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The Cys319 loop modulates the transition between dehydrogenase and hydrolase conformations in IMP dehydrogenase

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Abstract

X-ray crystal structures of enzyme-ligand complexes are widely believed to mimic states in the catalytic cycle, but this presumption has seldom been carefully scrutinized. In the case of *Trichomonas foetus* inosine 5'-monophosphate dehydrogenase (IMPDH), ten structures of various enzyme-substrate-inhibitor complexes have been solved. The Cys319 loop is found in at least three different conformations, suggesting that its conformation changes as the catalytic cycle progresses from the dehydrogenase step to the hydrolase reaction. Alternatively, only one conformation of the Cys319 loop may be catalytically relevant while the others are off-pathway. Here we differentiate between these two hypotheses by analyzing the effects of Ala substitutions at three residues of the Cys319 loop, Arg322, Glu323 and Gln324. These mutations have minimal effects on the value of k_{cat} (≤ 5 -fold) that obscure large effects (>10 -fold) on the microscopic rate constants for individual steps. These substitutions increase the equilibrium constant for the dehydrogenase step, but decrease the equilibrium between open and closed conformations of a mobile flap. More dramatic effects are observed when Arg322 is substituted with Glu, which decreases the rates of hydride transfer and hydrolysis by factors of 2000 and 130, respectively. These experiments suggest that the Cys319 loop does indeed have different conformations during the dehydrogenase and hydrolase reactions as suggested by the crystal structures. Importantly, these experiments reveal that the structure of the Cys319 loop modulates the closure of the mobile flap. This conformational change converts the enzyme from a dehydrogenase into hydrolase, suggesting that the conformation of the Cys319 loop may gate the catalytic cycle.

Keywords

IMP dehydrogenase; IMPDH; enzyme catalysis; enzyme kinetics; guanine nucleotide biosynthesis

X-ray crystal structures of enzyme-ligand complexes are widely believed to provide snapshots of enzyme-substrate interactions during a catalytic cycle, but such structures may instead depict off-pathway conformations that are not relevant to catalysis. Crystallization

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Supporting Information Available. The Supporting Information includes a description of the determination of K_C and figures showing the global fits for the progress curves of the R322A and E323A reactions. Supplemental materials may be accessed free of charge online at <http://pubs.acs.org>

conditions introduce a further complication, since pH, ionic strength and salt composition are seldom physiological, and therefore increase the chance that the crystallized conformation is not catalytically relevant. While site directed mutagenesis is often used to confirm the importance of active site residues, such experiments rarely go beyond the evaluation of k_{cat} and K_{m} , and thus are very blunt probes of residue function and the catalytic competence of a given conformation. Thus the presumption that X-ray crystal structures of enzyme-ligand complexes mimic catalytic states has seldom been thoroughly scrutinized.

IMP dehydrogenase (IMPDH) is an excellent system to address these questions. This enzyme catalyzes two distinct chemical transformations (Scheme 1; *Trichomonas foetus* IMPDH numbering will be used throughout): (1) a dehydrogenase reaction where the catalytic Cys319 attacks the 2-position of IMP, expelling a hydride to NAD^+ to form the thioimidate intermediate E-XMP*; (2) a hydrolysis reaction which converts E-XMP* to XMP. Ten x-ray crystal structures have been solved of enzyme-inhibitor complexes of *T. foetus* IMPDH alone, with resolutions from 2.0–2.2 Å, at pH 6.3–8.0, in either polyethylene glycol or sodium malonate precipitants (1–5). These structures suggest that IMPDH undergoes a series of conformational changes during the course of a catalytic cycle (6). A large conformational change separates the two chemical reactions: NADH departs after the dehydrogenase reaction and a mobile flap (residues 412–432) moves into the vacant dinucleotide site, carrying Arg418 into the active site where it acts as a general base to activate water (7). This closed conformation is captured in the crystal structure of the E•MZIP complex (2). Mutations of Arg418 and the neighboring Tyr419 disrupt the hydrolysis step, but have no effect on the hydride transfer step, confirming the catalytic relevance of the E•MZIP crystal structure (1,8,9).

Two other mobile structural elements are evident in the crystal structures: the loop that contains the catalytic Cys319 (residues 313–328) and the C-terminal segment (residues 481–503) (Figure 1). The Cys319 loop is trapped in at least three conformations, but is also disordered in several other crystals. The C-terminal segment is disordered in most crystals, but is found in a loose helix in E•RVP complexes and in an alpha helix in E•MZIP. With the exception of the E•MZIP complex, the catalytic relevance of these structures has not been established. The present manuscript addresses the catalytic relevance of the various conformations of the Cys319 loop, so these conformations will be described in more detail.

