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Co-Assembled Cytotoxic and Pegylated Peptide Amphiphiles Form Filamentous Nanostructures with Potent Anti-Tumor Activity in Models of Breast Cancer

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Abstract

Self-assembled peptide amphiphiles (PAs) consisting of hydrophobic, hydrogen bonding, and charged hydrophilic domains form cylindrical nanofibers in physiological conditions and allow for the presentation of a high density of bioactive epitopes on the nanofiber surface. We report here on the use of PAs to form multifunctional nanostructures with tumoricidal activity. The combination of a cationic, membrane-lytic PA co-assembled with a serum-protective, pegylated PA was shown to self-assemble into nanofibers. Addition of the pegylated PA to the nanostructure substantially limited degradation of the cytolytic PA by the protease trypsin, with an eight-fold increase in the amount of intact PA observed after digestion. At the same time, addition of up to 50% pegylated PA to the nanofibers did not decrease the *in vitro* cytotoxicity of the cytolytic PA. Using a fluorescent tag covalently attached to PA nanofibers we were able to track the biodistribution in plasma and tissues of tumor bearing mice over time after intraperitoneal administration of the nanoscale filaments. Using an orthotopic mouse xenograft model of breast cancer, systemic administration of the cytotoxic pegylated nanostructures significantly reduced tumor cell proliferation and overall tumor growth, demonstrating the potential of multifunctional PA nanostructures as versatile cancer therapeutics.

Keywords

Peptide amphiphile; cancer therapy; nanofiber; breast cancer; Pegylation

Nanotechnology offers great potential to improve conventional cancer therapies through enhancement of both drug delivery and antitumor activity.^{1, 2} Nanoparticles in the size range of tens to hundreds of nanometers can passively accumulate in tumors due to their leaky vasculature and poor lymphatic clearance, referred to as the enhanced permeability and

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retention (EPR) effect.³ In addition to size, the shape of the nanostructures plays an important role in tumor and cell uptake. Examples of non-spherical cell penetrating particles existing in nature include the filamentous viruses such as the tobacco mosaic virus and the potato virus X.⁴ Inspired by these examples from biology, Discher *et al.* have shown that cylindrical micelles have longer circulation times relative to spherical controls.⁵ Maximum tolerated doses also increase with cylinders compared to spherical block copolymers.⁶ Furthermore, particles with higher aspect ratios have shown increased uptake into cancer cells.⁷

Lipidated peptides with certain sequences were identified by our group as a broad class of small molecules that self-assemble into cylindrical nanostructures.^{8,9} This family of peptide amphiphiles (PAs) are synthesized using solid-phase peptide synthesis by attaching an alkyl tail to a short peptide sequence that is typically composed of a β -sheet forming region that drives cylinder formation and a bioactive signaling epitope, which may include charged residues to promote water solubility. PAs spontaneously self-assemble in aqueous solution into high aspect-ratio filaments through hydrophobic collapse of alkyl segments to form the core of the nanofiber and by hydrogen bonding among peptide segments to create β -sheet structures. The resultant nanostructures can be designed to display bioactive peptide epitopes at high density on their surfaces.¹⁰ PAs have been investigated *in vivo* as therapies in regenerative medicine, including central nervous system repair, bone and cartilage regeneration, angiogenesis for hind limb ischemia and myocardial infarction.^{11–16} In addition, our group has shown previously that treating mice bearing orthotopic human breast tumors with soluble PA nanostructures encapsulating hydrophobic chemotherapy in the core of the fibers have reduced tumor burden.¹⁷

The tunable nature of peptide-based therapeutics allows for the precise targeting of extracellular or intracellular components to induce cancer cell death.^{18,19} Additional potential advantages of therapeutic peptides include complete biodegradability into benign metabolites and ease of synthesis. Peptides have shown promise as potential cancer therapeutics in a variety of models. For example, the α -helical cationic peptide sequence, (KLAKLAK)₂, was shown to interact strongly with lipid membranes.²⁰ This peptide, commonly conjugated to cell-penetrating or targeting peptides, lyses either plasma or mitochondrial membranes, and has shown anti-tumor activity *in vivo*.^{21–23}

