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Oppositines A and B, Sesquiterpene Pyridine Alkaloids from a Sri Lankan *Pleurostyliya opposita*

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Abstract

Two new sesquiterpene pyridine alkaloids, oppositines A (**1**) and B (**2**), have been isolated from the plant, *Pleurostyliya opposita* (Celastraceae), collected in Sri Lanka. The compounds were isolated and purified by solvent/solvent partitioning, column chromatography and HPLC. Their structures were assigned on the basis of extensive 1D and 2D NMR studies as well as analysis by HRESIMS. Oppositines A (**1**) and B (**2**) showed moderate cytotoxicity against HCT116 cell lines with EC₅₀ values of 27 ± 2 and 26 ± 3 μM, respectively.

Poly-acetylated sesquiterpene pyridine alkaloids are ubiquitously found in the Celastraceae family of plants.^{1–4} Dihydro-β-agarofuran sesquiterpenes frequently incorporate pyridine dicarboxylic acids to bridge their sesquiterpenoid cores at the C-3 and C-13 positions. The dihydro-β-agarofuran skeleton is comprised of a decalin ring, a tetrahydrofuran ring, and methyl substituents located at the C-4, C-10, and two at the C-11 position. Evoninic, wilfordic, edulic, cathaic, and cassininic (often misnamed cassinic) acids are some of the pyridine carboxylic acids present in this class of sesquiterpenes. These compounds are of interest due to their wide range of biological activities including: cytotoxicity,¹ immunosuppression,² insecticidal,^{3,5,6} insect-antifeedant,^{5,7} anti-HIV,⁴ reversal of the multidrug resistance (MDR) phenotype,^{8–11} and antitumor.¹² As part of an ongoing search for bioactive compounds, two new sesquiterpene alkaloids, oppositines A (**1**) and B (**2**), were isolated from the Sri Lankan plant, *Pleurostyliya opposita* (Wallich, 1824) (Celastraceae).

The air dried stem bark of *P. opposita* was extracted with 1:1 MeOH/CH₂Cl₂. The extract was dried and partitioned between petroleum ether and aqueous MeOH, and the aqueous MeOH was subsequently partitioned with chloroform. The chloroform-soluble material was fractionated using size exclusion chromatography, preparative TLC, and reversed-phase HPLC resulting in oppositines A (**1**) and B (**2**).

Oppositine A (**1**) was obtained as a light-yellow glass. The molecular formula, C₄₄H₅₁NO₁₈, was determined by HRESIMS. Initial interpretation of the ¹H and ¹³C NMR data indicated that **1** contained five acetyls and one hydroxyl. The data also indicated that oppositine A (**1**) contained four quaternary carbons, eight methines, five methylenes, and three methyls. The existence of a benzoate was suggested by the signals at δ_H 7.80 (dd, *J* = 7.9, 1.3 Hz, H-18, H-18'), 7.52 (tt, *J* = 7.9, 1.3 Hz, H-20), 7.38 (t, *J* = 7.9 Hz, H-19, H-19'), as well as five carbons at δ 164.6 (C-16), 133.5 (C-20), 129.5 (C-18, C-18'), 129.1 (C-17), and 128.5 (C-19, C-19'). The signal at δ_C 164.6 (C-16) was assigned to the carbonyl carbon of the benzoate based on

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an HMBC correlation with the proton signal at δ_H 7.80 (H-18, H-18'). HSQC and HMBC data further supported the assignment of the benzoate group (Supplemental Table 1). The existence of a 2,3-disubstituted pyridine unit was suggested by the signals for three aromatic protons at δ_H 8.82 (dd, $J = 4.6, 1.9$ Hz, H-6), 8.39 (dd, $J = 8.0, 1.9$ Hz, H-4), 7.26 (dd, $J = 8.0, 4.6$ Hz, H-5), as well as five aromatic carbons at δ_C 168.1 (C-2), 153.8 (C-6), 138.2 (C-4), 124.9 (C-3), and 120.8 (C-5). HSQC and HMBC data further supported the assignment of the 2,3-disubstituted pyridine unit (Supplemental Table 1). A dicarboxy side chain containing two esters, one methine, two methylenes, one ethyl group, and the 2,3-disubstituted pyridine unit supported the presence of a cassininic acid moiety in oppositine A (**1**). The NMR data for this portion of the molecule are consistent with data reported for the analogous protons in the compounds cassinine¹³ and orthophosphonine.¹⁴ Based on COSY data, signals at δ_H 5.88, 5.22, and 5.01 were assigned to H-1, H-2, and H-3, respectively, and signals at δ_H 7.00, 2.32, 5.51, and 5.38 were assigned to H-6, H-7, H-8, and H-9, respectively. Taken together, these data indicate that **1** is a sesquiterpene pyridine alkaloid with a dihydro- β -agarofuran skeleton. Oppositines A (**1**) and B (**2**) have been drawn with the tetrahydrofuran ring on the top face of the molecule so as to conform to the depiction of other cassininic acid containing molecules.^{13,14} The tetrahydrofuran ring can also be drawn on the bottom side of the molecule, as is noted in a review of Celastraceae sesquiterpenoids.¹⁵

