A structural basis for Mg^{2+} homeostasis and the CorA translocation cycle

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We describe the CorA Mg^{2+} transporter homologue from *Thermotoga maritima* in complex with 12 divalent cations at 3.7 Å resolution. One metal is found near the universally conserved GMN motif, apparently stabilized within the transmembrane region. This portion of the selectivity filter might discriminate between the size and preferred coordination geometry of hydrated substrates. CorA may further achieve specificity by requiring the sequential dehydration of substrates along the length of its ~55 Å long pore. Ten metal sites identified within the cytoplasmic funnel domain are linked to long extensions of the pore helices and regulate the transport status of CorA. We have characterized this region as an intrinsic divalent cation sensor and provide evidence that it functions as a Mg^{2+} -specific homeostatic molecular switch. A proteolytic protection assay, biophysical data, and comparison to a soluble domain structure from *Archaeoglobus fulgidus* have revealed the potential reaction coordinate for this diverse family of transport proteins.

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Introduction

As the most abundant divalent cation within cells, Mg^{2+} enjoys diverse cellular roles. It helps to stabilize ribosomes and cellular membranes, and is indispensable for maintaining genomic stability (Hartwig, 2001). Mg^{2+} has the largest hydrated radius and the smallest ionic radius among the common biological cations, where its high charge density and relative structural rigidity probably reflects its distinct roles in biology. As such, Mg^{2+} transporters may constitute a unique class of integral membrane proteins (Kehres and Maguire, 2002; Maguire and Cowan, 2002). Only recently

have mechanisms of Mg^{2+} homeostasis and transport been revealed at the molecular level (Garcia Vescovi *et al*, 1996; Cromie *et al*, 2006; Lunin *et al*, 2006).

CorA represents the primary Mg^{2+} uptake system in most prokaryotes. It is constitutively expressed and functions to transport Mg^{2+} into cells against a concentration gradient. Named for the cobalt-resistant mutants in which it was first identified, CorA can also transport Co^{2+} and Ni^{2+} . The influx of Co^{2+} and Ni^{2+} through CorA may occur as leakage, as the concentrations required for transport would lead to toxicity (Kehres and Maguire, 2002). Under certain conditions, the CorA proteins can also mediate Mg^{2+} efflux (Snively *et al*, 1989; Liu *et al*, 2002; Kolisek *et al*, 2003), although the relevance of this is not understood. The prokaryotic ZntB subfamily actually mediates Zn^{2+} efflux rather than Mg^{2+} influx (Worlock and Smith, 2002). In yeast, the *alr1* family confers Al^{3+} resistance (MacDiarmid and Gardner, 1998) and encodes the major plasma membrane Mg^{2+} uptake system (Liu *et al*, 2002). The related *mrs2* genes are involved in mitochondrial Mg^{2+} uptake in yeast, plants, and mammals (Bui *et al*, 1999; Zsurka *et al*, 2001; Kolisek *et al*, 2003; Knoop *et al*, 2005). The *mrs2* family has been expanded in plants, and *Arabidopsis thaliana* contains at least 10 genes whose products are targeted to different cellular membranes (Li *et al*, 2001; Knoop *et al*, 2005). It appears that some of these systems are capable of transporting additional divalent cations, including Cu^{2+} , Fe^{2+} , Mn^{2+} , and Zn^{2+} (MacDiarmid and Gardner, 1998; Li *et al*, 2001).

The prokaryotic and eukaryotic CorA/Alr1/Mrs2 proteins can often functionally substitute for one another (Bui *et al*, 1999; Li *et al*, 2001; Zsurka *et al*, 2001; Kehres and Maguire, 2002). In fact, recent structural analysis has confirmed that these diverse transport proteins likely share a common structural scaffold (Lunin *et al*, 2006). Although their precise classification as either ion channels or ion transporters remains somewhat ambiguous (Snively *et al*, 1989; MacDiarmid and Gardner, 1998; Kehres and Maguire, 2002; Liu *et al*, 2002; Kolisek *et al*, 2003; Lunin *et al*, 2006), these proteins seem to utilize the membrane potential to drive Mg^{2+} uptake (Kolisek *et al*, 2003; Froschauer *et al*, 2004). The CorA transporters share a universally conserved GMN motif within their transmembrane region, and most prokaryotic proteins conserve an MPEL sequence in close proximity (Kehres and Maguire, 2002; Knoop *et al*, 2005). Mutations within the GMN motif are known to abolish Mg^{2+} transport; however, naturally occurring variants might be associated with the transport of other divalent cations (Knoop *et al*, 2005). In general, the sequence similarity among CorA orthologues is highest in the C-terminal membrane-spanning domain and lower within the N-terminal soluble region.

We report crystal structures of the full-length CorA homologue from the thermophilic Gram-negative bacterium *Thermotoga maritima* (TmCorA_{1–351}) and of a cytoplasmic domain from the thermophilic archaeon *Archaeoglobus fulgidus* (AfCorA_{1–263}) at 3.7 and 2.9 Å resolution, respectively.