Our group has previously described a PA that contains the membrane-lytic, highly cationic (KLAKLAK)₂ sequence as the bioactive component.²⁴ This PA (referred to here as KLAK PA) has the sequence C₁₆A₄G₃(KLAKLAK)₂ and assembles into cylindrical nanostructures capable of disrupting cell membranes.²⁴ An interesting characteristic of KLAK PA is that the delivery vector is also the therapy, whereas most other nanoscale delivery vehicles encapsulate traditional chemotherapies, or have incorporated biologic therapies into non-biodegradable nanoparticles.²⁵ We previously showed that KLAK PA nanostructures were more cytotoxic towards transformed cells compared with untransformed cells.²⁴ We speculate that this effect is due to the increased amount of negative charge on the surface of cancer cells.²⁶ KLAK PA nanoparticles may be passively targeted to tumors *in vivo* through the EPR effect, but it is possible that the KLAK epitope mediates active targeting: The R/KXXR/K epitope has shown potential for tumor targeting, which suggests that the membrane lytic KLAK sequence could also function to increase tumor uptake.²⁷

Because peptides can be quickly degraded after systemic delivery, ongoing work has investigated limiting proteolysis in circulation.^{28,29} A common strategy is to synthesize peptides using D-enantiomers that are more resistant to endogenous protease degradation.^{29,30} Another strategy to improve the therapeutic-delivery vehicle is to functionalize the surface with polymers that protect from proteolysis, lessen immune

response, and improve circulation time. Polyethylene glycol (PEG) is widely known to reduce the metabolism and elimination of pegylated molecules in the bloodstream.³¹ For instance, the covalent attachment of pegylated-lipid micelles to short peptide sequences has been shown to decrease the rate of peptide proteolysis³². Doxil, a pegylated liposome that contains doxorubicin, enhances the circulation time compared to free doxorubicin, and has a blood half-life of over 48 h.³³ Variations in the molecular weight of PEG on polymersomes³⁴ and density of PEG on polymeric micelles³⁵ have been shown to induce drastic changes in circulation time and biodistribution. Furthermore, pegylation of the protein interferon has improved circulation time and efficacy against Hepatitis C.³⁶ Pegylation of self-assembling peptides has been described previously,^{37, 38} and Hamley and coworkers have attached a PEG chain to a beta-sheet forming peptide that formed long, one-dimensional fibers.³⁹ However, the *in vivo* impact of pegylation on self-assembled peptide bioactivity has not yet been reported. Here we demonstrate the use of a co-assembled combination of pegylated PA (PEG PA) and the cytotoxic KLA PA to create an effective antitumor therapy.

Results and Discussion

TEM and SAXS characterization

The chemical structures of KLA PA and PEG PA are shown in Figure 1A. Conventional and cryogenic transmission electron microscopy (TEM) were performed on the KLA PA alone, PEG PA alone, and the KLA/PEG PA mixture. KLA PA appears to form short, cylindrical fibers that are roughly 10 nm in diameter and several hundred nm in length according to cryo-TEM (Figure 1B). Conventional TEM shows the presence of both short, bundled fibers, as well as longer single fibers (Figure 1C). Overall, according to both the cryo and conventional TEM, KLA PA forms relatively short fibers. PEG PA, on the other hand, forms fibers that are similar in diameter, but significantly longer in length (See Supporting Information for details, specifically Supplemental Figure S1). Three glutamic acids were included in the PEG PA structure to utilize electrostatic attraction during co-assembly.⁴⁰ The two PAs were mixed in a 1:1 molar ratio to produce a co-assembled structure. The PEG and KLA PAs were first dissolved and mixed in hexafluoroisopropanol (HFIP), lyophilized, and resuspended in PBS. HFIP, known to disrupt hydrogen bonding, was used to destabilize the assembled structures and enhance co-assembly of the two PAs. Cryo-TEM of the mixture shows that the longer fibers are the dominant morphology (Figure 1D) and the short, bundled structures that were observed in conventional TEM of KLA PA alone were no longer visible in the co-assembled structure (Figure 1E). The mixed KLA-PEG PAs form longer structures with a similar morphology compared to the PEG alone. KLA PA synthesized using D-enantiomers ("D-KLA") showed similar morphologies both with and without PEG PA by conventional TEM, suggesting that the assembled structures were independent of the chirality of KLA PA (Supplemental Figure S2). In both cases, short nanofibers were observed for KLA PA alone, while the mixing of PEG PA and KLA PA appeared to drastically increase the nanofiber length. We speculate that the difference in observed nanofiber lengths is due to the co-assembly of oppositely charged molecules, which decreases the electrostatic barrier to the formation of long cylindrical structures.

