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An Azeotropic Drying-Free Approach for Copper-mediated Radiofluorination without Addition of Base

Xiang Zhang*, Falguni Basuli, and Rolf E. Swenson

Imaging Probe Development Center, National Heart, Lung, and Blood Institute, National Institutes of Health, Rockville, MD

Abstract

Copper-mediated radiofluorination provides a quick and versatile approach for ^{18}F -labeling of arenes and heteroarenes. However, this method is known to be base-sensitive which has been a barrier for preparative scale radiosynthesis. In this report, we provide an approach for copper-mediated radiofluorination without azeotropic drying or adding a base. $[^{18}\text{F}]\text{Fluoride}$ trapped on a PS- HCO_3 Sep-Pak was quantitatively eluted with a solution of 4-dimethylaminopyridinium trifluoromethanesulfonate (DMAP-OTf) in anhydrous *N,N*-dimethylformamide (DMF). The eluted solution was directly used for copper-mediated radiofluorination. Twelve boronic ester substrates were tested, yielding fluorinated products in 27 – 83% radiochemical yield based on HPLC analysis. This approach was successfully applied to the radiosynthesis of $[^{18}\text{F}]\text{flumazenil}$, a well-known PET tracer for imaging central benzodiazepine receptors, with a radiochemical yield of 47%. This highly efficient protocol significantly augments the powerful copper-mediated radiofluorination approach.

1. Introduction

Positron emission tomography (PET) is one of the leading imaging techniques in both clinical and research settings.¹⁻⁴ It provides valuable functional information about a living subject, which helps the physicians and researchers for disease diagnosis, therapy monitoring, and drug evaluation. Among the PET radionuclides, fluorine-18 is undoubtedly the most popular isotope.^{4, 5} It has a convenient half-life of 110 min, which allows multi-step syntheses and regional dose transportation. In addition, fluorine-18 has a clean decay profile (97% positron emission) and low β^+ decay energy (0.633 MeV) which is optimal for high resolution PET imaging.⁶

The traditional radiofluorination strategy involves electrophilic or nucleophilic substitution, with the latter being commonly used for decades.^{4, 7-9} However, radiofluorination of electron rich arenes or heteroarenes remains challenging.^{10, 11} In the past decade, numerous approaches have been developed to prepare fluorine-18 labeled arenes or heteroarenes, including iodonium salt,¹²⁻¹⁴ ylide,^{15, 16} and transition metal mediated approaches.¹⁷⁻²⁴ Among these, the copper-mediated radiofluorination approach developed by Gouverneur group has gained a great deal of attention due to the tolerance of a wide range of functional

*Corresponding Author: Xiang Zhang.

groups on both electron-rich or electron-deficient substrates.²² By treating the pinacolyl arylboronate (arylBPin) precursor with Cu(OTf)₂py₄ and [¹⁸F]KF/K₂₂₂, a number of ¹⁸F-labeled arenes and heteroarenes were readily prepared.²² Further optimization of this efficient method has been performed by several research groups. For example, a “low-base” protocol was developed to efficiently produce PET tracers on a preparative scale.²⁵ Other substrates such as boronic acids and arylstannanes have also been successfully evaluated.^{23, 26} However, the excess base and phase transfer agent (PTA) can significantly diminish the conversion, since the Cu catalyst possess low stability under basic conditions.^{23, 25} It is still challenging to apply this method on scale-up applications and routine productions of PET tracers. Recently, Mossine *et al* reported the use of an organic base, 4-(dimethylamino)pyridine (DMAP), and a customized elution technique for Cu-mediated radiofluorination with up to 58% radiochemical yield (RCY) based on radio-TLC.²⁷ However, the isolated yield is relatively low, possibly due to activity loss during elution and azeotropic drying. In another recent publication a Et₄N·OTf elution approach was developed for successful ¹⁸F-fluorodestannylation through the copper-mediated process.²⁸ Overall, Cu-mediated radiofluorination is a significant improvement to the library of radiofluorination methods, and its full potential is being discovered to benefit radiochemistry communities.