The entire Cys319 loop is visible in four structures in addition to E•MZIP: E•IMP, E•IMP•TAD, E•RVP•MPA and E•RVP (PDB accession numbers 1ME9, 1LRT, 1ME7 and 1ME8, respectively; Figure 1) (1–4). E•RVP•MPA and E•RVP are very similar, so only the E•RVP•MPA conformation will be considered further here. E•IMP•TAD is believed to be a model for E•IMP• NAD^+ while E•RVP•MPA is a possible model for both E-XMP*• NADH^+ and E-XMP*_{open}. The ends of the Cys319 loop, residues 313–318 and 325–328, are very similar in all of the structures, and the positions of Cys319 and Ile320 change only slightly. In contrast, the position of Arg322 changes dramatically, and Thr321, Glu323 and Gln324 are also found in different positions. The most extreme positions of the side chains of Thr321, Arg322, Glu323 and Gln324 are separated by 8.8, 13, 6.7 and 5.6 Å, respectively. Similar conformations are observed in both E•IMP and E•IMP•TAD, though the side chain of Arg322 is found in different rotamers (Figure 1D and E): Arg322 is solvent exposed in E•IMP, but forms two hydrogen bonds to the carboxamide group of TAD in E•IMP•TAD. A second conformation of the Cys319 loop is found in E•RVP•MPA (Figure 1D): the helical arrangement of residues 322–324 has shifted; this segment makes only one hydrogen-bond to a protein residue, between Gln324 and Cys319, but this interaction is not likely to be available when Cys319 is part of E-XMP*. The E•RVP•MPA structure is considerably more ordered than E•IMP and E•IMP•TAD: the C-terminal segment is visible, and the

monovalent cation binding site is present. The order increases in E•MZIP: the Cys319 loop makes one turn of an alpha helix, the C-terminal segment forms an alpha helix and the flap occupies the dinucleotide site. This Cys319 conformation is stabilized by a hydrogen bonding network that includes 5 hydrogen bonds with Arg322 (Figure 1E).

These structures suggest the following model for the IMPDH catalytic cycle (Figure 2, Path 1): when NAD⁺ binds, Arg322 rotates to form two hydrogen bonds that facilitate hydride transfer. Formation of E-XMP* changes the conformation of the Cys319 loop to allow monovalent cation binding and the C-terminal segment to dock; when NADH departs, residues 321–324 and the C-terminal segment form alpha helices anchored by Arg322. This new conformation allows the flap to close so that E-XMP* is hydrolyzed. However, the catalytic relevance of these conformations is by no means assured: in addition to the caveats discussed above, Cys319 is oxidized to sulfenic acid in E•IMP, and E•RVP•MPA contains Na⁺ instead of K⁺. Further, the Cys319 loop has similar conformations in the structures of E•MZIP and the E-XMP*•MPA complex of Chinese hamster IMPDH (2,10), which suggests another plausible alternative (Path 2): the Cys319 loop has the same conformation in all E-XMP* complexes, i.e., the conformation found in E•MZIP. Path 2 implies that the movement of the flap is independent of the conformation of the Cys319 loop.

If the Cys319 loop has different conformations throughout the catalytic cycle as in Path 1, then mutations in residues 321–324 should have disparate effects on the dehydrogenase and hydrolysis steps (Figure 3A). Conversely, if the Cys319 loop has the same conformation in all E-XMP* complexes, these mutations should have similar effects on both steps (Figure 3B). At present, only the Ala mutation of Thr321 has been evaluated in detail (8) and this substitution has equivalent effects on both hydride transfer and hydrolysis. However, this mutation removes hydrogen-bonding interactions in two conformations of the Cys319 loop, so this result is not informative. Here we have examined the mechanistic consequences of mutations at Arg322, Glu323 and Gln324. The results are consistent with Path 1, suggesting that the crystal structures are indeed catalytically relevant. Moreover, these experiments have uncovered communication between the Cys319 loop and the Arg418 flap, suggesting that the conformation of the Cys319 loop may gate the closure of the flap.

MATERIALS AND METHODS

Materials

IMP was purchased from Sigma. NAD⁺ was purchased from Roche. DTT was purchased from Research Organics, Inc. Tris, glycerol, EDTA, and KCl were purchased from Fisher Scientific.

Synthesis of 2-²H-IMP

2-²H-Inosine was synthesized from AICAR and ²H-ethylorthoformate and phosphorylated with phosphorus oxychloride in trimethyl phosphate using known procedures as described previously (11).