Small angle X-ray scattering (SAXS), a technique that can be used to determine the size and shape of nanostructures in solution, confirmed the presence of cylindrical structures in solution. We sought to verify that there was no visible change in supramolecular structure when KLA PA and PEG PA molecules were mixed compared to solutions prepared only with PEG PA. Both the KLA-PEG PA mixture and PEG PA alone had a slope of -1 in the low-*q* range, which is indicative of cylinders in solution. In the cases of KLA PA without PEG PA, both KLA and D-KLA PA had slopes of approximately -2 in the low *q*-range,

which could be the result of bundling of multiple nanostructures together (Supplemental Figure S3). However, in order to accurately quantify the size of these nanostructures, core-shell cylinder models were used for each of the PAs. The KLAKE-PEG PA mixture was fit to a polydisperse core-shell cylinder model (Figure 1F). The total radius, which includes core and shell thickness, was found to be 4.9 nm according to the fit, which is similar to the 4.3 nm fit radius of PEG PA alone (Figure 1G). We suggest that since the PEG chain is well hydrated, its electron density is indistinguishable from the solvent, causing the radius to appear smaller than expected. Thus, combining the results from TEM and SAXS, two possibilities exist: either KLAKE PA is incorporated into the same nanostructure as PEG PA without significant disruption of fiber morphology, or the PEG PA nanostructure is dominating the scattering and microscopy images. The observed increase in nanostructure radius from 4.3 to 4.9 nm upon addition of the KLAKE PA lends support to the possibility that co-assembly of both molecules indeed occurs in the observed supramolecular filaments.

DOSY

Diffusion ordered spectroscopy (DOSY) measures the exponential attenuation of NMR signals while applying a pulsed-field gradient to determine the diffusion coefficient of each chemical species.⁴¹ It has been used to distinguish a variety of nanostructures based on diffusion coefficients, including carbon nanotubes.⁴² To assess changes in fiber diffusion constants, we performed DOSY NMR on both KLAKE PA alone and the mixture of KLAKE and PEG PAs. Because individual spectral peaks can be monitored by DOSY, one can observe changes in diffusion of a specific molecule in a mixture of two or more components. DOSY results show a shift in the diffusion coefficient of the KLAKE PA after addition of PEG PA (Figure 2A). All of the peaks in the blended sample with KLAKE and PEG PAs correspond to the same diffusion coefficient, suggesting that both PAs are on the same nanostructure. A slight upfield shift in the KLAKE PA alone peak at 1.2 ppm may be the result of more flexible peptides at the periphery of the nanostructure. One possible explanation for this effect is that the α -helical nature of the KLAKE sequence decreases the interaction with surrounding molecules, thus increasing the translational diffusion coefficient. Additionally, comparing the diffusion coefficients by DOSY for KLAKE PA with and without PEG PA showed a change in size after the addition of PEG PA, suggesting an interaction between the two molecules. To quantify this difference, we used the attenuation of the amplitude of an NMR signal from KLAKE PA, which decays exponentially at a rate that is proportional to the diffusion coefficient. The peak at 2.83 ppm, which was attributed to KLAKE PA using the ¹H 1-D spectra (Supplemental Figures S4 and S5), was fitted to an exponential decay (Figure S6) and showed the same change in diffusion coefficient (Figure 2B). The KLAKE PA alone diffusion coefficient was found to be $0.69 \times 10^{-10} \text{ m}^2/\text{s}$, whereas when PEG PA molecules were mixed with KLAKE PA, the diffusion coefficient was lowered to $0.54 \times 10^{-10} \text{ m}^2/\text{s}$. We attribute this shift in diffusion coefficient to the incorporation of KLAKE PA into a co-assembled nanostructure with PEG PA, which corresponded with the change in size and morphology as determined by TEM and SAXS. These data collectively are indicative of co-assembly of the two PA molecules in the same nanostructure.