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Our data indicate that a metal is stabilized within the transmembrane region and suggests that the CorA selectivity filter might discriminate between the size and preferred coordination geometry of hydrated cations. In addition, 10 metal-binding sites located within the cytoplasmic funnel domain are characterized as a physiologically tuned divalent cation sensor. This reveals a central role for CorA in prokaryotic Mg^{2+} homeostasis. A major conformational change is observed between AfCorA₁₋₂₆₃ and TmCorA₁₋₃₅₁, which allows us to propose an open state model and the structural basis for gating and transport in the CorA superfamily.

Results

Overall structure of CorA

TmCorA₁₋₃₅₁ presents a funnel-shaped pentameric assembly with approximate five-fold symmetry around a central pore. It stands over 100 Å in height and can be divided into two parts: a large N-terminal cytoplasmic funnel domain and a smaller C-terminal transmembrane (TM) domain (Figure 1A). Following a portion of the main chain without regular

secondary structure, the funnel domain has an α/β structure whose core is composed of a seven-stranded mixed β -sheet ($\beta 1$ – $\beta 7$). The β -sheet is flanked on one side by three short α -helices ($\alpha 1$ – $\alpha 3$) and on the other side by one short ($\alpha 4$) and two long α -helices ($\alpha 5$ – $\alpha 6$) (Figure 1B). This portion of the funnel domain is connected to $\alpha 7$, the supporting helix of the entire CorA assembly, which extends towards the membrane to form the central pore-lining TM helices (TM1). The five pore-lining helices are flanked by five additional TM helices (TM2, $\alpha 8$). The TmCorA_{1–351} model was refined to a 3.7 Å resolution with *R* and *R*_{free} values of 29.5 and 31.6%, respectively. Data collection and refinement statistics are presented in Supplementary Table 1, along with representative portions of the electron density map (Supplementary Figure 1).

Membrane and pore boundaries of CorA

The metal translocation pathway in CorA is surprisingly long, ~ 55 Å, from Asn314 to Asp277. This exceeds the ~ 30 Å hydrophobic stretch expected for a membrane bilayer, but is partially reconciled by considering what portion of the TM domain delineates the membrane boundary. Tyr344 and

Figure 1 Structure and sequence alignment of CorA. **(A)** Ribbon diagram of the TmCorA₁₋₃₅₁ pentamer. Each protomer is colored separately, and purple spheres represent bound Ca²⁺ ions. The funnel domain is cytoplasmic. **(B)** A TmCorA₁₋₃₅₁ protomer is shaded from blue to red (N- to C-terminus). Helical secondary structural elements are indicated. **(C)** Sequence alignment of prokaryotic CorA proteins from *T. maritima* (Tm), *A. fulgidus* (Af), *S. typhimurium* (St), *E. coli* (Ec), and *M. jannaschii* (Mj). Color coding is as follows: Purple – divalent cation sensor; blue – basic ring ($\alpha 7b$ and $\alpha 8$) and funnel domain ‘neutralizing residues’ ($\beta 4$ and $\alpha 6$); red – acidic ring ($\alpha 5$ and $\alpha 6$) and funnel domain ‘electrostatic sink residues’ ($\alpha 7a$ and $\alpha 7b$); orange – CorA elbow ($\alpha 5$ – $\alpha 6$ loop) and other potential hinges: Pro303 ($\alpha 7b$) and the GYP motif ($\alpha 8$); light green – pore lining residues ($\alpha 7b$); pink – hydrophobic girdle ($\alpha 7b$); light blue – hydrophobic belt ($\alpha 7b$); yellow – GMN motif and MPFL loop; dark green – membrane boundaries ($\alpha 8$). Two potential roles for Asp256 have been indicated. Charge ‘switching’ can partly account for the overall low sequence identity between CorA homologues.

Phe345 on TM2 create a pronounced aromatic ring at the intracellular lipid–aqueous interface, which aligns with Leu294 on TM1 (Figure 2A and D). The hydrophobic boundary on the periplasmic side of the membrane is more ambiguous (Smith *et al*, 1998; Lunin *et al*, 2006), but probably defined by the well-conserved Trp323–Lys324–Trp325 sequence (Figure 1C). Although this region is difficult to model (see Supplementary Materials and methods), electron density clearly restricts the main chain (Figure 2A), and the side chains of Trp323 and Trp325 should form a prominent aromatic ring at the periplasmic lipid–aqueous interface. A slight lengthening of TM2 allows room to model Glu316–Arg322 within the density observed connecting TM1 to TM2

(Figure 2A). This does not seem possible using a previously determined structure (TmCorA_{SGC}; PDB 2BBJ; space group *P*2₁2₁2), where Trp323–Lys324–Trp325 are modeled in a more extended conformation (Lunin *et al*, 2006).

