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Ether-Directed *ortho*-C–H Olefination with a Pd^{II}/MPAA Catalyst**

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Abstract

Weak coordination is powerful! A Pd^{II}-catalyzed olefination of *ortho*-C–H bonds of arenes directed by weakly coordinating ethers is developed using mono-protected amino acid (MPAA) ligands. This finding provides a method for chemically modifying ethers, which are abundant in natural products and drug molecules.

Keywords

C–H activation; ether; olefination; palladium; weak coordination

Cyclopalladation reactions promoted by strong π -chelation have served as a fruitful platform for studying and developing redox chemistry to establish catalytic cycles for C–H functionalization processes.^[1] However, these well-established C–H activation reactivities have several limitations from the viewpoint of both catalysis and synthetic applications.^[1f] First, the cyclopalladated intermediates are thermodynamically stable and less reactive in the subsequent functionalization step thereby making early discovery of suitable conditions for a diverse range of catalytic reactions difficult. Second, the substrates are typically limited to molecules containing nitrogen-, sulfur- or phosphorus-chelating groups.^[1a] Third, the strongly coordinating directing groups either outcompete ligands for vacant coordination sites or dominate the electronic properties of the metal centers, both of which are not desirable for developing ligand-controlled reactions. In response to these challenges, we and others have recently established weak coordination as a powerful means for directing catalytic C–H activation with Pd^{II}, Rh^{III} and Ru^{II} catalysts.^[2–3] The interplay between a suitable ligand and these types of weak coordination on Pd^{II} centers was also shown to accelerate C–H activation.^[4] Herein, we demonstrate for the first time that mono-protected amino acid ligands (MPAA) promote ether-directed C–H olefination,^[1e] which provides a

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method to functionalize readily available arylethyl ethers to afford novel cinnamate derivatives.

Arylether directed benzylic C–H activation:^[6a]

(1)

Alkylether directed aryl C–H activation:

(2)

Ether is one of the most common functional groups in natural products and drug molecules.^[5] The development of ether directed C–H functionalization reactions would be very useful and important. However, such reports are very rare in literatures,^[6,7] presumably due to the poor coordination of ethers to late transition metal centers. To the best of our knowledge, only one example of ether directed C–H functionalization with Pd^{II} catalyst is documented to date and it is arylether directed benzylic C–H amidation reported by Álvarez and et al. [Eq. (1)].^[6a] While this reaction is an intriguing example of using arylether directing groups to promote activation of benzylic C–H bonds, the use of simple alkylethers to direct aryl C–H activation via a six-membered cyclopalladation represents a distinct and unanswered challenge. Since alkylethers are widespread in natural products and drug molecules,^[5] we set out to develop C–H activation reactions using a combination of weak coordination of the alkylether moiety and ligand acceleration^[2h] to provide a new method for the functionalization of a major class of ethers, namely arylethyl ethers [Eq. (2)].

Taking into consideration the weakly coordinating ability of the ether moiety, we began to develop the olefination of ether **1a** using a weakly coordinating solvent hexafluoroisopropanol (HFIP).^[2h] Thus a mixture of ether **1a** (0.2 mmol), ethyl acrylate (**2a**) (0.3 mmol), 10 mol% Pd(OAc)₂ and Ag₂CO₃ (0.4 mmol) was stirred in 2 mL HFIP under 90 °C for 24 h (Table 1, entry 1). ¹H NMR analysis detected the formation of the mono-olefinated product **3a** in 11% yield. Guided by previous studies on Pd^{II}-catalyzed C–H activation using MPAA ligands,^[2h,4] we screened these ligands and observed significant improvement of the reaction with several ligands (entries 2–4).^[8] The olefinated products were formed in 92% total yield (mono/di = 1.9/1) when Ac-Gly-OH was used (entry 4). Other solvents except CF₃CH₂OH drastically decreased the yields (entries 5–7). Among the oxidants tested, Ag₂CO₃ was shown to be the most effective one, which was consistent with our previous report on Pd^{II}/Pd⁰ catalysis (entries 8–11).^[9] It is worth noting that the use of O₂ as the terminal oxidant was also possible with catalytic amount of Cu(OAc)₂ (entry 12). To improve the mono-selectivity, reaction conditions were further optimized (entries 13–16). Either lowering the reaction temperature or stopping the reaction at lower conversion led to improved mono-selectivity while maintaining good yields (entries 13 and, 14). The mono- and di-olefinated products were easily separated by silica gel chromatography.