Enzymatic degradation

The addition of PEG PA to the KLAKE PA fibers was intended to create a protective corona around the assembled nanostructure, which would limit non-specific adsorption of proteins and reduce proteolysis. Protection from proteolysis is an important characteristic for effective peptide nanostructures in serum. To test the effect of including PEG PA on enzymatic degradation, we exposed the co-assembled structure of KLAKE PA and PEG PA to trypsin for one hour. Using this assay, we also determined an optimal amount of PEG PA to be co-assembled with KLAKE PA. Trypsin, which cleaves peptides on the C-terminal side of positively-charged residues, was added to solutions of 250 μM KLAKE PA, which

contained varying concentrations of co-assembled PEG PA. Because PEG PA has no cationic amino acids, trypsin will only cleave peptides on KLAK PA. After removal of trypsin, concentrations of intact KLAK PA were measured by liquid chromatography-mass spectroscopy (LC-MS). Significantly decreased levels of degradation of KLAK PA were observed for solutions that contained at least 50 mol % PEG PA relative to 25 mol % PEG PA and 0 mol % PEG PA (Figure 3). As an additional control, KLAK PA and PEG PA were pre-assembled in separate solutions prior to mixing and the addition of trypsin. In this case, the presence of 50% (molar) PEG PA had no effect on degradation of KLAK PA and there was no statistically significant difference relative to the 0 mol % PEG PA control. These degradation results strongly suggest that KLAK PA and PEG PA are co-assembled in the same nanostructure. For further biological testing, 50 mol % of PEG PA (a 1:1 molar ratio of KLAK PA to PEG PA) was used because additional PEG PA did not significantly reduce proteolysis of KLAK PA.

***In vitro* cytotoxicity**

We tested the cytotoxicity of KLAK PA on human breast cancer cells to determine if mixing with PEG PA affected the ability of KLAK PA to kill cells through membrane lysis.²⁴ *In vitro* results show that the mixture of the PEG PA and KLAK PA has no negative effect on cytotoxicity relative to the activity of KLAK PA alone (Figure 4A). Flow cytometry was used with DAPI staining to quantify membrane permeabilization of MCF7 breast cancer cells as a function of time (Figure 4B). Mitochondrial membranes also show loss of integrity (Figure 4C) with co-assembled PA treatment. These results indicate that addition of 50 mol % PEG PA does not produce any change in the rate of membrane lysis by KLAK PA. As expected, PEG PA alone showed no cytotoxicity up to a concentration of 1 mM. Thus, while mixing of PEG PA with KLAK PA does appear to limit enzyme access to the cationic peptides of KLAK PA, it does not appear to inhibit any interaction of KLAK PA with cellular membranes. KLAK PA and D-KLAK PA showed no significant differences in cytotoxicity (Supplemental Figure S7).