The membrane boundaries of CorA defined by these two aromatic rings have important consequences. First, the highly conserved MPEL motif (Figure 1C) lies almost parallel with the membrane surface and constitutes the largest portion of the transporter exposed to the periplasmic space (Figure 2A). Second, the universal GMN motif is perpendicular to the plane of the membrane and lines the first portion of the metal translocation pathway (Figure 2A–D). Third, a metal identified within the pore sits ~10 Å into the membrane (Site A;

Figure 2 Membrane and pore boundaries. **(A)** CorA membrane boundaries. In a stereo representation, residues of the intracellular aromatic ring (Tyr344 and Phe345) are shown as stick models. Electron density calculated from a composite omit map contoured at 2σ shows the connectivity between TM1 and TM2 (i.e. the MPEL loop); it also indicates the relative position of the Trp323–Lys324–Trp325 aromatic ring. Purple spheres represent bound Ca^{2+} ions. Approximate distances are indicated. **(B)** Stereo view of the CorA pore. Side chains directed into the pore are shown as stick models; Gly312 and Gly309 are shown in full. Electron density calculated from a composite omit map at Site A indicates the presence of a hydrated cation; blue and purple are contoured at 3σ and 4.5σ , respectively. Electron density at Site B was calculated from a composite omit map and contoured at 3σ (blue). **(C)** Stereo representation of the Asn314 gate. The protein backbone of TmCorA_{1–351} (present work) is colored purple (Site A is included for reference) and TmCorA_{SGC} (PDB 2BBJ) is colored blue. The side chains of Asn314 and Phe315 are shown as stick models. When comparing the two structures, the side chains of Asn314 have moved up to 3 Å. **(D)** One protomer from TmCorA_{1–351} (purple; Site A is included for reference) and TmCorA_{SGC} (blue; Trp323–Lys324–Trp325 have been omitted for clarity) are shown. The side chains of Leu294, Tyr344, and Phe345 are shown as stick models. While both proteins were crystallized under similar conditions and in an apparently closed state, displacements of TM2 >1 Å are observed.

Figure 2A–D). Moreover, a second metal found ~ 50 Å below Site A is located within the funnel domain; this site marks the intracellular end of the CorA pore (Site B; Figures 1A, 2A and B).

The CorA pore

There are several intriguing features about the CorA pore. For instance, its unprecedented overall length (~ 55 Å) and its sparingly polar nature. Although the side chains of Asn314 and Asp277 ‘cap’ the pore entrance at either end, only the backbone carbonyls and amides from TM1, the hydroxyls of Thr305, Thr287, and Ser284, and the sulfur atoms from Met302 and Met291 contribute to its polar character (Figure 2B). There are no formal charges and, beyond the GMN motif (Figure 1C), no universally conserved residues line the metal translocation pathway. Another striking feature is the cross-sectional width of the pore; its diameter ranges from ~ 6 Å near Gly309 and Ala298 to between 3.5 and 2.5 Å at Met302 and Leu294.

The narrowest constriction along the pore occurs at Leu294. This position is highly conserved as a bulky hydrophobic residue in CorA (Figure 1C), and likely serves as the intracellular gate. We refer to this region as the ‘hydrophobic girdle’, and note that it aligns with the aromatic ring formed by Tyr344–Phe345 (Figure 2A, B and D). We refer to the constriction at Thr305 and Met302 as the ‘hydrophobic belt’, as it appears to be conserved in the prokaryotic proteins (Figures 1C and 2B). Based on their relative positions within the pore, these features presumably play important roles in determining the substrate specificity of this transporter. As the pore is too narrow for Mg^{2+} to pass while maintaining its first hydration shell (~ 5 Å), CorA must at least partially dehydrate substrates during transport or achieve a significant dilation of its pore. It is possible that side chain movements might allow for cation transport in the conformation of TmCorA_{1–351} observed in the crystal structure. We note, for example, that the side chains of Asn314 appear to gate the periplasmic pore entrance (Figure 2C; Lunin *et al.*, 2006).

Metals along the CorA pore

While final confirmation will require anomalous scattering data, the two metals found along the pore were assigned as calcium ions (Site A and Site B; Figure 2A–C) as TmCorA_{1–351} was crystallized in the presence of 200 mM CaCl_2 . These sites are separated by a staggering ~ 50 Å, which is either the closest approach between two cations within the CorA pore, or the closest approach between two bound calcium ions under the experimental conditions. As Ca^{2+} is neither a substrate nor a good inhibitor of CorA (Snively *et al.*, 1989), we favor the latter scenario. Moreover, this result would otherwise deviate from the proximal substrate arrangements that are found in other ion channel and ion transporter proteins (Zhou and MacKinnon, 2003; Gouaux and MacKinnon, 2005; Lobet and Dutzler, 2006; Shi *et al.*, 2006).

Although the exact coordination sphere cannot be deciphered at the current resolution, the density observed at Site A suggests that a hydrated cation is present (Figure 2B). This would be consistent with the hydrogen-bonding constraints for a hydrated divalent cation, as well as the cross-sectional width of the surrounding pore. Site A also seems reminiscent of the water-hydrated cation found within the central cavity of the KcsA K^+ channel (Zhou *et al.*, 2001); and movement of

the CorA Asn314 side chains may be compatible with the passage of hydrated substrates (Figure 2C). As judged by distance, the Site B metal is likely coordinated through water by the side chains of Asp277 (Figure 2B).

