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High-Resolution Mn EXAFS of the Oxygen-Evolving Complex in Photosystem II: Structural Implications for the Mn₄Ca Cluster

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Photosynthetic water oxidation is a fundamental chemical reaction that sustains the biosphere and takes place at a catalytic Mn₄Ca site within photosystem II (PS II), which is embedded in the thylakoid membranes of green plants, cyanobacteria, and algae.^{1–3} The recent X-ray crystal structures with resolution between 3.2 and 3.8 Å^{4–7} locate the electron density associated with the Mn₄Ca cluster in the multiprotein PS II complex, at different levels of detail. However, these resolutions do not allow an accurate determination of the positions of Mn^{4–7} and Ca,⁶ of distances of Mn–Mn/Ca, or the bridging and terminal ligand atoms; therefore, detailed structures critically depend on spectroscopic techniques, such as EXAFS (extended X-ray absorption fine structure)^{8–10} and EPR/ENDOR.^{11–13} In addition, our recent studies with PS II crystals show that the Mn₄Ca cluster is highly susceptible to radiation damage; hence, it is impossible to obtain meaningful Mn–Mn/Ca/ligand distances from XRD under the current conditions of X-ray dose and temperature.¹⁴ EXAFS experiments require a lower X-ray dose than XRD, and radiation damage can be precisely monitored and controlled, thus allowing for data collection from an intact Mn₄Ca cluster. EXAFS data also provide significantly higher Mn distance accuracy and resolution than the current 3.2–3.5 Å crystal structures.^{15–17} Here, we present data from a high-resolution EXAFS method using a novel multicrystal monochromator (Figure 1)¹⁸ that can distinguish two distances that are apart by $\Delta R = 0.09$ Å with an accuracy of ~ 0.02 Å. These data resolve a distance heterogeneity in the short Mn–Mn distances of the S₁ and S₂ state and thereby provide firm evidence for three Mn–Mn distances between ~ 2.7 and ~ 2.8 Å. This result gives clear criteria for selecting and refining possible structures from the repertoire of proposed models based on spectroscopic and diffraction data.

The new detection technique results in significant improvement in metal-backscatterer distance resolution from EXAFS. In general, EXAFS spectra of systems which contain several different metals have a limited EXAFS range because of the presence of the rising edge of the next element, thus limiting the EXAFS distance resolution (see Supporting

Information for details).¹⁹ For the Mn *K*-edge EXAFS studies of PS II, the absorption edge of Fe in PS II limits the EXAFS energy range (Figure 1). Traditional EXAFS spectra of PS II samples are collected as an excitation spectrum by electronically windowing the *K* α fluorescence (2p to 1s, at 5899 eV) from the Mn atom.^{15–17} The solid-state detectors that have been used over the past decade have a resolution of about 150–200 eV (fwhm) at the Mn *K*-edge, making it impossible to discriminate Mn fluorescence from that of Fe *K* α fluorescence (at 6404 eV). The presence of the obligatory 2–3Fe/PS II (Fe edge at 7120 eV) limits the data to a *k*-range of $\sim 11.5 \text{ \AA}^{-1}$ ($k = 0.51\Delta E^{1/2}$, the Mn edge is at 6540 eV and $\Delta E = 580 \text{ eV}$) (Figures 1 and 2, left). The Mn–Mn and Mn its the data to a ligand distances that can be resolved in a typical EXAFS experiment are given by $\Delta R = \pi/2k_{\text{max}}$, where k_{max} is the maximum energy of the photoelectron of Mn. The use of a high-resolution crystal monochromator allows us to selectively separate the Mn *K* fluorescence from that of Fe (Figure 1), resulting in the collection of data to higher photoelectron energies and leading to increased distance resolution of 0.09 $\text{\AA}$. The new detection scheme produces distinct advantages: (1) improvement in the distance resolution, and (2) more precise determination in the numbers of metal–metal vectors.

Earlier EXAFS studies of the *S*₁ and *S*₂ states of the OEC have shown that each Mn is surrounded by a first coordination shell of O or N atoms at 1.8–2.1 $\text{\AA}$ (peak I in Figure 2), a second shell of short Mn–Mn distances at $\sim 2.7 \text{ \AA}$ (peak II), and long Mn–Mn and Mn–Ca distances at $\sim 3.3 \text{ \AA}$ (peak III).⁸ These studies firmly established that the OEC in PS II is comprised of di- μ -oxo-bridged Mn₂ motifs characterized by a Mn–Mn distance of $\sim 2.7 \text{ \AA}$ (peak II). However, there was uncertainty about whether there are two or three di- μ -oxo-bridged Mn–Mn moieties^{20–22} because of the inherent error of the EXAFS method in the determination of the number of Mn–Mn vectors.¹⁷ Range-extended EXAFS makes it possible to resolve and discriminate between the Mn–Mn distances contributing to peak II, facilitating a more accurate determination of the total number of interactions because there must be an integral number of each of the resolved distances.