PA plasma and tissue biodistribution

We attempted to quantify the tissue specific and plasma PA concentration over time by covalently attaching the near IR dye Alexa 700 to the KLAK PA monomer. Following IP administration, blood from each mouse was collected at earlier time points by tail bleed, followed by intracardiac puncture as part of a terminal procedure at which time tissues were harvested. The plasma fluorescence signal of KLAK-PEG PA treated animals rose more rapidly relative to KLAK PA (Fig. 5A), and more signal was seen in all tissues at 1 h (Figure 5B, C), suggesting it is more rapidly mobilized from the peritoneal compartment than KLAK PA alone. Fluorescence of both KLAK PA alone and KLAK-PEG PA mixtures was predominantly localized to liver and kidney (Figure 5B, C). Fluorescence observed in lung was minimal, speaking to the solubility of the PA fibers (data not shown). Area under the curve calculations indicated that plasma (Figure 5A) bioavailability is improved by the addition of PEG PA. Nevertheless, the observed plasma half-lives are on the order of several hours both with and without the addition of PEG PA, which is better than typical times observed for free peptides, suggesting the PA fibers are intrinsically resistant to degradation and/or elimination.^{28, 29} A limitation of using a fluorescence-based approach to quantify *in vivo* PA distribution is that the dye signal may be from a mixture of intact and degraded KLAK PA; in addition, the dye may alter the absorption, distribution, and elimination of the fibers. Because of the membrane lytic action of KLAK PA, off-target destruction of circulating blood cells was a theoretical concern. However, 24 h after PA administration no significant difference in white or red blood cell number was observed (Supplemental Table S1). Among the many metabolic parameters measured in the treated mice, alkaline phosphatase levels were significantly lower in the KLAK and KLAK/PEG PA treated

animals compared with PBS treated controls, and the KLAK PA treated animals had significantly higher blood glucose levels (Supplemental Table S2).

***In vivo* anti-tumor activity**

An orthotopic mouse breast cancer model was used to measure the anti-tumor activity following intraperitoneal injection of KLAK PA. Treatment with both KLAK PA alone and KLAK/PEG PA significantly suppressed human MDA-MB-231 breast cancer growth in the mice relative to PBS alone (Figure 6A). There was a trend towards smaller tumors in the mice treated with the KLAK/PEG PA compared with those treated with the KLAK PA alone, but this difference was not statistically significant (Figure 6A). No significant change in average mouse weights were noted during treatment (Figure 6B), indicating that treatment was generally well-tolerated. Immunohistological staining for bromodeoxyuridine (BrdU) revealed a trend towards a reduction in the percentage of proliferating cells with all KLAK PA treatments that was statistically significant only when KLAK PA was given co-assembled with PEG (Figure 6C). Our BrdU staining results corroborate a previous report demonstrating that nanoparticle delivery of membrane lytic peptides targets proliferating cancer cells.²⁵ Tumors of D-KLAK PA treated mice trended smaller (Supplemental Figure S8), but unlike KLAK PA, only D-KLAK/PEG PA treatment resulted in statistically significant tumor inhibition (Supplemental Figure S8C). Because KLAK and D-KLAK PA performed similarly, we speculate that the clearance of KLAK PA is not determined by the stereochemistry of the molecule. This phenomenon has been observed when L-Lysine and D-Lysine were conjugated to monoclonal antibodies and injected systemically into mice, similar changes in renal clearance were observed.⁴³ An alternate possibility is that tumor targeting and anti-tumor action happens more rapidly than the PAs are cleared.

Conclusions

Self-assembled peptide amphiphiles offer a new and versatile platform as nanoscale cancer therapeutics. Here we have demonstrated that a membrane-lytic peptide amphiphile co-assembles with a pegylated peptide amphiphile into cylindrical nanostructures to create filaments with greater proteolytic resistance and similar *in vitro* anti-breast cancer cytotoxic activity as those based only the lytic peptide. Furthermore, systemic delivery of these nanoscale filaments decreased breast cancer growth over four weeks of administration in a mouse xenograft model and appeared to be well-tolerated by the animals. This work highlights the potential of PA nanostructures as a biodegradable and customizable tumoricidal nanotechnology platform. We expect that the anti-tumor activity of PA nanostructures may be further enhanced through the use of chemotherapeutic drugs safely encapsulated in the nanofiber core. Furthermore, systemic toxicity may be reduced and tumor activity enhanced by the co-assembly approach described here but using selective tumor-homing peptides.