Site-Directed Mutagenesis

The R322A, R322E, Q324A and E323A mutations were constructed in pTf1 which contains the *Tf*IMPDH gene in the pKK223-3 plasmid (12). Point mutations were created using the Quikchange kit (Stratagene, La Jolla, CA). The entire coding sequences were sequenced to ensure that no undesired mutations were introduced (Genewiz, Inc.).

Expression and Purification of IMPDH

The mutant IMPDH genes were expressed in *Escherichia coli* H712 cells, which lack endogenous IMPDH (13). All the mutants were purified using IMP affinity column chromatography with buffer A (50 mM Tris, pH 7.5, 1 mM DTT, 10% glycerol). IMPDH was eluted with 0.5 mM IMP in buffer A, dialyzed against buffer A to remove IMP, and stored at -20°C . Purity was $>95\%$ as assessed by SDS-PAGE. The concentration of IMPDH was determined by Bio-Rad assay using IgG as a standard with a correction factor of 2.6 as previously described (9).

Steady-State Kinetics

Standard IMPDH assays were performed in assay buffer (50 mM Tris-Cl, pH 8.0, 100 mM KCl, and 1 mM DTT) at 25°C . The production of NADH was monitored spectrophotometrically at 340 nm ($\epsilon = 6.22 \text{ mM}^{-1}\text{cm}^{-1}$). Initial velocities were fit by the Michaelis-Menten equation (eq 1) or an equation including uncompetitive inhibition (eq 2) using SigmaPlot (Systat Software, Inc.)(12):

$$v = V_m[S]/(K_m + [S]) \quad (1)$$

$$v = V_m/(1 + (K_m/[S]) + ([S]/K_{ii})) \quad (2)$$

where v is the initial velocity, V_m is the maximal velocity, K_m is the Michaelis constant for substrate (S), and K_{ii} is the inhibition constant for S.

Pre-Steady-State Kinetics

Pre-steady-state experiments were performed monitoring either NADH absorbance (340 nm) or fluorescence (340 nm excitation, 420 nm cutoff filter). Enzyme (3 μM for R322A, E323A and Q324A; 1.5 μM for R322E) and saturating IMP (300 μM for R322A, 1.5 mM for R322E, 1 mM for E323A and Q324A) were pre-incubated and combined with an equal volume of varying concentrations of NAD^+ .

Multiple-Inhibitor Kinetics

Multiple-inhibitor experiments with tiazofurin and ADP were performed as described previously (9). Reaction mixtures contained IMP and NAD^+ . For R322A 200 μM IMP, 1 mM NAD^+ was used. For R322E, 500 μM IMP and 1 mM NAD^+ was used. For Q324A, 200 μM IMP and 120 μM NAD^+ was used. For E323A, 1 mM IMP and 350 μM NAD^+ was used. Initial velocities were fitted to eq 3:

$$v = v_0/(1 + [I]/K_i + [J]/K_j + [I][J]/\alpha K_i K_j) \quad (3)$$

where v is the initial velocity, v_0 is the initial velocity in the absence of inhibitor, K_i and K_j are the inhibition constants for inhibitors I and J, respectively, and α is the interaction constant.

Equilibrium Dissociation Constants (K_d)

NAD^+ was successively added to 0.2–0.3 μM wild-type or R322E in assay buffer at 25°C . Protein fluorescence was measured with Hitachi F-2000 fluorescence spectrophotometer. To minimize inner filter effects, excitation wavelength = 295 nm and emission was monitored at 344 nm. Inner filter corrections were calculated by the formula

$$F_c/F_o = \text{antilog}[(A_{ex} + A_{em})/2] \quad (6)$$

Where F_c is the corrected fluorescence, F_o is the observed fluorescence, and A_{ex} and A_{em} are the absorbances at the excitation and emission wavelengths respectively. F_c/F_o was less than 2 for all concentrations. Since purines are fluorescent quenchers, nonspecific quenching was measured by titrating ligand against a solution of L-tryptophan as previously described (14).

RESULTS AND DISCUSSION

Mutant construction and initial characterization

We used site-directed mutagenesis to construct R322A, R322E, E323A and Q324A. These enzymes were expressed in an *E. coli* strain that lacks the endogenous IMPDH. The enzymes were purified using IMP affinity chromatography, which demonstrates that the mutations did not grossly perturb structure. All of the enzymes were active; steady state kinetic parameters are shown in Table 1. The mutant enzymes displayed at least 20% activity relative to wild-type, with the exception of R322E, which has 0.5% activity. All the mutations increase the value of K_m for IMP (unfortunately, the low activity of R322E precluded the determination of the K_m for IMP), and all except Q324A increase the value of the K_m for NAD^+ .