In a previous study, electron density within the pore near Gly309 and Ala298 had suggested the presence of bound ions (Lunin *et al.*, 2006). As TmCorA_{SGC} was crystallized in the presence of 300 mM $\text{Mg}(\text{NO}_3)_2$ (PDB 2BBJ; Lunin *et al.*, 2006), it is worth noting that Gly309 is directly adjacent to Site A (Figure 2B). If the cavity near Ala298 can accommodate a hydrated Mg^{2+} ion, as its dimensions suggest, then two hydrated metals may flank the hydrophobic belt at Thr305/Met302 (Figure 2B). Because Ca^{2+} is not a substrate of CorA, and there is no apparent density in the pore near Ala298 within our structure, these comparisons may indicate that the hydrophobic belt performs a critical role in substrate selectivity.

CorA electrostatics and the cation sensor

Many electrostatic surface properties of CorA require mention. TM2 ends with the highly conserved cytoplasmic sequence KKKKWL₃₅₁ (Figure 1C). Therefore, residues Lys346–Lys349 are located near the intracellular lipid–aqueous interface, below the Tyr344–Phe345 aromatic ring (Figure 2A). Together with Lys286 and Lys292 contributed by $\alpha 7$, TmCorA places a total of 30 lysine residues around the pentamer at this membrane proximal region. This ‘basic ring’ (Figure 3A) may play a role in protein folding (von Heijne, 1989) or as a potential pH sensor (Kloda *et al.*, 2006). Although some charge should be neutralized by the surrounding lipid head-groups, this feature will also contribute to the creation of a large dipole. Situated below the basic ring on the funnel domain exterior is an ‘acidic ring’ (Figure 3A), which is formed by a collection of conserved residues contributed by $\alpha 5$ and $\alpha 6$ (Figure 1C). The relative juxtaposition of these oppositely charged rings could propagate movements onto TM2, the periplasmic MPEL loop, and the GMN motif during the translocation cycle of CorA (Figure 2C and D). Below, we provide evidence demonstrating a conformational plasticity in CorA capable of effecting such an outcome.

Beyond Asp277 of the CorA pore, $\alpha 7$ lines the interior of the funnel domain with acidic and hydroxyl-bearing residues (Figure 1C). This feature could create an electrostatic sink that would help draw Mg^{2+} out of the pore; however, basic residues contributed by $\alpha 6$ and $\beta 4$ effectively neutralize the desired negative charge character in TmCorA (Figures 1C and 5C). In our open state model (see below), these basic residues are withdrawn and the funnel domain interior becomes extremely acidic (Figure 5D). This could be the key to overcoming the energetic barrier of Mg^{2+} transport through the CorA pore.

The interface between protomers within the funnel domain is also acidic (Figure 5C). As previously suggested (Lunin *et al.*, 2006), a cation is coordinated between Asp89’ from $\alpha 3'$ and Asp253 of $\alpha 7$ (‘ represents a residue from an adjacent protomer; Figure 3B). A second cation is found coordinated by Asp175 and Asp179 from $\alpha 5$, and probably through waters to Glu88’ of $\alpha 3'$ and His257 from $\alpha 7$ (Figure 3B). A portion of the ‘N-terminal flap’ (i.e. residues 1–25) closes over this second site and stabilizes it (Figures 3B and 5A). These 10 peripheral metal-binding sites appear to be part of TmCorA’s ‘divalent cation sensor’. Although Glu88, Asp89, Asp175,

Figure 3 Charged rings and the divalent cation sensor. (A) The electrostatic surface potential of the CorA exterior. Red – acidic regions; blue – basic regions. The view is similar to that in Figure 1A. Ca^{2+} ions were omitted from the calculation. (B) Stereo view of residues implicated in the divalent cation sensor and bound Ca^{2+} ions (purple spheres). Electron density calculated from composite omit maps are shown for each site, contoured at 3.5σ . The ‘green’ metal is coordinated directly by Asp89’ and Asp253. The ‘yellow’ metal is coordinated by Asp175 and Asp179, and probably through waters to Glu88’ and His257.

Asp179, Asp253, and His257 are highly conserved residues, none are universal (Figure 1C). We anticipate that analogous arrangements exist within all CorA proteins and have indicated some candidate positions (Figures 1C and 3B). Based on the ‘minimal’ sequences of the well-characterized proteins from *Salmonella typhimurium* (StCorA) and *Methanococcus jannaschii* (MjCorA; Figure 1C), the N-terminal flap interactions may not be essential.

Building an open state model

During efforts to obtain crystallographic phases for TmCorA_{1–351}, we crystallized soluble protein constructs from various prokaryotic homologues (Payandeh and Pai, 2006). The structure of AfCorA_{1–263} was solved to 2.9 Å resolution, and a Co^{2+} coordinated to Asp253 and His257 was confirmed experimentally (Figure 4A; see Supplementary Table 1, and Supplementary Materials and methods). It is now apparent that our soluble constructs truncated $\alpha 7$ within the funnel domain. Although the pentameric state is not preserved, and a domain-swapped dimer is observed instead, AfCorA_{1–263} superimposes well with equivalent portions of TmCorA (Figure 4A–C). For instance, the first ~200 residues comprising $\beta 1$ – $\beta 7$ and $\alpha 1$ – $\alpha 5$ are structurally equivalent (RMSD 1.5 Å over 172 C α ; Figure 4B and C). Where $\alpha 5$ is the handle, this subdomain resembles a hammer (Figure 4C).