Our group recently developed a “Radiofluorination on the Sep-Pak” method to quickly and efficiently prepare 6-[¹⁸F]fluoronicotinic acid-2,3,5,6-tetrafluorophenyl ester, a prosthetic group for peptide/protein labeling, without the addition of base or azeotropic drying.^{29, 30} Considering the base-sensitivity of the Cu-mediated radiofluorination process, a similar ¹⁸F-elution process could significantly aid this transformation. In this work, [¹⁸F]fluoride trapped on the Sep-Pak was eluted with an anhydrous organic solution of pyridinium salt such as pyridinium trifluoromethanesulfonate (Py·OTf). This approach does not require azeotropic drying, as no water was used for the elution of fluoride. Moreover, the absence of base increases the stability of both precursor and the copper reagent leading to better radiochemical yield. In this paper, we report our findings for an improved Cu-mediated radiofluorination through a no-base-added, azeotropic drying- free ¹⁸F-elution process.

2. Results and Discussion

To begin our study, the elution of [¹⁸F]fluoride from PS-HCO₃ Sep-Pak (Figure 1) and subsequent Cu-mediated fluorination was first tested with Py·OTf. Quantitative elution of [¹⁸F]fluoride was obtained (Table 1, entry 1) by slowly passing a DMF solution of Py·OTf through the Sep-Pak containing [¹⁸F]fluoride at 0.5 mL/min speed. The eluted [¹⁸F]pyridinium fluoride solution was subjected to the model reaction of 4-[¹⁸F]fluoroacetophenone synthesis via Cu-mediated radiofluorination. However, no product was obtained. Addition of pyridine as a co-eluent showed no effect on [¹⁸F]fluoride elution or radiochemical conversion (entry 2). To further optimize the protocol, 4-(dimethylamino)pyridinium trifluoromethanesulfonate (DMAP·OTf) was tested, as DMAP has been proven more efficient (10-fold higher efficiency) than pyridine in Cu-mediated radiofluorination.²⁷ DMAP was readily converted to the trifluoromethanesulfonic acid salt, DMAP·OTf, by mixing DMAP with trifluoromethanesulfonic acid, and used without purification. The fluoride elution efficiency (95%) was comparable with Py·OTf. More

importantly, the desired 4- ^{18}F fluoroacetophenone (entry 3) was prepared with a radiochemical yield of 41%.

This result prompted us to further optimize the radiosynthesis. Additional DMAP in the eluting solution was not productive, as it reduced the yield to 9% (entry 4). Increasing the reaction temperature to 120 °C improved the yield (entry 5). The effect of various amounts of DMAP·OTf was also tested for ^{18}F fluoride elution (entry 5-7). When 5 mg DMAP·OTf was used, the elution efficiency was slightly decreased to 83% (entry 6). However, larger amounts of DMAP·OTf (15 mg) reduced the radiochemical yield (entry 7). Therefore, we decided to perform the elution of ^{18}F fluoride with 10 mg of DMAP·OTf as this provided optimal balance between elution efficiency and yield. Various reaction temperatures were also evaluated, and slightly increased yield was observed at higher temperatures (entry 5 and entry 8). Further increasing the temperature to 130 °C had no effect on radiochemical yield (entry 9). Changing the solvent to *N,N*-dimethylacetamide (DMA) similarly showed no effect on radiochemical yield (entry 10). Based on the model reaction, we settled on a reaction temperature of 120 °C in DMF or DMA as the optimal conditions for the following studies. Variation of reaction time was not investigated in this optimization. All reactions were heated for 20 min, since it was the standardized time for most related works.^{22, 25, 27, 31}

The optimized protocol was tested on a substrate scope study of various boronic esters (Figure 2). The selected substrates contain electron deficient (**3a-b**) and electron rich arenes (**3d**, **3g**, **3j-l**), as well as heteroarenes such as quinoline, indole and pyridine moieties (**3c**, **3e-f**, **3h-i**). Using the optimized conditions, medium to high RCY was obtained for most substrates (up to 83%). The RCYs for few compounds such as **3c** and **3i** are significantly higher than previously reported, possibly due to the improved stability of the precursor under the no-base-added conditions.^{27, 31} The side-by-side comparison of their synthesis under literature “high-base” condition *vs.* this protocol was performed.^{25, 31} For **3c**, the literature method gave 31% RCY, whereas the RCY for the elution protocol was 83 ± 2%. Similar result was obtained for **3i**, which gave <2% RCY for the literature method *vs.* 27% for the elution protocol. The solvent effect was studied for compound **3h**, since the reaction in DMF resulted in a low RCY (19%). Switching the solvent to DMA improved the yield to 40%. Similar yield difference in DMF *vs.* DMA was reported in the literature,^{11, 31} possibly due to the undesired reaction of precursor with dimethylamine (resulting from thermal decomposition of DMF). RCY was determined by analytical HPLC of the crude product and the identity of product was confirmed by co-injection with the standard compounds.