The regiosubstitution of the various groups around the terpene skeleton was established based on HMBC data. In the HMBC spectrum of **1**, the methyl singlets at δ_H 2.31 (H-30), 2.19 (H-24), 2.10 (H-26), 1.38 (H-28), and 2.13 (H-22) correlated with the five acetyl carbonyl carbons at δ_C 170.4 (C-29), 170.0 (C-23), 170.0 (C-25), 168.9 (C-27), and 168.4 (C-21), respectively. An anisotropic effect caused by the benzoate at the C-1 position is thought to be the cause of H-28 resonating at such high field. The effect on H-28 is observed in similar compounds with a benzoate at the C-1 position.^{1,3,4} The attachment of the acetyl groups at C-2, C-6, C-8, C-9, and C-15 was established based on HMBC correlations from the proton signals at δ_H 5.22 (H-2), 7.00 (H-6), 5.51 (H-8), 5.38 (H-9), and 4.63 (H-15b) to the acetyl carbonyls at δ_C 168.4 (C-21), 170.0 (C-23), 170.0 (C-25), 168.9 (C-27), and 170.4 (C-29), respectively. The attachment of the benzoate at C-1 was supported by an HMBC correlation from the proton signal at δ_H 5.88 (H-1) to the carbonyl signal at δ_C 164.6 (C-16). The attachment of a methyl and a hydroxyl at C-4 was established on the basis of HMBC correlations between δ_H 1.56 (H-14) and δ_C 70.0 (C-4), as well as δ_H 4.96 (OH-4) and δ_C 70.0 (C-4). Additionally, HMBC correlations between the proton signals at δ_H 5.01 (H-3) and 5.96, 3.59 (H-13a, H-13b) and the carbon signals at δ_C 172.2 (C-12) and 166.4 (C-13), respectively, supported the attachment of the cassinate moiety at C-3 and C-13. Sesquiterpene pyridine alkaloids characteristically include diastereotopic methylenes attached to C-13 that resonate at very different chemical shifts.¹⁻⁴

The relative stereochemistry of oppositine A (**1**), as shown in Figure 1, was established on the basis of coupling constants, observed ROE correlations, and molecular modeling. A ROESY experiment showing ROE effects for H-1 to H-2, H-9, and H-12; H-14 to H-3, H-6, and H-15; and H-9 to H-8, and H-12, established the relative position of the substituent groups. All coupling constants associated with the terpene skeleton were less than 6 Hz (Supplemental Table 1), and supported the arrangements determined by the ROESY data. Therefore, the relative configuration in **1** was established as 1*R**, 2*R**, 3*S**, 4*S**, 5*S**, 6*R**, 7*R**, 8*R**, 9*S**, 10*R**, 11*S**. The stereochemistry at 7' was not determined.

Oppositine B (**2**) was obtained as a light-yellow glass. The molecular formula, C₄₄H₅₁NO₁₇, was determined by HRESIMS, and suggested that oppositine B (**2**) had one less oxygen than oppositine A (**1**). Comparison of the NMR data of **2** with **1** confirmed the absence of the hydroxyl at the C-4 position. Specifically, a methine signal at δ_H 2.72 (H-4), coupled to a methyl signal at δ_H 1.24 (H-14), showed HMBC correlations to carbons at δ_C 90.7 (C-5), 74.9

(C-3), 70.4 (C-2), and 50.9 (C-10). The relative stereochemistry of **2** was established on the basis of coupling constants, observed ROE correlations, and comparison with **1**. A ROESY experiment showing ROE effects for H-1 to H-2, and H-9; H-14 to H-3, H-6, and H-15; and H-9 to H-8, and H-12, established the relative position of the substituent groups. All coupling constants were less than 6 Hz (Supplemental Table 1), and supported the arrangements determined from the ROESY data. Therefore, the relative configuration in **2** was established as 1*R**, 2*S**, 3*R**, 4*R**, 5*S**, 6*R**, 7*R**, 8*R**, 9*S**, 10*R**, 11*S**. The stereochemistry at 7' was not determined.