Materials and Methods

PA synthesis and materials characterization

PAs were synthesized using standard Fmoc solid-phase synthesis conditions. Coupling reactions were performed using Fmoc-amino acids (4 equiv), HBTU (3.95 equiv) and diisopropylethylamine (DIEA) (6 equiv) in dimethylformamide (DMF). Synthesis of L-KLAK PA and D-KLAK PA was achieved using L-amino acids and D-amino acids, respectively. For the hydrophobic tail of PEG PA (sequence K(C₁₂)A₄G₃E₃-PEG), lauric acid was attached to the εamine of a lysine, which was deprotected by selective removal of the Mtt (Mtt = 4-methyltrityl) group using 4% trifluoroacetic acid (TFA) + 5% triisopropylsilane (TIPS) in CH₂Cl₂. CH₃O-PEG-COOH (MW 2000) was prepared as

previously described⁴⁴ and attached on resin at the N-terminus of the PEG PA. The alkyl tail of the KLAK PA was formed by reacting the N-terminus with palmitic acid (4 equiv), HBTU (3.95 equiv) and DIEA (6 equiv) in DMF. Following cleavage using a TFA/TIPS/H₂O mixture (95:2.5:2.5), PAs were purified by high-performance liquid chromatography (HPLC).

Purification by preparative-scale HPLC was carried out on a Varian Prostar 210 HPLC system, eluting with 2% acetonitrile (ACN) to 100% ACN in water on a Phenomenex C18 Gemini NX column (150 × 30 mm) with 5 μm pore size and 110 Å particle size. 0.1% NH₄OH or 0.1% trifluoroacetic acid, for acidic or basic PAs, respectively, were added to both mobile phases to aid PA solubility. Product-containing fractions were confirmed by ESI mass spectrometry (Agilent 6510 Q-TOF LC/MS), combined, and lyophilized after removing ACN by rotary evaporation. Amino acid analyses were performed by Commonwealth Biotechnologies (Richmond, VA).

Co-assembly of the PAs was achieved by dissolving the KLAK PA and PEG PA separately in hexafluoroisopropanol (HFIP), an organic solvent known to disrupt hydrogen bonds,^{45–47} and then mixed together for at least 15 minutes. Samples were lyophilized to dryness to form a powder, as previously reported by our group.⁴⁸ After lyophilization in HFIP, samples were dissolved in water, aliquoted, and lyophilized again. A KLAK PA-AlexaFluor 700 conjugate was synthesized by reacting a 5 molar excess of KLAK PA with NHS-AlexaFluor 700 (Invitrogen). KLAK PA was dissolved in DMSO at a concentration of 0.5 mM with 1% triethylamine (TEA), and AlexaFluor 700 was added drop-wise to a stirring solution and reacted overnight (the reaction was confirmed using ESI mass spectrometry). The KLAK PA-AlexaFluor 700 conjugate was dialyzed in H₂O overnight to remove any unreacted dye and subsequently lyophilized. KLAK PA-AlexaFluor 700 and PEG PA mixtures were co-assembled in HFIP, as described above.

Specimens for conventional transmission electron microscopy (TEM) were prepared by drop-casting samples on carbon type B copper grids (Ted Pella) followed by staining with a 2% uranyl acetate aqueous solution. Cryogenic TEM (cryo-TEM) specimens were prepared using an FEI Vitrobot by blotting in 95% humidity and subsequently plunging grids into liquid ethane. Images were taken for both conventional and cryo-TEM using a JEOL 1230 transmission electron microscope operating at 100 keV equipped with a Gatan camera.

Small angle X-ray scattering (SAXS) experiments were performed at the Advanced Photon Source, Argonne National Laboratory. The X-ray energy (15 keV) was selected using a double-crystal monochromator, and the SAXS CCD camera was offset in order to achieve a wide range of scattering angles. Samples were dissolved at a concentration of 2 mM and placed in 1.5 mm quartz capillary tubes. The typical incident X-ray flux on the sample was $\sim 1 \times 10^{12}$ photons/s with a 0.2 × 0.3 mm² collimator, estimated by a He ion channel, and samples were irradiated for 5 s. The 1D scattering profiles were obtained by radial integration of the 2D patterns, with scattering from the capillaries subtracted as background. Scattering profiles were then plotted on a relative scale as a function of the scattering vector $q = (4\pi/\lambda) \sin(\theta/2)$, where θ is the scattering angle.