To examine the substrate scope of this reaction, we methylated a wide range of readily available arylethanol with MeI to give the corresponding ethers. Ethers derived from primary, secondary and tertiary alcohols were olefinated to give the desired cinnamates in 55–96% yields (**3a–h**) (Table 2). It is worth noting that the corresponding hydroxyl-directed olefination reaction with primary and secondary arylethyl alcohols suffered from partial oxidation of the alcohols to aldehydes and ketones and decomposition of the substrates, resulting in an unproductive reaction.^[4b] Both electron-donating (**3i–k**) and withdrawing groups (**3l–p**) on the arenes were tolerated. Except for a highly reactive arene (**3j**), *meta*-substituted arenes were typically olefinated in greater than 95% mono-selectivities (**3f**, **3k**, and **3o**). Olefination of the methyl ether of the diol also proceeded to give **3q** in 66% total yield. Not surprisingly, the replacement of the methyl group by a benzyl group in the ether substrates resulted in olefination on both arenes to give an inseparable mixture. However, olefination of *meta*-trifluoromethylbenzyl ether proceeded selectively to give the desired product **3r** in 51% yield, suggesting alkylethers other than methyl ethers are also reactive. Moreover, when a hydroxyl group is desired for further functionalization, the methyl ether products can be demethylated with BBr₃ using a literature procedure (Scheme 1).^[10]

The scope of the olefin coupling partners was also examined (Table 3). Olefination of ether **1e** or **1f** with various olefin coupling partners such as , -unsaturated ester, amide, phosphonate and ketone gave corresponding cinnamates in good yields (**5a–5d**). Although the di- or tri-substituted olefins are usually not compatible with *ortho*-C–H olefination,^[4a] modest yields (47%, 34%) were obtained with the ether directing group (**5e**, **5f**).

In summary, we have developed an ether-directed C–H olefination with Pd^{II}/MPAA catalysts, further demonstrating the potential of weak coordination as a useful tool for promoting C–H activation. This reaction can provide a new method for chemically modifying readily available and synthetically useful ethers. We are currently developing the asymmetric variant of this reaction by tuning the chiral MPAA ligands.

Supplementary Material

Refer to Web version on PubMed Central for supplementary material.

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Scheme 1.
Demethylation of the olefinated product.

Table 1

Optimization of reaction conditions.^[a,b]

Entry	Ligand	Solvent	Oxidant	Yield (%) [mono/di]
1	–	HFIP	Ag ₂ CO ₃	11 [1/0]
2	Ac-Ile-OH	HFIP	Ag ₂ CO ₃	38 [11.7/1]
3	Ac-Ala-OH	HFIP	Ag ₂ CO ₃	73 [4.6/1]
4	Ac-Gly-OH	HFIP	Ag₂CO₃	92 [1.9/1]
5	Ac-Gly-OH	<i>tert</i> -Amyl-OH	Ag ₂ CO ₃	trace
6	Ac-Gly-OH	CF ₃ CH ₂ OH	Ag ₂ CO ₃	69 [6.7/1]
7	Ac-Gly-OH	DCE	Ag ₂ CO ₃	trace
8	Ac-Gly-OH	HFIP	AgOAc	86 [1.7/1]
9	Ac-Gly-OH	HFIP	Ag ₂ O	4 [1/0]
10	Ac-Gly-OH	HFIP	Cu(OAc) ₂	13 [1/0]
11	Ac-Gly-OH	HFIP	O ₂ (1 atm)	7 [1/0]
12 ^[c]	Ac-Gly-OH	HFIP	O ₂ /Cu(OAc) ₂	41 [20/1]
13 ^[d]	Ac-Gly-OH	HFIP	Ag₂CO₃	90 [4.3/1]
14 ^[e]	Ac-Gly-OH	HFIP	Ag ₂ CO ₃	73 [5.6/1]
15 ^[f]	Ac-Gly-OH	HFIP	Ag ₂ CO ₃	90 [2.6/1]
16 ^[g]	Ac-Gly-OH	HFIP	Ag ₂ CO ₃	71 [3.7/1]