Figure 4 Structure and comparison of AfCorA_{1–263}. (A) The ‘domain-swapped’ molecules of the AfCorA_{1–263} crystal structure are viewed along the screw axis of its P6₁22 unit cell, where one two-fold axis is perpendicular to the plane of the paper. The orange spheres represent bound Co^{2+} ions coordinated by Asp253 and His257 (these side chains are not shown). A portion of the unbiased experimental electron density from a 3.2 Å Se-MAD experiment is shown over the CorA elbow region of the yellow molecule, contoured at 1.5σ . (B) Superposition of the equivalent residues from TmCorA_{1–351} (purple) demonstrates the domain-swapped nature of the AfCorA_{1–263} structure and implicates the CorA elbow as a region of interest. (C) Independent alignment of the hammer domains from AfCorA_{1–263} and TmCorA_{1–351} demonstrate their structural likeness, as does the independent superposition of $\alpha 6$ and $\alpha 7$.

Superposition of the hammer domains reveals that $\alpha 6$ and $\alpha 7$ from AfCorA_{1–263} are markedly displaced from their positions in the full-length transporter structure (Figure 4B). However, superposition of only $\alpha 6$ and $\alpha 7$ reveals a similar backbone

trace (RMSD 1.5 Å over 54 C α ; Figure 4C). From this analysis, the loop connecting $\alpha 5$ – $\alpha 6$ seems largely responsible for the conformational change observed in AfCorA_{1–263}. We refer to this loop as the ‘CorA elbow’ (Figure 4A–C).

Beyond considering AfCorA_{1–263} a crystallization artifact, a model for TmCorA_{1–263} was constructed based on the AfCorA_{1–263} structure and positioned onto the full-length transporter. Guided by steric considerations alone (i.e. clashes), only one model for the presumed ‘open state’ appears feasible (Figure 5B). Strikingly, $\alpha 6$ would rotate $\sim 105^\circ$ around $\alpha 7$, so we refer to $\alpha 6$ as the ‘swivel helix’. A closer examination of AfCorA_{1–263} reveals that $\alpha 6$ is slightly rotated, and becomes unwound towards the membrane proximal region (Figure 4C). Thus, the swivel helix itself contains a ‘swing’. When viewed end on, transition to the open state model can be described as the counterclockwise rotation of $\alpha 6$ around $\alpha 7$ with a concurrent $\sim 180^\circ$ clockwise swing of the membrane proximal portion of $\alpha 6$; this is coupled to a $\sim 30^\circ$ opening of the CorA elbow (Figure 5A and B). The hammer domains presumably slide along the periphery of $\alpha 7$ during these motions. It is important to note that the loop connecting $\alpha 6$ – $\alpha 7$ appears to remain rigid (Figure 4C). Therefore, movements of the swivel helix should impart significant torque along $\alpha 7$.

A two-state model

In addition to the AfCorA_{1–263} crystal structure, various lines of evidence support the presented open state model. There is significant conservation within the $\alpha 5$ – $\alpha 6$ region and a glycine or proline residue is highly conserved in the CorA elbow (Figure 1C). Moreover, the elbow also appears to be a dynamic structural element. In TmCorA_{SGC} and TmCorA_{1–351}, this is suggested by elevated crystallographic B-factors and poor electron density. As the elbow was untraceable in a 1.85 Å soluble domain structure of TmCorA_{SGC–266} (PDB 2BBH; space group *P*4₁2₁2; Lunin *et al.*, 2006), this trend is not resolution dependent. A report that *Escherichia coli* CorA could only be isolated in a stable form upon the addition of Mg²⁺ (Eshaghi *et al.*, 2005) supports our view that the CorA elbow may function (like a spring) in connection with the metal site occupancy at the divalent cation sensor. Although the cations located at the protomer interface (Figure 3B) may clearly play a role in stabilizing TmCorA (Supplementary Figures 2 and 3), they are not required to maintain its pentameric state (Supplementary Figure 4). These results are consistent with the notion that CorA gating responds to low intracellular substrate concentrations.

We sought evidence of a conformational change in TmCorA based on the observation that the N-terminal flap, residues

Figure 5 Closed and open state models. (A) The crystal structure of TmCorA_{1–351}. A view looking into the funnel domain from the cytoplasmic side. Protomers are individually colored and purple spheres represent bound Ca²⁺ ions. $\alpha 5$, $\alpha 6$, $\alpha 7$, and the N-terminal flap (i.e., N-flap) are indicated. (B) A model for the CorA open state. The N-terminal flap residues were removed and Ca²⁺ ions in the pore are shown for reference only. (C) Electrostatic surface potential of the CorA closed state (same view as part A). Red – acidic regions; blue – basic regions. Bound Ca²⁺ ions were omitted from the calculation. The basic character is largely established by the ‘neutralizing residues’ contributed from $\alpha 6$. Note, the acidic interface between protomers houses the divalent cation sensor. (D) Electrostatic surface potential of the CorA open state model (same view as part B). Red – acidic regions; blue – basic regions. Ca²⁺ ions were omitted from the calculation.