Substitutions of Arg322, Glu323 and Gln324 perturb the equilibrium between open and closed conformations

As described previously (8,9,15,16), the equilibrium (K_c) between open and closed conformations of the flap was probed with multiple inhibitor experiments: if the closed conformation is favored, the binding of tiazofurin in the nicotinamide subsite will shift the enzyme to the open conformation, increasing the affinity of ADP for the adenosine subsite. This synergistic interaction provides a measure of $E\text{-XMP}^*_{\text{open}}$, and thus of the value of K_c (see Supporting Information for detailed explanation). E323A is no longer inhibited by tiazofurin ($K_i > 200$ mM), so the value of K_c could not be determined for this enzyme.

Synergistic interactions are observed between tiazofurin and ADP for R322A, E323A and Q324A (Figure 4 and Table 2). Surprisingly, the value of K_c is significantly decreased in all three mutant enzymes (Table 2) even though Arg322, Glu323 and Gln324 do not interact directly with the flap. All three residues participate in extensive hydrogen bonding networks when the flap is in the closed conformation (Figure 1E), and have few interactions in complexes where the flap is open (Figure 1B–D). These observations suggest that flap closure requires that the Cys319 loop adopts the conformation of the $E\bullet\text{MZP}$ complex. Therefore the conformation of the Cys319 loop controls the closing of the flap, effectively gating the hydrolysis reaction.

Experimental Strategy

A series of isotope effect and pre-steady state experiments were performed to delineate the effects of these mutations on both chemical transformations. The hydride transfer step was probed by monitoring the reactions with $2\text{-}^2\text{H}\text{-IMP}$ while the hydrolysis step was probed by performing the reaction in D_2O . Pre-steady state reaction progress curves were collected by mixing $E\bullet\text{IMP}$ with varying concentrations of NAD^+ ; experiments monitored NADH production with absorbance and/or fluorescence. The absorbance at 340 nm reports on both $E\text{-XMP}^*\bullet\text{NADH}$ and free NADH. Fortuitously, $E\text{-XMP}^*$ quenches the fluorescence of NADH in the $E\text{-XMP}^*$ complex, so only free NADH has a fluorescent signal (excitation

wavelength = 340 nm, emission wavelength = 400 nm). The progress curves were initially fit to equations containing single or double exponential and steady-state terms, and the resulting values were used as starting points for global fits to the mechanism of Scheme 1 using Dynafit (17). Figure 5 shows the global fit for Q324A as an example; fits for the other enzymes are included in the Supporting Information. The global fits were validated by determining the values of k_{cat} , K_m and K_{ii} for NAD^+ using the initial rates simulated with the microscopic rate constants; good agreement with the experimental values was obtained in all cases (compare Tables 1 and 3). In addition, the isotope effects were simulated with the assumption that the intrinsic isotope effects are 3, and again the simulated values are in good agreement with experimental values.

Despite the gross similarities in the steady state parameters of R322A, E323A and Q324A, each mutation has distinct effects on the microscopic rate constants, as summarized in Figure 6 and described in detail below in order of increasing perturbation. The low activity of R322E permitted only rudimentary characterization of the microscopic rate constants, which nonetheless reveal that this mutation has a large effect on both chemical transformations.

Characterization of E323A

The E323A mutation has only subtle effects on the IMPDH reaction. A burst of NADH is observed in the pre-steady state, which indicates that the hydride transfer step is fast relative to subsequent steps, as observed with wild-type IMPDH. A somewhat larger S.I.E. is observed than with the wild-type enzyme, which suggests that the hydrolysis step has become more rate-limiting. Global fits of the reaction progress curves substantiate these conclusions (Table 3 and Figure 6): the release of NADH is slightly faster than in wild-type IMPDH, and the hydrolysis step is slightly slower. The global fit uncovers a significant change in the hydride transfer step: the rate ($k_7 + k_8$) decreases by a factor of 4 while the equilibrium (k_7/k_8) increases from 0.6 to 3. Intriguingly, this 5-fold increase in stability of $\text{E-XMP}^*\cdot\text{NADH}$ is similar in magnitude to the decrease in the value of K_c described above (a factor of 3); both of these effects can be explained if this mutation stabilizes open conformations of E-XMP^* relative to $\text{E-XMP}^*_{\text{closed}}$ as illustrated in Figure 3A. This result is consistent with the x-ray crystal structures: Glu323 makes hydrogen bonds to water molecules in the closed conformation of the $\text{E}\cdot\text{MZP}$ complex but is solvent exposed in the other x-ray crystal structures (Figure 1). Therefore these results are consistent with Path 1.

