Discovery of a Novel 2,6-Disubstituted Glucosamine Series of Potent and Selective Hexokinase 2 Inhibitors

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S Supporting Information

ABSTRACT: A novel series of potent and selective hexokinase 2 (HK2) inhibitors, 2,6-disubstituted glucosamines, has been identified based on HTS hits, exemplified by compound **1**. Inhibitor-bound crystal structures revealed that the HK2 enzyme could adopt an “induced-fit” conformation. The SAR study led to the identification of potent HK2 inhibitors, such as compound **34** with greater than 100-fold selectivity over HK1. Compound **25** inhibits *in situ* glycolysis in a UM-UC-3 bladder tumor cell line via ¹³CNMR measurement of [3-¹³C]lactate produced from [1,6-¹³C₂]glucose added to the cell culture.

KEYWORDS: Hexokinase 2 inhibitor, crystal structure, structure–activity relationship selectivity

Inhibition of glycolysis has been considered as a therapeutic approach for anticancer treatment for decades.¹ As the first enzyme in the glycolytic pathway, hexokinase-2 (HK2) plays two pivotal roles in cancer. HK2 catalyzes the first functionally irreversible reaction in the glycolytic pathway to form glucose-6-phosphate (G6P) from glucose and ATP. It drives the “Warburg effect” by committing glucose to enter both the glycolytic and pentose phosphate pathways. Second, HK2 stabilizes the Mitochondrial Outer Membrane (MOM) via interaction with the voltage-dependent anion channel (VDAC) to prevent apoptosis.² The HK2 gene is amplified and HK2 RNA is overexpressed in tumors.³ Epigenetic changes such as HK2 DNA demethylation have been reported as well. The most profound changes in HK2 expression have been observed in liver, kidney, endometrial, and pancreatic cancers.⁴ HK2 expression is regulated by many somatic stimuli: HIF1- α , growth factors, c-MYC, and p53 mutations. Akt stimulates HK2 association with mitochondria to promote mitochondrial integrity and prevent apoptosis. HK2 expression correlates with tumor stage and a worse prognosis. Studies with HK2 knockdowns using siRNA or shRNA both *in vitro* and *in vivo* have shown inhibition of tumor growth.^{5,6}

Recently, Hay and his collaborators provided compelling evidence for the role of HK2 in tumorigenesis and maintenance, as well as metabolic insights via Hk2 depletion using a genetic mouse model of nonsmall cell lung carcinoma (NSCLC) induced by expression of activated KRAS (KRAS-LA2) and a mouse model of breast cancer induced by

expression of activated ERBB2/Neu (MMTV-neu).⁷ They also showed that global Hk2 ablation in adult mice was well-tolerated without significant physiological consequences.

While HK2 is a potential target for cancer treatment, it has been considered intractable for the past 50 years due to its extremely polar active site, the complexity of its protein functions, and the uncertainty associated with finding a HK2 selective inhibitor over the housekeeping HK1 isozyme. Encouraged by the strong rationale, we initiated research to identify HK2 selective small-molecule inhibitors for potential cancer treatment or in combination with existing drugs to sensitize chemotherapy and targeted therapy.

Compound **1** was one of the glucosamine derivatives identified from the high throughput screen (Supporting Information) with purified HK2, and we first synthesized a variety of C-2 amides to examine potency and HK2 vs HK1 selectivity.⁸ As shown in Table 1, compound **1** is weakly active against HK2, but has no selectivity over HK1. Using a continuous coupled assay (formation of G6P coupled to G6P dehydrogenase), compound **1** was found to be competitive with glucose ($K_i = 2.9 \pm 0.33 \mu\text{M}$) and noncompetitive versus MgATP (Supporting Information). In a separate double inhibition study (continuous coupled assay for ADP formation with pyruvate kinase and lactate dehydrogenase), compound **1**

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Table 1. SAR of C-2 Substituted Glucosamines

#	R1	R2	Enzyme IC ₅₀ (μM) ^a	
			HK2	HK1
1		OH	6.3	2.0
2		OH	16	13
3		OH	3.2	0.79
4		OH	0.79	0.040
5		OH	16	160
6		OH	2.0	2.0
7		OMe	>400	>200
8		H,H	>400	>200