Oppositines A (**1**) and B (**2**) showed moderate cytotoxicity against HCT116 cell lines with EC₅₀ values of 27 ± 2 and 26 ± 3 μM, respectively. Oppositines A (**1**) and B (**2**) were also tested for anti-HIV activity and HIF-1α inhibition, but were inactive up to a concentration of 10 μg/mL in both assays. Antifungal activity assays for oppositines A (**1**) and B (**2**) are currently in progress.

To date only two cassininic acid metabolites have been reported, cassinine¹³ and orthophosphonine.¹⁴ Oppositine A (**1**) differs from cassinine in hydroxylation at the C-4 carbon and the position of the benzoate at the C-1 carbon rather than the C-6 carbon. Oppositine B (**2**) differs from cassinine in the position of the benzoate. Previous investigations of *P. opposita* have yielded lupane and spermidine derivatives.^{16–18} This is the first reported incidence of dihydro-β-agarofuran sesquiterpenes from *P. opposita*, although a plethora of such compounds has been reported from the Celastraceae family.

Experimental Section

General Experimental Procedures

Optical rotations were measured on a JASCO DIP-370 digital polarimeter. UV spectra were acquired in spectroscopy grade MeOH using a Hewlett Packard 8452A diode array spectrophotometer. IR spectra were recorded on a JASCO FT/IR-420 spectrometer. NMR spectra were recorded on a Varian INOVA spectrometer operating at 500 MHz and 125 MHz, for ¹H and ¹³C, respectively. Chemical shifts are reported in ppm and were referenced to residual solvent. High-resolution ESIMS analyses were performed on either a ThermoFinnigan LTQ-FT or a Micromass Q-tof micro. Initial purification was performed on Sephadex LH-20 and HPTLC-Platten-Sil 60, Merck. HPLC was performed on a Beckman System Gold (diode array detector).

Plant Material

Stem bark of *Pleurostyliia opposita* (Wallich, 1824) (Celastraceae) was obtained from a tree growing on the University of Colombo Campus, Sri Lanka, in July 2005, and was identified by Professor A.S. Seneviratne. A voucher specimen of the sample is on file at the Department of Plant Sciences, University of Colombo, Sri Lanka.

Extraction and Isolation

The stem bark (155 g) was air dried and soaked in MeOH-CH₂Cl₂ (1:1 v/v). The crude extract was concentrated to dryness *in vacuo* and redissolved in a solution of MeOH/H₂O (9:1). The aqueous MeOH solution was extracted with petroleum ether, then diluted with water to a composition of MeOH/H₂O (6:4), and finally extracted with CHCl₃. The CHCl₃ soluble material was chromatographed on Sephadex LH-20 using EtOAc/MeOH/H₂O (40:10:4). The first of the two fractions collected was subjected to preparative TLC on silica gel using hexane/EtOAc (2:8). The third of the four fractions obtained (UV visualization) was purified using a C₁₈ monolithic column employing a gradient of 10% CH₃CN/H₂O to 50% CH₃CN/H₂O over

2 h and resulted in oppositine A (**1**, 3.5 mg, 0.002%) eluting at 98.9 min and oppositine B (**2**, 1.5 mg, 0.001%) eluting at 107.3 min.