Characterization of Q324A

The Q324A mutation also has fairly subtle effects on the IMPDH reaction. As with E323A, a burst of NADH production is present in the pre-steady-state; a larger S.I.E. is observed, suggesting that the hydrolysis step has become even more rate-limiting. Global fits of the reaction progress curve reveal that while this mutation decreases the rate of the hydride transfer step by a factor of 2, and the value of k_{HOH} decreases by a factor of 8. Like E323A, this mutation increases the equilibrium for hydride transfer to 10. Again, the magnitude of this change is similar to the change in K_c (a factor of 12), suggesting that this substitution also stabilizes the open conformations of E-XMP^* as in Figure 3A. As above, these results are generally consistent with the structural information. Gln324 is solvent exposed in the $\text{E}\cdot\text{IMP}\cdot\text{TAD}$ complex (Figure 1C), which suggests that residue is not involved in the hydride transfer reaction. In contrast, Gln324 makes two hydrogen bonds in the structure of $\text{E}\cdot\text{MZP}$ complex, one of which is to a structural water that makes two additional hydrogen bonds to protein residues (Figure 1E). Therefore these findings also consistent with Path 1.

Characterization of R322A

When 2-²H-IMP was the substrate, a small isotope effect was observed on k_{cat} , suggesting that hydride transfer may be partially rate-limiting. A burst of free NADH is observed in the pre-steady state when the reaction is monitored with fluorescence, indicating that a step subsequent to NADH release must be rate-limiting. Surprisingly, no solvent deuterium isotope effect was observed for the reaction of R322A, indicating that the hydrolysis step is no longer rate-limiting. The only likely rate-limiting step remaining is k_{close} , the closure of the flap.

Global fits of the reaction progress curves were performed to two mechanisms; the first assumed that the conformational change is fast and the second assumed that k_{close} is rate-limiting. Good fits were obtained to both mechanisms (Table 3), and the values of k_{cat} , K_{m} NAD⁺ and K_{ij} NAD⁺ derived from both mechanisms were in good agreement with experimental values. However, a significant SIE is expected on k_{cat} when the conformational change is fast, but a SIE is not observed experimentally. When the conformational change is slow, a small SIE on k_{cat} is predicted. Therefore these results suggest that the substitution of Arg418 with Ala has a profound effect on the rate of flap closure (Figure 6). The global fit also indicates that the R322A mutation significantly perturbed the rate of hydride transfer, though perhaps not as much as would be expected if two hydrogen bonds are lost to the carboxamide group of NAD⁺. Thus these results provide further support for Path 1, but cast some doubt on the catalytic relevance of the E•IMP•TAD complex.

Characterization of R322E

We constructed and characterized R322E to gain further insight into the role of Arg322. This substitution will clearly disrupt the hydrogen bonding network that stabilizes the E•XMP*closed structure, and should thus decrease the hydrolysis step. If the interaction between Arg322 and NAD⁺ is part of the catalytic cycle, this substitution will also profoundly perturb hydride transfer. In contrast, if Arg322 is solvent exposed during hydride transfer as in the E•IMP structure, then this substitution may have a modest effect on hydride transfer.

Unlike the other IMPDH mutants, R322E dramatically decreased the value of k_{cat} , by a factor of 200. A SIE is observed on k_{cat} , suggesting that the hydrolysis step is partially rate-limiting. A significant ^D k_{cat} is also observed, indicating that hydride transfer is also partially rate-limiting. Assuming intrinsic isotope effects of ~3 on both steps, the values for k_7 is ~0.015 and k_{HOH} is ~0.03, indicating that this mutation decreases the rate of hydride transfer by a factor of 2000 and the rate of hydrolysis by a factor of 130. These results are consistent with Path 1 and suggest that Arg322 may play an important role in hydride transfer as implied by E•IMP•TAD.

The binding of NAD⁺ was monitored via intrinsic protein fluorescence to further probe the defect in hydride transfer. We previously reported that NAD⁺ binding curve for the wild-type enzyme was sigmoidal and best fit a model with two binding constants (12). However, the binding curve becomes hyperbolic with a correction for non-specific quenching (16), with $K_{\text{d}} = 27 \pm 6 \mu\text{M}$. NAD⁺ binds to R322E with similar affinity, $K_{\text{d}} = 37 \pm 10 \mu\text{M}$. ADP also binds to both wild-type and mutant enzymes with similar affinity (Table 2). In contrast, the mutation decreases the affinity of tiazofurin by a factor of >3. These observations suggest that while NAD⁺ still binds to R322A, the nicotinamide is no longer positioned correctly for efficient hydride transfer, providing further evidence for the catalytic relevance of the E•IMP•TAD structure.