^aIC₅₀ values given are means of at least 2 experiments.

was also shown to bind simultaneously with G6P (Supporting Information). From structural studies with HK1, G6P is known to bind separately from glucose in a nearby “allosteric” site with the pyranose ring at the putative position of the ATP-bound Mg²⁺ cation.⁹ Further modifications of the C-2 amides indicated that bulky substitutions at the *meta*-positions of the benzene ring improve HK2 potency (see 3 and 4). However, such modifications impacted HK1 potency more; for example, compound 4 was a very potent HK1 inhibitor with an IC₅₀ of 40 nM. On the other hand, a bulky aliphatic amide as in compound 5 appeared to enhance HK2 selectivity (IC₅₀ = 16 and 160 μM, respectively, for HK2 and HK1), but suffered from weak potency. Compound 6 with a 3,5-dinitrobenzamide

is equally active against HK2 and HK1 with an IC₅₀ value of 2.0 μM.

We also examined the importance of the C1-hydroxy group for inhibitor potency. Unfortunately, both the 1-methoxy analog (7) and the 1-deoxy analog (8)¹⁰ were inactive.

To understand the binding of these inhibitors with HK2, we conducted crystallography studies, and obtained the very first ligand bound human HK2 cocrystal structure at 2.76 Å with compound 1 and glucose-6-phosphate (G6P).¹¹ The structure is consistent with the double inhibition kinetic study, which showed G6P and compound 1 could bind simultaneously to HK2. Both N- and C-terminal domains show compound 1 at their corresponding active sites, with the glucosamine ring placed in the glucose-binding pocket (Figure 1), which is consistent with the glucose competitive mode of action of these compounds. All donor–acceptor interactions of the hydroxyl groups of glucose with HK2 are also retained in the interaction of compound 1 with HK2. Since the 1-hydroxyl group forms a tight hydrogen bonding network with a water molecule and the side chains of Gln739 and Glu742, it is not surprising that the 1-methoxy (7) and 1-deoxy (8) modifications are not tolerated.

The cocrystal structure of HK2 with compound 1 reveals that the enzyme binding site is highly flexible. Comparison with the HK2-glucose cocrystal structure (2NZT)¹² shows that, in both crystals (Figure 1A), the glucose ring binds tightly to the big lobe of the enzyme via hydrogen bonds to Glu708, Gln739, and Glu742. But for compound 1, the active-site loop (residues 616–633, colored red in Figure 1), which contains the catalytic Lys621, is unable to fully close over the pyranose ring, as it is displaced outward by 3 Å due to steric clash with the C-2 amide side chain. This prevents the enzyme from phosphorylating the C-6 hydroxyl group of compound 1. Because of such displacement, the narrow pocket extending from the C-2 equatorial position of the pyranose ring is considerably widened and can now easily accommodate bulkier amide substituents, a clear example of enzyme “induced fit” by a ligand. This wider pocket is lined with side chains from Pro605, Lys618, Lys738, and Gln739 and is highly lipophilic, which explains the increased potency of the biphenyl-derivative 4 (Table 1). A hydrogen bond between the C-2 amide carbonyl and the Thr620 side chain helps to direct the C-2 substituent into this lipophilic pocket. The structure also suggests that the extension of the molecule into the G6P binding site through derivatization at the C-6 position could be a way to improve potency of the glucosamine series. Since three oxygen atoms of

Figure 1. (A) Compound 1 in HK2 with G6P overlaid with 2NZT; (B) Compound 1 and G6P interactions with HK2 in the C-terminal catalytic pocket.

the phosphate of G6P form H-bond interactions with Thr536, Thr680, and Thr863, a polar group to mimic one of the H-bond interactions might be important.