^[a] Conditions: **1a** (racemic, 0.2 mmol), **2a** (0.3 mmol), Pd(OAc)₂ (10 mol%), ligand (20 mol%), oxidant (0.4 mmol), solvent (2 mL), 90 °C, 24 h, in a 35 mL sealed tube.

^[b] total yield of (*o*)-mono- and (*o*, *o'*)-di-olefinated products determined with CH₂Br₂ as internal standard, ratios of mono/di are in brackets.

^[c] O₂ (1 atm), Cu(OAc)₂ (0.1 mmol), 36 h.

^[d] 80 °C.

^[e] 12 h.

^[f] 18 h.

^[g] **2a** (0.2 mmol). Ac = acetyl, Ile = L-isoleucine, Ala = L-alanine, Gly = glycine, *tert*-Amyl-OH = *tert*-amyl alcohol, DCE = 1,2-dichloroethane, HFIP = hexafluoroisopropanol.

Table 2

Pd^{II}-catalyzed ortho-C–H olefination of ethers.^[a,b]

 3a_{mono} , 70% 3a_{(o,o')di} , 16%	 3b_{mono} , 60% 3b_{(o,o')di} , 25% ^[c]	 3c , 55% ^[c]
 3d_{mono} , 53% 3d_{(o,o')di} , 10% ^[d]	 3e , 76% ^[d,e]	 3f , 82% ^[e]
 3g_{mono} , 72% 3g_{(o,o')di} , 24% ^[c]	 3h_{mono} , 62% 3h_{(o,o')di} , 32% ^[e]	 3i , 56% ^[e]
 3j_{mono} , 46% 3j_{(o,o')di} , 25% ^[f]	 3k , 84% ^[c,e]	 3l , 60% ^[c,e]
 3m , 50% ^[c]	 3n , 51% ^[c]	 3o , 65% ^[g]
 3p_{mono} , 58% 3p_{(o,o')di} , 17% ^[g]	 3q_{mono} , 54% 3q_{(o,o')di} , 12% ^[c]	 3r , 51%

^[a] Conditions: **1** (racemic, 0.2 mmol), **2a** (0.3 mmol), Pd(OAc)₂ (10 mol%), Ac-Gly-OH (20 mol%), Ag₂CO₃ (0.4 mmol), HFIP (2 mL), 80 °C, 24 h, in a 35 mL sealed tube.

[b] Isolated yields of *ortho* olefinated products; *meta* or *para* olefinated regioisomers are also detected as minor products; for **1c**, **1m** and **1r**, (*o,o'*)-diolefinated products were observed but not isolated; see SI for details.

[c] 48 h.

[d] **2a** (0.25 mmol).

[e] 90 °C.

[f] 100 °C.

[g] See SI for condition variations.

Table 3

Scope of olefins.^[a,b]

^[a] Conditions: **1** (racemic, 0.2 mmol), **2** (0.25 mmol), Pd(OAc)₂ (10 mol%), Ac-Gly-OH (20 mol%), Ag₂CO₃ (0.4 mmol), HFIP (2 mL), 90 °C, 24 h, in a 35 mL sealed tube.

^[b] Isolated yields of *ortho* olefinated products; *meta* or *para* olefinated regioisomers are also detected as minor products; see SI for details.

^[c] **2** (0.4 mmol).

^[d] 48 h.

^[e] Pd(OAc)₂ (15 mol%) and Ac-Gly-OH (30 mol%) used.

$[\alpha]_D^{25}$ only one diastereomer isolated; relative configuration of **5e** not determined.

$[\alpha]_D^{25}$ 100 °C.