Figure 2 shows the *k*³-weighted Mn EXAFS spectra and the corresponding Fourier transforms (FT) of *S*₁ and *S*₂ states obtained using a solid-state Ge detector compared with the range-extended EXAFS data collected well past the Fe *K*-edge using a high-resolution crystal monochromator. The improvement in the range of the *k*-space data results in higher resolution and better precision in the fits.

Table 1 summarizes one and two Mn–Mn shell fits to peak II in the *S*₁ and *S*₂ states. The overall fit quality is monitored by comparing Φ and ϵ^2 parameters (for details, see Supporting Information). There is significant improvement in the fit quality that contains two Mn–Mn distances of ~ 2.7 and $\sim 2.8 \text{ \AA}$ over one-shell fits with an average Mn–Mn distance of $\sim 2.74 \text{ \AA}$. This is illustrated by two different two-shell fits in which *N*₁ and *N*₂ were fixed at different ratios: (1) *N*₁:*N*₂ = 1:1, corresponding to a scenario of two Mn–Mn interactions; and (2) *N*₁:*N*₂ = 2:1, corresponding to three Mn–Mn interactions. The latter case is clearly preferable based on the quality of the fits to both the *S*₁ and *S*₂ state data.

The current observation by range-extended EXAFS supports models that contain three Mn–Mn vectors: two at $\sim 2.7 \text{ \AA}$ and one at $\sim 2.8 \text{ \AA}$. Models for the Mn cluster compatible with these parameters and containing one or two Mn–Mn interactions at 3.3 $\text{\AA}$ are shown in Figure 3.

Preliminary studies using oriented PS II membranes show much better resolution for Fourier peak III and raise the possibility of resolving the Mn–Mn/Ca contributions in future studies using this high-resolution EXAFS methodology. In conclusion, two Mn–Mn distances of

2.73 and 2.82 Å in a 2:1 ratio can be resolved using this new method, proving the presence of three short Mn–Mn vectors in the S₁ and S₂ states of the Mn₄Ca cluster.

Supplementary Material

Refer to Web version on PubMed Central for supplementary material.

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References

1. Debus RJ. *Biochim. Biophys. Acta*. 1992; 1102:269–352. [PubMed: 1390827]
2. Rutherford, AW.; Zimmermann, JL.; Boussac, A. *The Photosystems: Structure, Function, and Molecular Biology*. Barber, J., editor. Elsevier B. V.: Amsterdam; 1992. p. 179-229.
3. Ort, DR.; Yocum, CF. *Oxygenic Photosynthesis: The Light Reactions*. Kluwer Academic Publishers: Dordrecht, The Netherlands; 1996.
4. Zouni A, Witt HT, Kern J, Fromme P, Krauss N, Saenger W, Orth P. *Nature*. 2001; 409:739–743. [PubMed: 11217865]
5. Kamiya N, Shen JR. *Proc. Natl. Acad. Sci. U.S.A.* 2003; 100:98–103. [PubMed: 12518057]
6. Ferreira KN, Iverson TM, Maghlaoui K, Barber J, Iwata S. *Science*. 2004; 303:1831–1838. [PubMed: 14764885]
7. Biesiadka J, Loll B, Kern J, Irrgang KD, Zouni A. *Phys. Chem. Chem. Phys.* 2004; 6:4733–4736.
8. Yachandra VK, Sauer K, Klein MP. *Chem. Rev.* 1996; 96:2927–2950. [PubMed: 11848846]
9. Penner-Hahn JE. *Struct. Bonding*. 1998; 90:1–36.
10. Sauer K, Yano J, Yachandra VK. *Photosynth. Res.* 2005; 85:73–86. [PubMed: 15977060]
11. Peloquin JM, Britt RD. *Biochim. Biophys. Acta*. 2001; 1503:96–111. [PubMed: 11115627]
12. Hasegawa K, Ono TA, Inoue Y, Kusunoki M. *Chem. Phys. Lett.* 1999; 300:9–19.
13. Kulik LV, Epel B, Lubitz W, Messinger J. *J. Am. Chem. Soc.* 2005; 127:2392–2393. [PubMed: 15724984]
14. Yano J, Kern J, Irrgang KD, Latimer MJ, Bergmann U, Glatzel P, Pushkar Y, Biesiadka J, Loll B, Sauer K, Messinger J, Zouni A, Yachandra VK. *Proc. Natl. Acad. Sci. U.S.A.* 2005; 102:12047–12052. [PubMed: 16103362]
15. Scott, RA. *Structural and Resonance Techniques in Biological Research*. Rousseau, DL., editor. Orlando: Academic Press; 1984. p. 295-362.
16. Cramer, SP. *X-ray Absorption: Principles, Applications and Techniques of EXAFS, SEXAFS, and XANES*. Koningsberger, DC.; Prins, R., editors. New York: Wiley-Interscience; 1988. p. 257-320.
17. Yachandra VK. *Methods Enzymol.* 1995; 246:638–675. [PubMed: 7752941]
18. Bergmann, U.; Cramer, SP. *SPIE Conference on Crystal and Multilayer Optics*. Vol. Vol. 3448. San Diego, CA: SPIE; 1998. p. 198-209.
19. Glatzel P, de Groot FMF, Manoilova O, Grandjean D, Weckhuysen BM, Bergmann U, Barrea R. *Phys. Rev. B: Condens. Matter*. 2005; 72 Art. No. 014117.
20. Robblee JH, Messinger J, Cinco RM, McFarlane KL, Fernandez C, Pizarro SA, Sauer K, Yachandra VK. *J. Am. Chem. Soc.* 2002; 124:7459–7471. [PubMed: 12071755]
21. Dau H, Iuzzolino L, Dittmer J. *Biochim. Biophys. Acta*. 2001; 1503:24–39. [PubMed: 11115622]