1–25, would clash in our open state model if maintained in its ‘closed state’ conformation (Figure 5A and B). As portions of the N-terminal flap are ‘invisible’ in all available crystal structures, and the well-characterized StCorA and MjCorA proteins lack this sequence (Figure 1C), we reasoned that this region may become disordered (and therefore susceptible to exogenous protease) in the open state of the transporter. Indeed, incubation of TmCorA in the presence or absence of Mg^{2+} reveals differential susceptibility to trypsin. Surprisingly, relative to 4°C, incubation at 37°C protects TmCorA from digestion. This result is more obvious with extended incubation times, where one finds dramatic protection of TmCorA in the presence of Mg^{2+} at 4°C (Figure 6A and B). After ruling out the possibility of cold denaturation (Supplementary Figures 2, 3, and 4), the most straightforward interpretation is that TmCorA becomes ‘resistant’ to trypsin in a divalent cation-dependent manner. Thus, the trypsin-protected conformation presumably reflects the closed state observed in the TmCorA_{1–351} crystal structure (Figures 1A, 5A, and Supplementary Figure 3). The differential protease susceptibility of TmCorA is suggestive of a significant conformational change, and the degradation

pattern is consistent with two defined conformational states (Figure 6A and B).

A thermodynamic trap may explain the increased susceptibility of TmCorA to trypsin at 4°C. Based on our open state model, we assume that the lower temperature slows the return of the CorA elbow and swivel helix to their closed state positions (Figure 5A and B); and consequently, increase the susceptibility of TmCorA to trypsin. Relative to the EDTA conditions at 37°C, that TmCorA is susceptible even in the presence of high Mg^{2+} concentrations is somewhat unexpected (Figure 6A and B). Although the activation of trypsin by certain divalent cations needs to be considered (Green and Neurath, 1953), this result might suggest that an intermittent opening of the transporter occurs under conditions in which Mg^{2+} concentrations are not physiologically limiting. Nonetheless, our data and model would predict that opening and closing of CorA occurs in a concerted manner involving all protomers.

Probing a global divalent cation sensor

Employing the established protease protection assay, titration experiments support the notion that CorA contains a divalent

Figure 6 Protease susceptibility assay. (A, B) Trypsin susceptibility of TmCorA_{1–351} after (A) 1 h and (B) 15 h. Reactions were performed as described in Materials and methods. Control samples provided in the first two lanes are representative of the different experimental conditions. Third lane – molecular weight ladder: 94, 67, 43, 30, 20, and 14 kDa, respectively. (C, D) Based on the results of (A, B), trypsin susceptibility was investigated at 4°C with 15 h incubations. 5 mM EDTA control. Cation concentrations from left to right are: 0, 0.05, 0.1, 0.5, 1, 2.5, and 5 mM. (E) The experiments described in (C, D) were also used to compare the wild-type (wt) TmCorA_{1–351} protein to the Asp253-to-Lys253 mutant (DK). Reaction conditions are indicated.

cation sensor. Specifically, the monovalent cations: Li^+ , Na^+ , K^+ , Rb^+ , and Cs^+ offered TmCorA no protection at the concentrations tested (Figure 6C). With respect to Mg^{2+} , concentrations that afford TmCorA protection are centered within a physiologically meaningful range, that is, ~ 0.5 mM (Figure 6D; Hartwig, 2001; Maguire and Cowan, 2002; Kolisek *et al*, 2003; Froschauer *et al*, 2004; Cromie *et al*, 2006). This result suggests that the divalent cation sensor functions as a molecular switch, and also acts as a global Mg^{2+} homeostatic meter. The minimum protection threshold for TmCorA is $\sim 50 \mu\text{M}$ Mg^{2+} (Figure 6D); remarkably, this set point lies just below the extracellular concentrations required to activate the PhoP/PhoQ two-component system to transcribe the *mgtA* and *mgtB* Mg^{2+} transporter genes (Garcia Vescovi *et al*, 1996; Cromie *et al*, 2006). Concentrations that offer TmCorA no protection, that is, $< 50 \mu\text{M}$ Mg^{2+} (Figure 6D), are the exact conditions that derepress *mgtA* translation (Cromie *et al*, 2006). Therefore, together with previous work (Garcia Vescovi *et al*, 1996; Cromie *et al*, 2006), this study begins to reveal the hierarchy and coordination of prokaryotic Mg^{2+} homeostasis at the molecular level. It is tempting to speculate that an analogous regulatory mechanism exists within the eukaryotic CorA orthologues (Kolisek *et al*, 2003).

The substrates Co^{2+} and Ni^{2+} also offer TmCorA protection, but at relatively high concentrations (i.e. $\sim 50 \mu\text{M}$; Figure 6D). In order to appreciate the relevance to CorA biology, one must consider the extremely low intracellular concentrations of these cations (Outten and O'Halloran, 2001), and also the freely available intracellular concentrations of the cation in question. For example, there is < 1 free Zn^{2+} ion per cell (Outten and O'Halloran, 2001); the corresponding numbers for Ca^{2+} are ~ 30 , $\text{Mg}^{2+} \sim 90\,000$, and for Na^+ and $\text{K}^+ > 2 \times 10^6$ (Maguire and Cowan, 2002). This implies that the CorA divalent cation sensor is ' Mg^{2+} -specific' under physiological settings.