Fluorine-18 labeled flumazenil is a well-known tracer for quantitative evaluation of central benzodiazepine receptors (Scheme 1).^{32, 33} Recently, Preshlock *et al.* reported the radiosynthesis of this tracer via copper-mediated radiofluorination with a RCY of 19%.¹¹ In this study, we tested the labeling efficiency with the DMAP·OTf elution method. The reaction was optimized with low amounts of ^{18}F fluoride (0.37 - 0.76 GBq, 10-20 mCi) at varying temperatures, a 51% RCY was obtained at 110 °C with DMAP·OTf in DMA as eluent. When the radiosynthesis was performed on a production scale (6.7 - 7.4 GBq) with HPLC purification, ^{18}F flumazenil was successfully obtained with 26 - 47% RCY (isolated, *n* = 3) in 55-60 min. The molar activity is 100 - 126 GBq/μmol at the end of synthesis. The

identity of [^{18}F]flumazenil was confirmed by co-injection with an authentic nonradioactive standard on the analytical HPLC (Figure 3).

We believe our protocol increased the stability of both the precursor and the copper reagent. As a result, less mazenil pinacol boronate (**4**) and $\text{Cu}(\text{OTf})_2$ were needed in the radiosynthesis, which will significantly reduce the cost of the overall procedure. In our improved [^{18}F]flumazenil synthesis, 2 mg of mazenil pinacol boronate precursor and 3 mg of $\text{Cu}(\text{OTf})_2$ were sufficient to give the purified product in 47% RCY. Low reagent loading and fewer side reactions should generally simplify the purification process, especially for reactions with base-sensitive precursors.

3. Conclusions

Copper-mediated radiofluorination is an efficient method of incorporating fluorine-18 on both electron-rich and electron-deficient substrates. However, this method requires base to elute fluorine-18 from Sep-Pak and azeotropic drying. The current method requires neither the base to elute fluorine-18 from Sep-Pak nor azeotropic drying. Fluorine-18 eluted from the Sep-Pak with DMAP-OTF can be used directly for radiolabeling. This method will be beneficial for base sensitive precursors and labeled tracers. Moreover, stability of the copper reagent will be higher under this reaction conditions. The current method was successfully validated on 12 aryl- and heteroaryl-boronic ester substrates with a range of electronic properties. Using this approach, the radiosynthesis of [^{18}F]flumazenil was achieved with 47% isolated RCY in 55-60 min synthesis time. The total synthesis time is shorter than the literature reported method (75-80 min) due to no need of azeotropic drying of fluorine-18.³³ This highly efficient method will significantly boost the application scope of the powerful copper-mediated radiofluorination. Further evaluation of this method on other PET tracers is in progress.

4. Methods

4.1 General

Unless otherwise noted, all chemicals and solvents were purchased from Sigma-Aldrich (Milwaukee, WI, USA) or Combi-blocks (San Diego, CA, USA) and used without further purification. Flumazenil and its boronate precursor were purchased from ABX GmbH (Radeberg, Germany). Non-carrier added [^{18}F]fluoride was obtained from the National Institutes of Health cyclotron facility (Bethesda, MD, USA). Chromafix PS- HCO_3 anion-exchange Sep-Pak cartridges were purchased from Synthra (Hamburg, Germany) and the packing material was reduced to half (~20 mg) for better elution efficiency. Ultra-pure water was produced with the Milli-Q[®] Integral water purification system (Billerica, MA, USA). The Fusion 100 syringe pump was purchased from Chemyx (Stafford, TX, USA).