Based on this hypothesis, we set out to modify the 6-position as depicted in Scheme 1, in which the 1-hydroxy of compound

Scheme 1. Synthesis of 2,6-Disubstituted Glucosamines

Conditions and Reagents: (a) allyl alcohol, TsOH·H₂O, 110 °C, 72 h, 93%; (b) TsCl, py, rt, 3 h; (c) NaN₃, DMF, 90 °C, 17 h; 52%; (d) Ph₃P, H₂O, THF, rt, 72 h, 43%; (e) benzenesulfonyl chloride, py, rt, 20 h; (f) PdCl₂, MeOH, rt, 4 h; 19%; (g) allyl alcohol, TsOH·H₂O, 110 °C, 2 h; (h) TsCl, py, 0 °C, 6 h, 32%; (i) NaN₃, DMF, 80 °C, 24 h; (j) Ph₃P, H₂O, THF, rt, 24 h; (k) 2,3-dichlorobenzenesulfonyl chloride, py, rt, 24 h, 30%; (l) iodine, THF/H₂O 1:1, rt, 2 h; then solid sodium thiosulfate/sodium carbonate; (m) 4'-cyano-5-methoxy-[1,1'-biphenyl]-3-carboxylic acid, 2-(1H-benzotriazol-1-yl)-1,1,3,3-tetramethyluronium hexafluorophosphate (HBTU), diisopropylethylamine, CH₂Cl₂, rt, 24 h, 26%; (n) PdCl₂, MeOH, rt, 24 h; 46%.

4 was protected with an allyl group in a reaction catalyzed by TsOH hydrate or, alternatively, by BF₃·Et₂O at lower temperature (80 °C) and with shorter reaction time (18 h). The 6-hydroxy of compound 9 was then converted to the tosylate, which was treated with sodium azide in DMF to form compound 10. Reduction of the azide to a primary amine 11 with Ph₃P in THF/H₂O provided a key intermediate for C6 elaboration. As an example, sulfonamide 12 was formed by slow addition of benzenesulfonyl chloride to a solution of compound 11 in pyridine, followed by a Pd-mediated allyl deprotection in MeOH and reverse phase HPLC purification.

In order to develop a robust synthetic route to modify the 2-position of the 2,6-disubstituted glucosamine series, we introduced pent-4-enamide at the beginning to give compound 13.^{13,14} It could be deprotected under mild conditions later in the synthetic scheme. As depicted in Scheme 1, Method B, using the same transformations, the 6-hydroxy of compound 13 was converted to amine 14. Treatment of 2,3-dichlorobenzenesulfonyl chloride with 14 followed by deprotection of the pent-4-enamide with iodine afforded a second useful intermediate 15 for 2-position modification. As an example, compound 15 was converted to amide 16 via a coupling reaction with 4'-cyano-5-methoxy-[1,1'-biphenyl]-3-carboxylic acid¹⁵ in the presence of HBTU and DIEA, followed by a Pd-mediated deprotection of the allyl group to give the final compound 17 after reverse phase HPLC purification.

Various functional groups at the 6-position, including sulfonamide, amide, and urea, are tolerated (Supporting Information Table 1). Among them, the 6-sulfonamide (compound 18) is particularly promising with improved HK2 inhibitory activity.

Given how much has been known about HK2/HK1 substrate binding pockets and the fact that they have been considered as intractable targets for the last 50 years, it is remarkable that these dramatic modifications to glucose are tolerated and result in increased inhibition of both HK2/HK1 enzymatic activities. We therefore pursued this approach together with X-ray crystallography to understand the structural biology of HK2.

Our hypothesis to improve potency against HK2 by modification of the 6-position of the glucosamine series was further validated with additional 6-sulfonamide analogs. As demonstrated in Table 2, *ortho*-chloride (23) improves HK2

Table 2. SAR of 2-Amido-6-benzenesulfonamide glucosamines^a

#	R1	R2	Enzyme IC ₅₀ (μM) ^a	
			HK2	HK1
12			0.79	0.16
18			2.0	2.0
23			0.10	0.016
24			0.32	0.16
25			0.025	0.032
26			0.10	0.020
27			0.13	0.0079
28			0.050	0.025
29			0.63	1.6
30			0.010	0.020
17			0.025	0.0079

^aIC₅₀ values given are means of at least 2 experiments

and HK1 potency by 8- to 10-fold, respectively, compared to the parent benzenesulfonamide **12**, while 2,3-dichlorobenzene-sulfonamide **25** is the first potent HK2 inhibitor with an IC_{50} value less than 100 nM (HK2 IC_{50} = 25 nM, HK1 IC_{50} = 32 nM). Heterocyclic sulfonamides are also equally potent, exemplified by 4-chloropyridine-3-sulfonamide (**26**), compared to its corresponding chlorobenzenesulfonamide **23**.