Implications for the catalytic cycle of IMPDH

X-ray crystal structures of enzyme-ligand complexes are generally presumed to provide snapshots of the catalytic cycle. In the case of IMPDH, two plausible models arise from the structures (Figure 3). In Path 1, the conformation of the Cys319 loop gates flap closure while the flap movement is independent of the Cys319 loop conformation in Path 2. The above experiments find that mutations of Arg322, Glu323 and Gln324 have differential effects on hydride transfer, flap closure and hydrolysis that are most consistent with Path 1. These results provide evidence for the catalytic relevance of the E•IMP•TAD, E•RVP and E•RVP•MPA structures.

Supplementary Material

Refer to Web version on PubMed Central for supplementary material.

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Abbreviations

IMP	inosine 5'-monophosphate
XMP	xanthosine 5'-monophosphate
IMPDH	IMP dehydrogenase
TAD	tiazofurin adenine dinucleotide
TAZ	tiazofurin
RVP	ribavirin monophosphate
MPA	mycophenolic acid
MZP	mizoribine monophosphate

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Figure 1.

The conformations of the Cys319 loop. A. Overlay of the structures of monomers of E•IMP (cyan, 1ME9), E•IMP•TAD (salmon, 1LRT), E•RVP•MPA (blue, 1ME8) and E•MZIP (magenta, 1PVN). Only the side chains of Cys319 and Arg322 are displayed. IMP, RVP and MZIP are displayed; these molecules occupy the same space, so only IMP is labeled. TAD and MPA are omitted for clarity. The flap is only visible in the E•MZIP structure, where it occupies the NAD site. Panels B–E show a different view of the Cys319 loop in these structures. All residues that interact with Arg322, Glu323 and Gln324 are shown; residues from the neighboring monomer are denoted with '. All panels display the same set of residues, unless those residues are disordered (e.g., residues 484'–489' are disordered in E•IMP). Green lines denote hydrogen bonds. IMPDHs are activated by monovalent cations, but the monovalent cation binding site is visible only in the structures of the E•RVP•MPA and E•MZIP complexes. B. E•IMP; C. E•IMP•TAD, a putative model for E•IMP•NAD⁺; D. E•RVP•MPA, a putative model for the open states of E-XMP*, e.g., E-XMP*•NADH and E-XMP*_{open}. Note that while the 481'–489' segment has a helical structure, it does not form an alpha helix as in the E•MZIP complex in the next panel. E. E•MZIP, a model for E-XMP*_{closed}. This figure was produced with UCSF Chimera (19).

Path 1: Cys319 loop has distinct conformations in open and closed E-XMP complexes. Flap closure is coupled to the conformation of Cys319 loop.*

Path 2: Cys319 loop has the same conformation in all E-XMP complexes. Flap closure is independent of the conformation of Cys319.*

Figure 2.

Model for conformational transitions of the IMPDH catalytic cycle. The Cys319 loop is depicted in blue, disordered segments are shown in gray and the monovalent cation is shown in orange. The presence of hydrogen bonds is indicated by the dashed lines.

Figure 3.

Hypothetical effects of mutations of on the various complexes in the IMPDH catalytic cycle. The wild-type reaction coordinate is depicted in black and the consequences of mutations in red. A. If the Cys319 loop has a different conformation in the open states of E-XMP*, e.g., E-XMP*•NADH and E-XMP*_{open}, than in E-XMP*_{closed}, then the mutations should have different effects on these states. For example, the R332A mutation removes 5 hydrogen bonds from E-XMP*_{closed}, and so should significantly destabilize this conformation. However, Arg322 is solvent exposed in the open conformations of E-XMP*, so this mutation should have little effect on these states. B. If the Cys319 loop has the same conformation in all E-XMP* states, then a mutation that destabilizes E-XMP*_{closed} will destabilize all other states.

Figure 4. Multiple inhibitor experiment with R322A. ADP concentrations = 0 (filled circles), 2.5 (open circles), 5 (filled squares) and 10 mM (open squares).

Figure 5.