NMR data were recorded on a Bruker 400 MHz spectrometer (Billerica, MA, USA). High resolution mass spectrometry (HRMS) was carried out on an Agilent TOF mass spectrometer (Santa Clara, CA, USA). Radio-TLC was performed on an Eckert-Ziegler AR-2000 radio-TLC Imaging Scanner (Hopkinton, MA, USA). High performance liquid chromatography (HPLC) purification and analytical HPLC were conducted on the Agilent

1260 HPLC system equipped with multi-wavelength UV detector along with a flow count radiodetector (Eckert & Ziegler, B-FC-3500 diode).

HPLC conditions:

Method A: Phenomenex Luna C18 (2) column, 100×4.6 mm, 5 µm. Mobile phase: A: water (0.1% NH₄OH); B: acetonitrile (0.1% NH₄OH). Gradient: 20 – 80% B in 10 min; 1 mL/min.

Method B: Phenomenex Luna C18 (2) column, 100×4.6 mm, 5 µm. Mobile phase: 25% acetonitrile in water (0.1% formic acid); 1.0 mL/min.

Method C: Phenomenex Luna C18 (2) column, 250×10 mm, 5 µm. Mobile phase: 22% acetonitrile in water (25 mM NH₄OAc); 4 mL/min.

4.2 Chemical synthesis

4-Dimethylaminopyridinium trifluoromethanesulfonate (DMAP-OTf)—A solution of trifluoromethanesulfonic acid (735 mg, 4.9 mmol) in dichloromethane (10 mL) was cooled to 0 °C. A solution of 4-dimethylaminopyridine (598 mg, 0.49 mmol) in dichloromethane (5 mL) was added dropwise and the mixture was stirred for 15 min. The precipitated product was collected by vacuum filtration and washed with cold dichloromethane (10 mL). The product was further dried under high vacuum overnight (1.3 g, quantitative yield). No further purification was necessary. ¹H NMR (400 MHz, DMSO) δ 13.15 (s, 1H), 8.21 (d, *J* = 7.6 Hz, 2H), 6.98 (d, *J* = 7.7 Hz, 2H), 3.19 (s, 6H). ¹³C NMR (101 MHz, DMSO) δ 157.43, 139.56, 121.15 (d, *J* = 322.4 Hz), 107.41, 40.09. HRMS (ESI): Calcd for C₇H₁₀N₂ (M+H)⁺ 123.0922, found 123.0919.

4.3 Radiochemistry

General procedure for the Cu-mediated radiofluorination via DMAP-OTf elution—[¹⁸F]fluoride (0.37 – 0.76 GBq, 10 – 20 mCi) in target water was diluted with 2 mL water and passed through a short PS-HCO₃ cartridge (see Section 4.1). The cartridge was washed with anhydrous acetonitrile (6 mL) and dried for 2 min. The [¹⁸F]fluoride on the cartridge was eluted (0.5 mL/min, manually or via syringe pump) with a DMAP-OTf (10 mg) solution in DMF or DMA (1 mL) to a reaction vial containing the boronate precursor (10 mg) and Cu(OTf)₂ (3 mg). The cartridge was further eluted with DMF or DMA (0.3 mL) to the same vial. The reaction mixture was heated at 120 °C for 20 min, an aliquot (~50 µL) was diluted with water (50 µL) and analyzed by analytical HPLC (method A) and radio-TLC for RCY determination. The identity of the product was confirmed by co-injection with unlabeled standard compound.

General procedure for the Cu-mediated radiofluorination via traditional azeotropic drying process for 3c and 3i—[¹⁸F]fluoride (0.37 - 0.76 GBq, 10 - 20 mCi) in target water was diluted with 2 mL water and passed through a short PS-HCO₃ cartridge (see Section 4.1). [¹⁸F]fluoride was eluted from the cartridge with the eluent (2.8 mg K₂CO₃, 12 mg K₂₂₂ in 1 mL acetonitrile and 280 µL water). The solution was azeotropically dried under vacuum and nitrogen flow at 110 °C. The azeotropic drying process was repeated by adding acetonitrile (1 mL × 3). To the dried activity was added the

boronic ester precursor (60 μ mol), tetrakis(pyridine)copper(II) triflate ($\text{Cu}(\text{OTf})_2(\text{py})_4$, 15 mg, 22 μ mol) in DMF or DMA (0.3 mL). The resulting mixture was heated under 110 $^\circ\text{C}$ for 20 min. An aliquot ($\sim$ 50 μL) was diluted with water (50 μL) and analyzed by analytical HPLC (method A). The identity of the product was confirmed by co-injection with unlabeled standard compound.