Oppositine A (1): light yellow glass, $[\alpha]_D^{26} +20.3$ (c 0.13, MeOH); UV (MeOH) $\lambda_{\max}$ (log ϵ) 270 (3.47) 230 (4.14) 206 (4.08) nm; IR (film) $\nu_{\max}$ 3453 (br), 2961, 1744, 1584, 1451, 1369, 1247, 1156, 1097, 757, 713 cm^{-1} ; ^1H NMR (CDCl_3 , 500 MHz) δ 8.82 (1H, dd, $J = 4.6$, 1.9 Hz, H-6'), δ 8.39 (1H, dd, $J = 8.0$, 1.9 Hz, H-4'), 7.80 (2H, dd, $J = 7.9$, 1.3 Hz, H-18, H-18'), 7.52 (1H, tt, $J = 7.9$, 1.3 Hz, H-20), 7.38 (2H, t, $J = 7.9$ Hz, H-19, H-19'), 7.26 (1H, dd, $J = 8.0$, 4.6 Hz, H-5'), 7.00 (1H, br s, H-6), 5.96 (1H, d, $J = 11.9$ Hz, H-13a), 5.88 (1H, d, $J = 4.1$ Hz, H-1), 5.51 (1H, dd, $J = 5.8$, 4.1 Hz, H-8), 5.38 (1H, d, $J = 5.8$ Hz, H-9), δ 5.35 (1H, d, $J = 13.5$ Hz, H-15a), 5.22 (1H, dd, $J = 4.1$, 2.4 Hz, H-2), 5.01 (1H, d, $J = 2.4$ Hz, H-3), 4.96 (1H, br d, $J = 1.4$ Hz, OH-4), 4.63 (1H, d, $J = 13.5$ Hz, H-15b), 4.32 (1H, dddd, $J = 11.7$, 9.5, 4.9, 2.7 Hz, H-7'), 3.59 (1H, d, $J = 11.9$ Hz, H-13b), 2.32 (1H, d, $J = 4.1$ Hz, H-7), 2.31 (3H, s, H-30), 2.26 (2H, m, H-10a', H-11a'), δ 2.19 (3H, s, H-24), 2.13 (3H, s, H-22), 2.10 (3H, s, H-26), 1.98 (1H, ddd, $J = 13.1$, 7.4, 2.7 Hz, H-8a'), 1.87 (2H, m, H-10b', H-11b'), 1.72 (1H, m, $J = 7.4$ Hz, H-8b'), 1.71 (3H, s, H-12), 1.56 (3H, br d, $J = 1.4$ Hz, H-14), 1.38 (3H, s, H-28), 0.68 (3H, t, $J = 7.4$, H-9'); ^{13}C NMR (CDCl_3 , 125 MHz) δ 172.2 (C-12'), 170.4 (C-29), 170.0 (C-23, C-25), 168.9 (C-27), 168.4 (C-21), 168.1 (C-2'), 166.4 (C-13'), 164.6 (C-16), 153.8 (C-6'), 138.2 (C-4'), 133.5 (C-20), 129.5 (C-18, C-18'), 129.1 (C-17), 128.5 (C-19, C-19'), 124.9 (C-3'), 120.8 (C-5'), 94.0 (C-5), 85.0 (C-11), 75.7 (C-3), 73.7 (C-6), 73.5 (C-1), 71.4 (C-9), 70.0 (C-2, C-4), 69.8 (C-13), 68.9 (C-8), 60.1 (C-15), 52.5 (C-10), 50.8 (C-7), 42.2 (C-7'), 31.8 (C-10'), 31.3 (C-11'), 28.8 (C-8'), 22.9 (C-14), 21.6 (C-24), 21.4 (C-30), 21.0 (C-26), 20.9 (C-22), 19.9 (C-28), 18.0 (C-12), 11.9 (C-9); HRESIMS m/z 882.3183 $[\text{M} + \text{H}]^+$ (calcd for $\text{C}_{44}\text{H}_{52}\text{NO}_{18}$, 882.3184).