1H-Diffusion Ordered Spectroscopy (DOSY) was performed using a Bruker Avance 600 MHz spectrometer at ambient temperature. For these experiments samples were dissolved at a constant total concentration of PA (5 mM KLAK PA alone or 2.5 mM KLAK PA and 2.5 mM PEG PA in the mixed case) in 99.9% D₂O (Sigma), and 32 points were measured with a 7 μs 90 degree pulse. Diffusion data were processed and analyzed using DOSY Toolbox.⁴⁹

***In vitro* enzyme degradation kinetics**

Enzyme degradation was performed using trypsin, which was conjugated to agarose beads (Sigma). KLAK PA alone or KLAK PA with PEG PA (n=3 for each group) were mixed in PBS for one hour with 2 units of trypsin under constant agitation at room temperature. The samples were subsequently centrifuged at 500 g for 30 seconds, and the supernatant was removed from the trypsin beads. Samples were diluted 10 fold into a solution with an internal standard, C₁₆V₃A₃K₃-(OEG)₅-NH₂, with a concentration that was held constant for all injections. LC/MS was performed, and the total ion counts were quantified for intact KLAK PA and compared to undigested KLAK PA controls.

***In vitro* breast cancer cytotoxicity**

The *in vitro* growth of cell cultures was measured using the CellTiter 96® Aqueous One Solution Cell Proliferation Assay (Promega, Madison WI). The Cell Titer Assay is a 3-(4,5-dimethylthiazol-2-yl)-5-(3-carboxymethoxyphenyl)-2-(4-sulfophenyl)-2H-tetrazolium (MTS)-based assay and was used according to the supplier's instructions. For each well to be assayed in a 96-well plate containing 100 µl of media per well, 20 µl of the CellTiter96® Aqueous One Solution was added. The plate was incubated for 1–3 h at 37 °C, and absorbance was read using a Molecular Devices microplate reader (490 nm).

Membrane permeability assays

Cells were labeled with 50 nM Mitotracker Red (Invitrogen) to stain mitochondrial membranes according to the manufacturer's instructions. After trypsinization, cells (1 × 10⁶ cells/ml) were resuspended in PBS and 100 ng/ml DAPI was added. Fluorescence intensity at 25° C was measured using a DakoCyomation CyAn.

***In vivo* orthotopic breast cancer model**

1 × 10⁶ MDA-MB-231 human breast ductal carcinoma cells were injected intraductally into the 4th mammary fat pads of 4–5-week old female athymic nude mice (Harlan Sprague-Dawley, Madison, WI) to establish orthotopic xenograft tumors. Four weeks after tumor implantation, mice were randomized into treatment groups (8 mice per group) and administered either PBS vehicle (Sigma-Aldrich), KLAK PA 14 mg/kg/dose, or KLAK PA 14 mg/kg/dose encapsulated with 1 molar equivalent of PEG PA by intraperitoneal (IP) injection twice per week for six total injections. IP administration of nanoparticles has been shown to cause a delay in the rise of circulating nanoparticle concentration relative to intravenous (IV) administration, however in preliminary experiments we found similar levels of fluorescent PA signal in mouse mammary tumor xenografts following either IP or IV administration (data not shown), which given the lower success rate of mouse tail vein injections led us to chose IP treatments for our experiments.⁵⁰ Tumors were measured after each injection with Vernier calipers, and tumor volume was calculated using the equation: tumor volume (mm³) = (length × width²) × π/6 as previously described.⁵¹ All animal experiments were conducted under protocols approved by the Animal Care and Use Committee of Northwestern University.

PA plasma kinetics and tissue biodistribution

We determined the fate of KLAK PA-AlexaFluor 700 either alone or co-assembled with PEG PA in athymic nude mice unilaterally bearing MDA-MB-231 tumors as described above. KLAK PA was injected at a concentration of 20 mg/kg (1.8 mM) with 2.5 µg of conjugated AlexaFluor 700 dye. For the KLAK PA with PEG PA, the concentration of KLAK and dye were kept constant, and 27 mg/kg (1.8 mM) of PEG PA was added. Blood was collected either as a terminal procedure *via* cardiac puncture or by tail vein bleeds into heparinized capillary tubes. Following centrifugation at 10,000 RPM for 10 min, a

Molecular Devices SpectraMax M5 plate reader was used for quantification of plasma fluorescence. Mice were euthanized at the specified time points and tissues were harvested and stored in 4% paraformaldehyde for subsequent fluorescence quantification using a Caliper Life Sciences IVIS Spectrum Bioluminescence System. Blood collected from animals 24 h following PA or PBS administration was sent to RADIL for metabolic profiling and complete blood count.