Radiosynthesis of [^{18}F]flumazenil—[^{18}F]fluoride (6.7 GBq, 188 mCi) in target water was diluted with 2 mL water and passed through a short PS- HCO_3 cartridge. The cartridge was washed with anhydrous acetonitrile (6 mL) and dried for 2 min. The [^{18}F]fluoride on the cartridge was eluted (0.5 mL/min) with a DMAP-OTf (10 mg) solution in DMA (1 mL) to a reaction vial containing mazenil pinacol boronate (2 mg) and $\text{Cu}(\text{OTf})_2$ (3 mg). The cartridge was further eluted with DMA (0.3 mL) to the same vial. The reaction mixture was heated at 110 $^\circ\text{C}$ for 20 min, then diluted with aqueous NH_4OAc solution (10 mM, 2 mL), and purified by HPLC using semi-preparative column (method C). The product peak was collected between 26 - 28 min. The radiosynthesis including [^{18}F]fluoride catch on the Sep-Pak, radiofluorination, and purification is completed in 55-60 min.

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References

- [1]. Ametamey SM, Honer M, and Schubiger PA. Molecular imaging with PET. *Chem Rev* 2008;108:1501–16. [PubMed: 18426240]
- [2]. Phelps ME. Positron emission tomography provides molecular imaging of biological processes. *Proc Natl Acad Sci* 2000;97:9226–9233. [PubMed: 10922074]
- [3]. Matthews PM, Rabiner EA, Passchier J, and Gunn RN. Positron emission tomography molecular imaging for drug development. *Br J Clin Pharmacol* 2012;73:175–86. [PubMed: 21838787]
- [4]. Miller PW, Long NJ, Vilar R, and Gee AD. Synthesis of ^{11}C , ^{18}F , ^{15}O , and ^{13}N radiolabels for positron emission tomography. *Angew Chem Int Ed Engl* 2008;47:8998–9033. [PubMed: 18988199]
- [5]. van der Born D, Pees A, Poot AJ, Orru RVA, Windhorst AD, and Vugts DJ. Fluorine-18 labelled building blocks for PET tracer synthesis. *Chem Soc Rev* 2017;46:4709–4773. [PubMed: 28608906]
- [6]. Levin CS and Hoffman EJ. Calculation of positron range and its effect on the fundamental limit of positron emission tomography system spatial resolution. *Phys Med Biol* 1999;44:781–99. [PubMed: 10211810]
- [7]. Alauddin MM. Positron emission tomography (PET) imaging with ^{18}F -based radiotracers. *Am J Nucl Med Mol Imaging* 2012;2:55–76. [PubMed: 23133802]
- [8]. Jacobson O, Kiesewetter DO, and Chen X. Fluorine-18 radiochemistry, labeling strategies and synthetic routes. *Bioconjug Chem* 2015;26:1–18. [PubMed: 25473848]
- [9]. Cai L, Lu S, and Pike VW. Chemistry with [^{18}F]fluoride ion. *Eur J Org Chem* 2008;2008:2853–2873.
- [10]. Tredwell M and Gouverneur V. ^{18}F -labeling of arenes. *Angew Chem Int Ed Engl* 2012;51:11426–37. [PubMed: 23086547]