Given the architecture of the divalent cation sensor (Figure 3B), one unanticipated finding is that cobalt(III)-hexamine offers TmCorA strong protection from trypsin proteolysis (Figure 6D). Cobalt(III)hexamine has ammine moieties tightly bound to cobalt, is a potent inhibitor of CorA transport, and is nearly the size of a hydrated Mg^{2+} ion (Kucharski *et al*, 2000; Kolisek *et al*, 2003). When considering that neither Mg^{2+} , Co^{2+} , Ni^{2+} , or Ca^{2+} provided such protection (Figure 6D), this result could be consistent with the proposal that cobalt(III)hexamine and other hydrated cations occupy at least one position within the CorA pore (Figure 2B). As such, this finding might suggest that substrate dehydration is a requisite and early event in the CorA translocation cycle.

Finally, we created an Asp253-to-Lys253 TmCorA mutant (DK). A lysine side chain should mimic a bound cation and lead to the formation of a salt bridge with Asp89' (Figure 3B); thus, we predicted that the DK mutant would be resistant to trypsin in a divalent cation-independent manner. Indeed, even in the complete absence of Mg^{2+} , the DK mutant is markedly protected from trypsin digestion (Figure 6E). We also find that the DK mutant is considerably more stable than the wild-type protein (Supplementary Figure 2); and, in contrast to the wild-type protein (Payandeh and Pai, 2006), the DK mutant crystallizes readily in the absence of added divalent cations (data not shown). Taken together, these results support the notion of a conformational change

in CorA, and implicate the involvement of the divalent cation sensor.

Discussion

CorA points of interest

Lunin *et al*, have proposed that substrate specificity in CorA is achieved by four consecutive properties of the transporter. First, the highly conserved periplasmic or extracellular MPEL sequences constitute part of the selectivity filter. Second, the Asn314 side chains of the GMN motif interact initially with hydrated substrates, and subsequently dehydrate them (at least partially). Third, the polar lining of the pore entrance compensates for the energetic cost of Mg^{2+} dehydration. Fourth, Leu294 is critical for achieving specificity by imposing a constriction that would allow transport of the uniquely small Mg^{2+} ion, a feature they suggest will gate the transporter (Lunin *et al*, 2006).

Several aspects of the above proposal are not supported by our data. For instance, our maps indicate that the MPEL sequences lie almost perpendicular to the pore entrance (Figure 2A); this position does not seem particularly conducive to being part of the selectivity filter. Furthermore, TmCorA_{SGC} was solved in the presence of Mg^{2+} (Lunin *et al*, 2006), conditions that one would expect to order the selectivity filter (Zhou and MacKinnon, 2003); however, the MPEL loops in TmCorA_{SGC} could not be modeled (Lunin *et al*, 2006). Moreover, the well-characterized MjCorA is extremely deviant within the MPEL motif, conserving only the proline residue (Figure 1C). Thus, we assign this loop a role in linking potential movements of TM2 to the CorA pore, but leave open the possibility that it may have other functional and/or structural roles (e.g. increasing the local cation concentration, or being involved in protein folding).

Lunin *et al* (2006) suggest that Asn314 plays a critical role in dehydrating incoming Mg^{2+} ions, and that this is a key basis for selectivity in CorA. While we assume that Asn314 plays an important role in 'screening' potential substrates, these side chains have moved up to $3 \text{ \AA}$ in our structure relative to TmCorA_{SGC} (Figure 2C). This seems more consistent with a role in gating (Dutzler *et al*, 2003). In fact, the entire GMN motif backbone has shifted in our structure, as if to be pulled 'towards' TM2 (Figure 2C and D). These displacements might be more representative of the open state; they may also illustrate consequences associated with even minor movements of TM2 (Figure 2D) and the MPEL loop. Finally, our data indicate that a hydrated cation exists within the pore beyond Asn314 (Figure 2B).

A proposal for specificity and transport in CorA

What is the hydration status of Mg^{2+} within the CorA pore? It is unlikely that a completely dehydrated Mg^{2+} ion will travel the distance in CorA as, beyond energetic considerations, this would produce a large discrepancy between the pore dimensions and the ionic radius of Mg^{2+} . It also seems unlikely that the Leu294 gate will function as the selectivity filter in CorA. For one, Leu294 sits at the intracellular lipid-aqueous interface (Figure 2A and D); moreover, the pore constriction at Leu294 is only $\sim 2.5 \text{ \AA}$, making selection between the ionic forms of Mg^{2+} and the non-CorA substrates Ca^{2+} , Mn^{2+} , Zn^{2+} , and Fe^{2+} (Snively *et al*, 1989; Papp and Maguire, 2004) nearly impossible.

As Mg^{2+} has the smallest ionic radius and the largest hydrated radius among the common biological cations (Maguire and Cowan, 2002), we anticipate that CorA will exploit these properties to achieve substrate specificity and transport. We propose that the CorA pore may contain multiple hydrated metal 'binding' sites intervened by discrete hydrophobic constrictions (Figure 2B). While reducing the overall energetic cost of transporting a completely dehydrated Mg^{2+} ion, this scenario might afford CorA the opportunity to maximize specificity by sensing both the hydrated and ionic radii of its substrates. We suggest a 'linked-beads' type of translocation model, where cations move in discrete and concerted steps along the translocation path. Incoming substrates should force existing metals along the pore; and an intracellular electrostatic sink will help to draw cations out of it. This mechanism could permit the rapid initiation and cessation of transport, conceivably trapping potential substrates during transit (Figure 2A–D). While consistent with a large body of evidence for other proteins (Zhou *et al.*, 2001; Gouaux and Mackinnon, 2005; Lobet and Dutzler, 2006; Shi *et al.*, 2006), this proposal can accommodate the high flux rates measured in various members of the CorA superfamily (Snively *et al.*, 1989; Liu *et al.*, 2002; Kolisek *et al.*, 2003).