Some polar substitutions were introduced into the benzenesulfonamide to potentially mimic H-bond interactions between the protein and G6P, since the 6-sulfonamides were designed to extend into the polar G6P binding pocket. As an example, a *para*-nitrile increases the potency of compound **27** against HK1 by 20-fold to 7.9 nM, while the increase of potency against HK2 is less profound, only 6-fold.

Interestingly, a *meta*-carboxylic acid also increases both HK2 and HK1 potency by 16-fold and 6-fold (compound **28** versus **12**), respectively. The acid–protein interactions appear to be specific, since compound **29**, bearing an ester group is far less potent than its corresponding acid **30**. Compound **30** is 63-fold more potent against HK2 (IC_{50} = 10 nM), and 80-fold more potent against HK1 (IC_{50} = 20 nM).

In addition, consistent with the SAR that bulky amides increase HK1 inhibitory potency in the monosubstituted 2-glucosamine chemical series, exemplified in Table 1 by compound **5**, it is not surprising that compound **17** is very potent against HK1, but HK2 potency remains the same as that of compound **25**.

Throughout the course of optimization, we have been seeking the “clues” in SAR to improve HK2 vs HK1 selectivity even though the sequences of these two proteins are 72% identical and 84% similar, and the glucose binding sites are well conserved among all hexokinases.¹⁶ Careful SAR analysis revealed consistent trends for HK2 selectivity: (a) 2,3-disubstituted sulfonamides at the 6-position; (b) less bulky amides at the 2-position; and (c) sulfonamide at the 2-position. Combination of any two of the elements could afford a compound as selective as 500-fold against HK2 versus HK1, such as compound **36**. This has been proven to be a tractable SAR to yield selective HK2 inhibitors exemplified by compounds **31–36** (Table 3 and Supporting Information Figure 5). Importantly, most selective compounds are also very potent; for example, 2,6-bis-sulfonamide **34** has a HK2 potency of 7.9 nM while its HK1 potency is only 1.0 μ M. Therefore, it has 125-fold selectivity against HK2 versus HK1.

With various tool molecules in hand, we successfully obtained some critical cocrystal structures to reveal the binding modes of these 2,6-disubstituted glucosamines. As shown in Figure 2, the cocrystal structure of compound **30** in HK2,¹⁷ a dimer with 4 molecules of ligands in both N- and C-terminal catalytic pockets, indeed supports our hypothesis that substitution at the 6-position can extend into the G6P pocket and increase potency. Similar to compound **1**, compound **30** also forms H-bond interactions with Gln742 and Glu708. However, it does not engage Lys621 because the catalytic loop (aa616–633) is no longer in proximity, since the pocket is even more open. The overlay (Figure 2B) of both cocrystal structures reveals: (1) the carboxylic acid group of compound **30** mimics the phosphate of G6P to form similar H-bond interactions with Thr680 and Thr863 but not Thr536; (2) the catalytic loop (aa616–633, salmon and red loops on the right) and the other loop that interacts with G6P (aa533–538, salmon and red loops on the left) are wide open to accommodate the

Table 3. HK2 Selective Glucosamine Derivatives

#	R1	Enzyme IC_{50} (μ M) ^a	
		HK2	HK1
31		0.040	2.0
32		0.13	10
33		0.025	5.0
34		0.0079	1.0
35		0.050	>20
36		0.13	63

^a IC_{50} values given are means of at least 2 experiments.

conformation change induced by 2,6-disubstituted glucosamines.