- [11]. Preshlock S, Tredwell M, and Gouverneur V. ^{18}F -Labeling of arenes and heteroarenes for applications in positron emission tomography. *Chem Rev* 2016;116:719–66. [PubMed: 26751274]
- [12]. Pike VW and Aigbirhio FI. Reactions of cyclotron-produced [^{18}F]fluoride with diaryliodonium salts—a novel single-step route to no-carrier-added [^{18}F]fluoroarenes. *J Chem Soc Chem Commun* 1995:2215–2216.
- [13]. Shah A W Pike V, and A. Widdowson D. The synthesis of [^{18}F]fluoroarenes from the reaction of cyclotron-produced [^{18}F]fluoride ion with diaryliodonium salts. *J Chem Soc Perkin Trans 1* 1998:2043–2046.
- [14]. Ross TL, Ermert J, Hocke C, and Coenen HH. Nucleophilic ^{18}F -fluorination of heteroaromatic iodonium salts with no-carrier-added [^{18}F]fluoride. *J Am Chem Soc* 2007;129:8018–8025. [PubMed: 17536798]
- [15]. Cardinale J, Ermert J, Humpert S, and Coenen HH. Iodonium ylides for one-step, no-carrier-added radiofluorination of electron rich arenes, exemplified with 4-((^{18}F]fluorophenoxy)-phenylmethyl)piperidine NET and SERT ligands. *RSC Advances* 2014;4:17293–17299.
- [16]. Rotstein BH, Stephenson NA, Vasdev N, and Liang SH. Spirocyclic hypervalent iodine(III)-mediated radiofluorination of non-activated and hindered aromatics. *Nat Commun* 2014;5:4365. [PubMed: 25007318]
- [17]. Lee E, Kamlet AS, Powers DC, Neumann CN, Boursalian GB, Furuya T, et al. A fluoride-derived electrophilic late-stage fluorination reagent for PET imaging. *Science* 2011;334:639–42. [PubMed: 22053044]
- [18]. Kamlet AS, Neumann CN, Lee E, Carlin SM, Moseley CK, Stephenson N, et al. Application of palladium-mediated ^{18}F -fluorination to PET radiotracer development: overcoming hurdles to translation. *PLoS One* 2013;8:e59187. [PubMed: 23554994]
- [19]. Lee E, Hooker JM, and Ritter T. Nickel-mediated oxidative fluorination for PET with aqueous [^{18}F]fluoride. *J Am Chem Soc* 2012;134:17456–17458. [PubMed: 23061667]
- [20]. Ren H, Wey HY, Strebl M, Neelamegam R, Ritter T, and Hooker JM. Synthesis and imaging validation of [^{18}F]MDL100907 enabled by Ni-mediated fluorination. *ACS Chem Neurosci* 2014;5:611–5. [PubMed: 24845956]
- [21]. Ye Y, Schimler SD, Hanley PS, and Sanford MS. $\text{Cu}(\text{OTf})_2$ -mediated fluorination of aryltrifluoroborates with potassium fluoride. *J Am Chem Soc* 2013;135:16292–5. [PubMed: 24160267]
- [22]. Tredwell M, Preshlock SM, Taylor NJ, Gruber S, Huiban M, Passchier J, et al. A general copper-mediated nucleophilic ^{18}F -fluorination of arenes. *Angew Chem Int Ed Engl* 2014;53:7751–5. [PubMed: 24916101]
- [23]. Mossine AV, Brooks AF, Makaravage KJ, Miller JM, Ichiishi N, Sanford MS, et al. Synthesis of [^{18}F]arenes via the copper-mediated [^{18}F]fluorination of boronic acids. *Org Lett* 2015;17:5780–3. [PubMed: 26568457]
- [24]. Sergeev ME, Morgia F, Lazari M, Wang C, Jr., and van Dam RM. Titania-catalyzed radiofluorination of tosylated precursors in highly aqueous medium. *J Am Chem Soc* 2015;137:5686–94. [PubMed: 25860121]
- [25]. Zlatopolskiy BD, Zischler J, Krapf P, Zarrad F, Urusova EA, Kordys E, et al. Copper-mediated aromatic radiofluorination revisited: efficient production of PET tracers on a preparative scale. *Chemistry* 2015;21:5972–9. [PubMed: 25708748]
- [26]. Makaravage KJ, Brooks AF, Mossine AV, Sanford MS, and Scott PJ. Copper-mediated radiofluorination of arylstannanes with [^{18}F]KF. *Org Lett* 2016.
- [27]. Mossine AV, Brooks AF, Ichiishi N, Makaravage KJ, Sanford MS, and Scott PJH. Development of customized [^{18}F]fluoride elution techniques for the enhancement of copper-mediated late-stage radiofluorination. *Sci Rep* 2017;7:233. [PubMed: 28331174]
- [28]. Zarrad F, Zlatopolskiy BD, Krapf P, Zischler J, and Neumaier B. A practical method for the preparation of ^{18}F -labeled aromatic amino acids from nucleophilic [^{18}F]fluoride and stannyl precursors for electrophilic radiohalogenation. *Molecules* 2017;22: 2231.