Statistical analysis

Statistical significance was assessed using Graph-Pad Prism 5.0b software using the appropriate statistical test: two-way ANOVA with Bonferroni post-tests for *in vivo* tumor cytotoxicity and peripheral smear/metabolic profiling studies, one-way ANOVA with Dunnett post-tests for tissue fluorescence studies, or one-way ANOVA with Bonferroni post-test for the enzyme degradation studies.

Supplementary Material

Refer to Web version on PubMed Central for supplementary material.

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Figure 1. PA Characterization

(A) Chemical structure of “KLAK PA” with sequence palmitoyl- $A_4G_3(KLAKLAK)_2$ and “PEG PA” with the sequence PEG₂₀₀₀- $E_3G_3A_4K(C_{12})$. Cryo-TEM of KLAK PA alone (B) KLAK with PEG (D) show a significant difference in average length. Conventional TEM images show fiber formation for both KLAK alone (C) and KLAK PA with PEG PA (E). SAXS confirms the presence of cylindrical structures in solution for both KLAK mixed with PEG PA (F) and PEG PA alone (G).

Figure 2. DOSY of Co-Assembled Structures

(A) Overlaid DOSY results for KLAK PA alone (red) and KLAK with PEG PA show a shift in the diffusion coefficient. (B) Using the KLAK PA peak at 2.83 ppm, the KLAK PA alone diffusion coefficient shows an increase upon mixing with PEG PA.

Figure 3. Enzymatic Degradation

(A) The percentage of intact KLAK PA, as measured by LC/MS, increases with increasing concentrations of PEG PA. The closed arrow points to the KLAK PA peak, and the open arrow points to the internal standard, kept constant in all runs. (B) Quantification of intact KLAK PA. The addition of PEG PA prevents trypsin degradation. The 50(◆) bar represents a solution of KLAK and PEG PA that were pre-assembled separately prior to mixing and exposing to trypsin. (***) $p < 0.001$, (*) $p < 0.05$.

Figure 4. *In Vitro* Cytotoxicity

(A) MDA-MB-231, MCF7, and SKBR3 human breast cancer cells were treated as indicated for 24 h and viability was quantified using a MTS assay. MCF7 human breast cancer cells were treated as indicated and intracellular DAPI (B) and Mitotracker red (C) fluorescence was determined by flow cytometry to assess the plasma and mitochondrial membrane integrity.

Figure 5. PA Plasma and Tissue Biodistribution

(A) Quantification of plasma fluorescence, n=6 per time point. (B) Tissue fluorescence following intraperitoneal administration of KLAK PA covalently modified with Alexa 700 to tumor bearing mice; tissues from the 6 h time point are shown. (C), Quantification of fluorescence as average radiant efficiency (n=6 animals per time point) comparing the signal in tissues from animals treated with KLAK PA alone or KLAK/PEG PA; the dashed line represents background fluorescence observed in control untreated animals. Statistical significance in all panels as measured by Student's t test is noted.

Figure 6. *In Vivo* Anti-Tumor Activity

(A) The growth of MDA-MB-231 human breast cancer orthotopic tumors is inhibited by intraperitoneal treatment (inverted arrows) of KLAK PA nanostructures. Both the KLAK and KLAK/PEG PA treated tumors were statistically smaller as determined by 2-way ANOVA. (B) Mouse weights across groups were similar suggesting the nanostructure treatment was well tolerated. (C) BRDU staining as a measure of cellular proliferation was statistically lower by 1-way ANOVA in the L-KLAK PA treated tumors when co-assembled with PEG PA.