Progress curves for the Q324A reaction. The production of NADH is monitored. $[\text{E}] = 3 \mu\text{M}$, $[\text{IMP}] = 500 \mu\text{M}$, $[\text{NAD}^+] = 250 \mu\text{M}$ (gray), $500 \mu\text{M}$ (pink), $1000 \mu\text{M}$ (purple), $2000 \mu\text{M}$ (green), $4000 \mu\text{M}$ (orange) and $8000 \mu\text{M}$ (blue). Simulations using the parameters derived in the global data fitting using Dynafit are shown in black (Table 3).

Figure 6.

The effects of the Cys319 loop mutations on the IMPDH reaction. The table is colored according to the magnitude of change induced by the mutation. No color, factor of 1–3; blue, factor of 2–10; green, factor of 10–100; orange, factor >100.

The IMPDH reaction and structures of inhibitors. Both the hydride transfer and hydrolysis reactions are likely to be addition/eliminations; however, single rate constants can describe these steps since the lifetimes of the tetrahedral intermediates are fleeting. Therefore the tetrahedral intermediates have been omitted for clarity. Rate constants as defined previously (9,12,16).

Table 1

Steady-state kinetic parameters and isotope effects for reactions of *T. foetus* IMPDH mutants.^a

Enzyme	W.T ^e	R322A	R322E	E323A	Q324A
k_{cat} (s ⁻¹)	1.9 ± 0.2	0.36 ± 0.01	0.009 ± 0.001	1.10 ± 0.03	0.43 ± 0.03
K_m IMP (μM)	1.7 ± 0.4	7.6 ± 0.8	n.d. ^g	31 ± 8	28 ± 6
K_m NAD ⁺ (μM)	150 ± 30	910 ± 90	2600 ± 700	300 ± 20	120 ± 20
K_i NAD ⁺ (mM)	7 ± 2	n.a. ^f	n.a. ^f	n.a. ^f	2.4 ± 0.3
Dk_{cat}^b	1.05 ± 0.05	1.5 ± 0.2	2.2 ± 0.8	n.d. ^g	n.d. ^g
Dk_{cat}/K_m^b	2.0 ± 0.5	2.1 ± 0.1	n.d. ^g	n.d. ^g	n.d. ^g
S.I.E. k_{cat}^c	1.6 ± 0.1	1.1 ± 0.1	1.5 ± 0.1	2.6 ± 0.2	2.2 ± 0.1
S.I.E. k_{cat}/K_m^c	0.7 ± 0.1	0.4 ± 0.1	n.d.	0.6 ± 0.2	1.3 ± 0.7
rate-limiting step(s) ^d	$k_9 + k_{HOH}$	$k_H + k_{close}$	$k_H + k_{HOH}$	k_{HOH}	k_{HOH}

^a Conditions as described in Materials and Methods.

^b Isotope effect using 2-²H-IMP.

^c Solvent deuterium isotope effect.

^d See Scheme 1; k_9 , NADH release; k_H , hydride transfer = $k_7 + k_8$; k_{close} , flap closure; k_{HOH} , hydrolysis of E-XMP[®].

^e Data from (12).

^f n.a., not applicable- no NAD⁺ inhibition observed at concentrations up to 5 mM.

^g no data

Table 2

Determination of K_c for the *T. foetus* IMPDH mutants. The value of the interaction constant α was determined using multiple inhibitor experiments with tiazofurin (TAZ) and ADP as described in Materials and Methods and Supplemental Information. The value of α approximates the fraction of E-XMP*_{open}, allowing the calculation of K_c .

Mutant	K_c	α	K_i ADP (mM)	K_i TAZ (mM)
W.T ^a	140	0.007 ± 0.002	31 ± 2	69 ± 9
R322A	11	0.08 ± 0.03	17 ± 4	91 ± 18
R322E	n.a. ^b	n.a. ^b	50 ± 40	>200
E323A	49	0.02 ± 0.02	25 ± 7	300 ± 100
Q324A	6	0.14 ± 0.08	7 ± 2	90 ± 25

^aData from (9,18).

^bn.a., not applicable.

No inhibition was observed with tiazofurin in the presence or absence of ADP.

Table 3

Microscopic rate constants for the reactions of the IMPDH mutants. Global fits of the reaction of Scheme 1 were obtained for reaction progress curves for R322A, E323A and Q324A using Dynafit (17). The rate constants of R322E were estimated from the values of k_{cat} and isotope effects, with the assumption that the intrinsic isotope effects are 3.