Oppositine B (2): light yellow glass, $[\alpha]_D^{26} +19.5$ (c 0.07, MeOH); UV (MeOH) $\lambda_{\max}$ (log ϵ) 268 (3.68) 228 (4.35) 204 (4.27) nm; IR (film) $\nu_{\max}$ 2928, 1744, 1567, 1451, 1369, 1226, 1157, 1074, 757, 713 cm^{-1} ; ^1H NMR (CDCl_3 , 500 MHz) δ 8.78 (1H, br d, $J = 3.9$ Hz, H-6'), 8.20 (1H, br d, $J = 6.8$ Hz, H-4'), 7.85 (2H, br d, $J = 7.7$ Hz, H-18, H-18'), 7.53 (1H, br t, $J = 7.7$ Hz, H-20), 7.39 (2H, br t, $J = 7.7$ Hz, H-19, H-19'), 7.24 (1H, dd, $J = 6.8$, 3.9 Hz, H-5'), 6.65 (1H, br s, H-6), 6.00 (1H, d, $J = 4.0$ Hz, H-1), 5.67 (1H, d, $J = 11.3$ Hz, H-13a), 5.53 (1H, dd, $J = 5.9$, 3.8 Hz, H-8), 5.37 (1H, d, $J = 5.9$ Hz, H-9), 5.33 (1H, br d, $J = 4.0$ Hz, H-2), 5.28 (1H, d, $J = 13.2$ Hz, H-15a), 4.83 (1H, br s, H-3), 4.63 (1H, d, $J = 13.2$ Hz, H-15b), 4.00 (1H, m, H-7'), 3.59 (1H, d, $J = 11.3$ Hz, H-13b), 2.72 (1H, q, $J = 7.9$ Hz, H-4), 2.41 (1H, d, $J = 3.8$ Hz, H-7), 2.23 (3H, s, H-30), 2.23 (1H, m, H-11a'), 2.21 (3H, s, H-24), 2.20 (1H, m, H-10a'), 2.12 (3H, s, H-26), 2.09 (3H, s, H-22), 1.98 (1H, m, $J = 7.2$, H-8a'), 1.87 (1H, m, H-10b'), 1.80 (1H, m, H-11b'), 1.79 (1H, m, $J = 7.2$ Hz, H-8b'), 1.65 (3H, s, H-12), 1.42 (3H, s, H-28), 1.24 (3H, d, $J = 7.9$ Hz, H-14), 0.76 (3H, t, $J = 7.2$, H-9'); ^{13}C NMR (CDCl_3 , 125 MHz) δ 172.8 (C-12'), 170.2 (C-29), 170.0 (C-25), 169.7 (C-23), 168.9 (C-27), 168.7 (C-21), 165.5 (C-13'), 165.4 (C-2'), 164.7 (C-16), 152.6 (C-6'), 138.0 (C-4'), 133.4 (C-20), 129.5 (C-18, C-18'), 129.3 (C-17), 128.5 (C-19, C-19'), 127.4 (C-3'), 120.9 (C-5'), 90.7 (C-5), 82.9 (C-11), 74.9 (C-3), 74.2 (C-6), 74.0 (C-1), 71.5 (C-9), 70.4 (C-2), 69.0 (C-8), 68.8 (C-13), 60.2 (C-15), 50.9 (C-10), 50.2 (C-7), 42.1 (C-7'), 36.9 (C-4), 30.7 (C-11'), 30.3 (C-10'), 29.0 (C-8'), 21.6 (C-24), 21.4 (C-30), 21.1 (C-26), 20.9 (C-22), 19.9 (C-28), 18.1 (C-12), 14.9 (C-14), 12.0 (C-9'); HRESIMS m/z 888.3087 $[\text{M} + \text{Na}]^+$ (calcd for $\text{C}_{44}\text{H}_{51}\text{NO}_{17}\text{Na}$, 888.3055).

Cell culture and MTT assay.¹⁹

The cancer cell line HCT-116 (human colon tumor) was grown at 37 °C, 5% CO_2 in McCoy's 5a medium containing 10% fetal bovine serum, penicillin (50 IU/mL), streptomycin (50 $\mu\text{g}/\text{mL}$) and 2 mM L-glutamine. Cells were plated at a density of 5000 cells/well in a 96-well plate. Twenty-four hours after plating, the cells were exposed to medium containing the compound at a concentration of 0.03 μM to 100 μM for 48 h at 37 °C. After the cells were exposed to the

compound, 100 μ L of fresh culture medium containing MTT at a final concentration of 0.3 mg/mL were added to each well and incubated for 3 h at 37 °C. The absorbance of each well was measured by a microplate reader (Multi-skan labsystems) at 570 nm. The percentage cytotoxicity was calculated by comparison of the A_{570} reading from treated versus control cells.

Supplementary Material

Refer to Web version on PubMed Central for supplementary material.

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- 1** Opposite A, R = OH
2 Opposite B, R = H

Figure 1.
Key ROE correlations supporting the relative stereochemistry of oppositine A (**1**).