- [29]. Basuli F, Zhang X, Jagoda EM, Choyke PL, and Swenson RE. Facile room temperature synthesis of fluorine-18 labeled fluoronicotinic acid-2,3,5,6-tetrafluorophenyl ester without azeotropic drying of fluorine-18. *Nucl Med Biol* 2016;43:770–772. [PubMed: 27693671]
- [30]. Basuli F, Zhang X, Woodroffe CC, Jagoda EM, Choyke PL, and Swenson RE. Fast indirect fluorine-18 labeling of protein/peptide using the useful 6-fluoronicotinic acid-2,3,5,6-tetrafluorophenyl prosthetic group: A method comparable to direct fluorination. *J Label Compd Radiopharm* 2017;60:168–175.
- [31]. Taylor NJ, Emer E, Preshlock S, Schedler M, Tredwell M, Verhoog S, et al. Derisking the Cu-mediated ^{18}F -fluorination of heterocyclic positron emission tomography radioligands. *J Am Chem Soc* 2017;139:8267–8276. [PubMed: 28548849]
- [32]. Ryzhikov NN, Gomzina NA, Fedorova OS, Vasil'ev DA, Kostikov AP, and Krasikova RN. Preparation of [^{18}F]flumazenil, a potential radioligand for PET imaging of central benzodiazepine receptors, by isotope exchange. *Radiochemistry* 2004;46:290–294.
- [33]. Ryzhikov NN, Seneca N, Krasikova RN, Gomzina NA, Shchukin E, Fedorova OS, et al. Preparation of highly specific radioactivity [^{18}F]flumazenil and its evaluation in cynomolgus monkey by positron emission tomography. *Nucl Med Biol* 2005;32:109–116. [PubMed: 15721755]

Figure 1.
[^{18}F]fluoride elution approach from PS- HCO_3 cartridge.

Figure 2.
Substrate scope study. Reaction condition: precursor (10 mg), DMAP·OTf (10 mg),
Cu(OTf)₂ (3 mg), DMF or DMA (1.3 mL), 120 °C, 20 min.

Figure 3.

HPLC analysis of a) $[^{18}\text{F}]$ flumazenil; b) $[^{18}\text{F}]$ flumazenil, co-injected with the nonradioactive standard using Method B. black line, in-line radiodetector; gray line, UV detector at 254 nm.

Scheme 1. Radiosynthesis of [^{18}F]flumazenil with DMAP·OTf elution method.

Table 1.

Model reaction of Cu-mediated fluorination.^a

Entry	Eluting agent	Amount (mg)	Co-eluent	Solvent	Elution efficiency (%)	T (°C)	RCY ^b (%)
1	Py·OTf	10	n/a	DMF	>95	120	0
2	Py·OTf	10	Py ^c	DMF	>95	120	0
3	DMAP·OTf	10	n/a	DMF	>95	110	41
4	DMAP·OTf	10	DMAP ^d	DMF	>95	110	9
5	DMAP·OTf	10	n/a	DMF	>95	120	51 ± 2 (n = 3)
6	DMAP·OTf	5	n/a	DMF	83	120	49
7	DMAP·OTf	15	n/a	DMF	>95	120	43
8	DMAP·OTf	10	n/a	DMF	>95	105	40
9	DMAP·OTf	10	n/a	DMF	>95	130	51
10	DMAP·OTf	10	n/a	DMA	>95	120	53

^aNote: 10 mg ArBpin precursor (40 μmol), 0.37 - 0.76 GBq of [¹⁸F]fluoride were used for each reaction.^bRCY were determined by analytical HPLC (method A) and radio-TLC.^c10 μL of pyridine.^d5 mg of DMAP.