Parameter	Wild-type ^a	R322A ^b conform fast	R322A ^c conform slow	R322E	E323A	Q324A
NAD^+ on, k_5 ($\times 10^{-3} \text{ M}^{-1} \text{ s}^{-1}$)	29 ± 2	5 ± 1	6 ± 1	≥ 3.5	300 ± 200	32 ± 6
NAD^+ off, k_6 (s^{-1})	4 ± 1	19 ± 3	26 ± 4	≥ 740 ($\geq K_m/k_5$)	540 ± 390	140 ± 40
Hydride transfer forward, k_7 (s^{-1})	34 ± 2	7.5 ± 0.9	7.8 ± 0.8	~ 0.015	16.3 ± 0.4	41 ± 3
Hydride transfer reverse, k_8 (s^{-1})	59 ± 5	5.7 ± 0.8	5.3 ± 0.6	n.d.	5.2 ± 0.4	3.7 ± 0.4
NADH release, k_9 (s^{-1})	8.5 ± 0.4	2.9 ± 0.3	2.9 ± 0.1	fast	12.7 ± 0.5	4.9 ± 0.2
k_{close}	14000	35 ± 760	1.4 ± 0.3	fast	140 ± 140	30 ± 80
k_{open}	1000	3 ± 25	n.a.	fast	9 ± 5	6 ± 30
NAD^+ inhibit on, k_{11} ($\times 10^{-6} \text{ M}^{-1} \text{ s}^{-1}$)	2	0.1 ± 2	0.06 ± 0.5	n.a.	0.06 ± 0.06	0.01 ± 0.02
NAD^+ inhibit off, k_{12} (s^{-1})	27 ± 3	20 ± 500	80 ± 600	n.a.	35 ± 46	7 ± 50
k_{HOH} (s^{-1})	4	1 ± 3	4 (fixed)	0.03	3.4 ± 0.3	0.48 ± 0.01
Calc K_c	n.a.	7	n.a.	n.a.	15	5
sim K_m NAD^+ (μM) ^c	110	430	510	n.a.	290	100
sim k_{cat} (s^{-1}) ^d	2.0	0.3	0.3	n.a.	1.8	0.4
sim K_i NAD^+ (μM) ^c	4300	>5000	>5000	n.a.	>5000	5000
sim $D k_{\text{cat}}^{d,e}$	1.1	1.1	1.1	n.a.	1.3	1
sim $^D k_{\text{cat}}/K_m^{d,e}$	1.2	1.6	1.9	n.a.	2.6	1
sim $^{D2O} k_{\text{cat}}^{d,e}$	2.1	2.1	1.2	n.a.	2.1	2.5
sim $^{D2O} k_{\text{cat}}/K_m^{d,e}$	1.1	0.6	0.6	n.a.	1.1	0.9

^aValues from (9), where the data from (12) was reprocessed using Dynafit. In (12), the values k_5 and k_6 were assigned from reactions with an inactive mutant. However, the mutation perturbed NAD^+ binding, so we now rely entirely on the global fit to wild-type data. The value of k_{HOH} was measured experimentally in the reaction with acetylpyridine adenine dinucleotide and fixed during the global fit; the values of k_{close} and k_{open} were fixed at arbitrarily fast values such that $K_{\text{close}} = 140$, and the value of k_{11} is increased by a factor of $(1 + K_c)$ from (12) to account for the conformational change. Note that while these values are in good agreement with experiment in the value of k_{cat} and associated isotope effects, they are a rather poor match for isotope effects on k_{cat}/K_m (compare with Table 1).

- ^b Mechanism assumes that open to closed conformational change is in rapid equilibrium with $K_C = 11$. The values of k_{closed} , k_{open} , k_{12} , are unconstrained- equally good fits are obtained as long as these values are fast relative to the other parameters.
- ^c Mechanism assumes that closing of the flap is partially rate-limiting. The values of k_{open} and k_{HOH} are unconstrained by the data, and equally good fits are obtained as long as $k_{\text{open}} \leq 0.1$ and $k_{\text{HOH}} \geq 4$.
- ^d The reactions were simulated using the parameters above, saturating IMP and $[\text{NAD}^+] = 25, 50, 100, 250, 500, 1000$ and $2000 \mu\text{M}$. The resulting initial velocities were fit to Michaelis-Menten with or without NAD^+ substrate inhibition term to derive the values of k_{cat} , K_m , NAD^+ and $K_{ii} \text{NAD}^+$.
- ^e Assumes intrinsic isotope effects = 3, no changes in other steps; note that significant inverse solvent isotope effects are predicted on k_{cat}/K_m because more E-XMP* accumulates in the presence of D_2O , which in turn is trapped by NAD^+ in E-XMP*• NAD^+ , decreasing the value of K_m .