22. Kusunoki, M.; Takano, T.; Ono, T.; Noguchi, T.; Yamaguchi, Y.; Oyanagi, H.; Inoue, Y. Photosynthesis: From Light to Biosphere. Mathis, P., editor. Vol. Vol. II. Dordrecht, The Netherlands: Kluwer Academic Publishers; 1995. p. 251-254.

Figure 1.

Left: A schematic representation of the detection scheme. Mn and Fe $K\alpha_1$ and $K\alpha_2$ fluorescence peaks are ~5 eV wide and split by ~11 eV (not shown). The multichannel monochromator with ~1 eV resolution is tuned to the $K\alpha_1$ peak (red). The fluorescence peaks broadened by the Ge detector with 150–200 eV resolution are shown below (blue).¹⁷ Right: The PS II Mn K -edge EXAFS spectrum from the S_1 state sample obtained with a traditional energy-discriminating Ge detector (blue) compared with that collected using the high-resolution crystal monochromator (red). Fe present in PS II does not pose a problem with the high-resolution detector (the Fe edge is marked by a green line). Inset: The inset shows the schematic for the crystal monochromator used in a backscattering configuration.¹²

Figure 2.

Left: The k^3 -weighted Mn EXAFS spectra from PS II samples in S₁ and S₂ states obtained with conventional EXAFS detection (blue) and with a high-resolution spectrometer (range-extended EXAFS, red). Green line at $k = 11.5$ Å⁻¹ denotes the spectral limit of conventional EXAFS imposed by the presence of the Fe *K*-edge. Right: Comparison of FTs of the Mn EXAFS spectra in conventional EXAFS collected up to the Fe edge at 11.5 Å⁻¹ with range-extended EXAFS recorded up to 15.5 Å⁻¹ in the S₁ (green) and S₂ states (black).

Figure 3. Models for the Mn₄Ca cluster (Ca not shown) compatible with the high-resolution Mn EXAFS data with three short 2.7–2.8 Å Mn–Mn distances and one or two longer Mn–Mn distances at ~3.3 Å.

Table 1

One- and Two-Shell Fits of Fourier Peak II from Range-Extended EXAFS Data from the S₁ and S₂ States^a

	$R_{(Mn-Mn)}$ (Å)	N	$\sigma^2 \times 10^3$ (Å) ²	$\sigma \times 10^3$	$\epsilon \times 10^5$
one shell	2.72	1.21	3.0	0.18	0.060
two shells	2.68	0.60	1.0	0.16	0.054
S ₁ State					
$N_1:N_2 = 1:1$	2.77	0.60	1.0		
$N_1:N_2 = 2:1$	2.71	0.87	1.0	0.11	0.035
	2.81	0.44	1.0		
one shell	2.75	1.20	2.6	0.13	0.042
two shells	2.71	0.60	1.0	0.11	0.038
S ₂ State					
$N_1:N_2 = 1:1$	2.79	0.60	1.0		
$N_1:N_2 = 2:1$	2.73	0.84	1.0	0.09	0.031
	2.82	0.42	1.0		

^a N_1 and N_2 are the numbers of interactions for two shells with fixed σ^2 (Debye–Waller parameter); Φ and ϵ^2 are fit-quality parameters; for details see Supporting Information.