To our surprise, some 2,6-disubstituted glucosamines, although shown to be mutually exclusive (i.e., competitive) with G6P in double inhibition mechanistic experiments (Supporting Information), do not extend into the G6P pocket. Instead, the 6-position substituents point into the opening of the catalytic pocket to form a “folded” conformation, exemplified by the cocrystal structure of compound **27** in HK2,¹⁸ shown in Figure 2C. In addition to common H-bond interactions with Glu742 and Glu708, the cyano group of compound **27** forms a H-bond with Lys621. The overlay of the cocrystal structures of compound **27** and **30** in HK2 (Figure 2D) shows minimal change in the protein structures, and the catalytic pocket can accommodate both extended and folded conformations of 2,6-disubstituted glucosamines. This also explains the mutual exclusivity of “folded” compounds with G6P, since the loop (aa533–538, salmon and red loops on the left in Figure 2B) that interacts with G6P is “pushed” open, thereby preventing G6P from binding within the pocket.

To further establish that glycolysis is inhibited by an HK2 inhibitor in a tumor cell line, the following three experiments (details in Supporting Information) were conducted in UM-UC-3 cells, a bladder tumor cell line with high protein expression of HK2 and minimal expression of HK1 (Western blot, Supporting Information Figure 6): (1) inhibition of G6P production using a more stable 2-fluoro-2-deoxy-glucose (2FDG) as a surrogate for glucose; (2) reduction of glucose consumption; and (3) reduction of downstream metabolite lactate (Supporting Information Figure 7A–C). Compound **25** demonstrated inhibition of glycolysis in all three experiments with an IC_{50} value of 0.20 μ M in the first experiment.¹⁹ Similarly, the selective compounds **33** and **36** had IC_{50} values of 0.47 and 0.52 μ M, respectively, in the first experiment. Since HK2 is the dominant isozyme in this cell line, these results

Figure 2. (A) Compound 30 ligand interactions in HK2 in the C-terminal catalytic pocket. (B) Overlay of compound 30 and 1 cocrystal structures with HK2 in the C-terminal catalytic pocket. The salmon and red loops on the left are aa533–538. The salmon and red loops on the right are aa617–633. (C) Compound 27 ligand binding interactions in HK2 in the C-terminal site. (D) Overlay of compounds 27 and 30 in HK2 in the C-terminal catalytic pocket.

provide evidence for HK2 driven cellular activities of this novel 2,6-disubstituted glucosamine series. Although the glucose pockets are likely very different, inhibition of glucose transporters cannot be ruled out in these experiments and may require further investigation.²⁰

In summary, we have successfully optimized HK2 HTS hit compound 1 to inhibitors with nM potency and developed a tractable SAR to achieve a high level of HK2 versus HK1 selectivity. The 2,6-disubstituted glucosamine series was rationally designed based on a cocrystal structure of compound 1 and G6P (Figure 1) to “link” the glucose and G6P binding pockets. A 6-sulfonamide linker turned out to be optimal. The hypothesis of using an acid (compound 30) to mimic the phosphate of G6P was confirmed by the cocrystal structure of compound 30 with HK2 (Figure 2). We also discovered a tractable SAR via the combination of several features that lead to selective HK2 inhibitors over HK1. Several cocrystal structures of HK2 with inhibitors captured different conformations of the enzyme. These cocrystal structures reveal the flexibility of the HK2 protein and that the catalytic site can adopt an “induced-fit” conformation with inhibitors. Using compound 25 as an example in a tumor cell line such as UM-UC-3 that predominately expresses HK2, we have demonstrated that a potent HK2 small molecule inhibitor can inhibit glycolysis in tumor cells. To our best knowledge, compounds such as 33, 34, and 35 are the most advanced tool molecules to date in terms of potency and selectivity of HK2 over HK1 to study the role of HK2 enzymatic activity in tumorigenesis without compromising systemic homeostasis or normal metabolic functions of HK1.

■ ASSOCIATED CONTENT

§ Supporting Information

The Supporting Information is available free of charge on the ACS Publications website at DOI: 10.1021/acsmedchemlett.5b00214.

Biological assays, biological data, and experimental procedures, and additional SAR. (PDF)

Data for 17 (PDF)

Data for 32 (PDF)

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Notes

The authors declare no competing financial interest.

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