CorA is known to transport Co^{2+} and Ni^{2+} , which have comparable ionic radii and share a similar hexacoordinate preference with Mg^{2+} (Harding, 2004); cobalt(III)hexammine, a potent inhibitor of CorA transport, is nearly the same size as a hydrated Mg^{2+} ion. These considerations imply that the small ionic radii of Mg^{2+} , Co^{2+} , and Ni^{2+} are important for their transport, and that CorA interacts with hydrated substrates, at least initially. In addition to sensing size, the commonalities between Mg^{2+} , Co^{2+} , Ni^{2+} , and cobalt(III)hexammine also suggest that CorA might identify substrates through preferences in their coordination geometry. However, as most cations are quite sparse in biological settings, perhaps CorA may only need to select between Na^+ , K^+ , Ca^{2+} , and Mg^{2+} . An analogous suggestion has been made for the Alr1 proteins (MacDiarmid and Gardner, 1998) and the CIC superfamily (Gouaux and Mackinnon, 2005; Lobet and Dutzler, 2006).

We believe that Site A represents a principal component of the selectivity filter in CorA (Figure 2B). We first consider Ca^{2+} , assigned in our structure, and will extend these considerations to Na^+ and K^+ . Although Ca^{2+} is neither a substrate nor a good inhibitor of CorA (Snively *et al.*, 1989), it still seems capable of proceeding beyond Asn314 and appears hydrated at Site A (Figure 2B). It is notable that the hydrated radii of Na^+ , K^+ , and Ca^{2+} are similar, and $\sim 1.5\text{--}2\text{ \AA}$ smaller than that of Mg^{2+} (Maguire and Cowan, 2002). Thus, it is possible that ions with a small hydration shell might not interact with CorA strongly enough to be transported, and may simply diffuse away. As a very large hydration shell would clash with the pore walls, steric restriction should occlude any larger species (Kucharski *et al.*, 2000). Hence, the CorA pore is probably optimized for hydrated substrates that fit 'just right' at Site A to promote their transit past the hydrophobic belt (Figure 2B). The cation hexammine inhibitors (Kucharski *et al.*, 2000) might then plug the pore at Site A because they cannot undergo the necessary 'dehydration' at the Thr305/Met302 hydrophobic belt. Compared to inhibition at Asn314 (Lunin *et al.*, 2006),

this model may explain why cobalt(III)hexammine has a higher affinity for CorA than does Mg^{2+} itself (Kucharski *et al.*, 2000).

Additional factors likely play a role in the selectivity, regulation, and transport cycle of CorA. For example, a second dehydration event at the hydrophobic girdle could impose further specificity. Also, movements impinging onto TM2 could regulate the status of the Asn314 gate through the MPEL loop. This would promote a reloading of substrates into Site A and the translocation of metals along the pore.

The CorA transport cycle

If we assume a two-state model (Figures 5 and 7), there are at least three consequences to consider in the closed-to-open state transition of TmCorA. First, movement of the swivel helix will impart a dramatic acidic charge character at the intracellular end of the pore (Figure 5D). This should provide critical energetic contributions to promote transport and draw Mg^{2+} through the pore. Second, given the rigid nature of the loop connecting the swivel helix to $\alpha 7$ (Figure 4C), considerable torque will be generated along $\alpha 7$. In fact, the open state model requires a 'translation' of $\alpha 7$ by $\sim 1/3$ of a helical turn. This twist would be an effective mechanism to impose a gating force onto Leu294, a residue $\sim 55\text{ \AA}$ away from the $\alpha 6\text{--}\alpha 7$ loop. Third, the acidic ring on the funnel domain exterior (Figure 3A) must undergo significant rearrangement during these movements, which is likely to have an impact upon the basic ring that sits atop TM2 (Figure 3A). We predict that this will influence the relative position and/or orientation of TM2 within the membrane bilayer, although perhaps only by a small amount (Figure 2D). In sum, we propose that the outcome of these conformational changes will simultaneously prime the CorA funnel domain interior, the Leu294 gate, and the Asn314 periplasmic gate for Mg^{2+} transport (Figure 7).

What are the components leading to opening of the transporter? The loss of cation coordination at the divalent cation sensor should set in motion events leading to the open state. As the charge character of the protomer–protomer interface is quite acidic (Figure 5C), loss of cations would result in significant charge repulsion. Perhaps this perturbation allows the CorA elbow to withdraw the swivel helix from its closed state position on route to the open state.

What are the components leading to closure of the transporter? Torque generated on $\alpha 7$ should be one force driving a return to the closed state. Recoiling of the $\alpha 6$ 'swing' and the CorA elbow could be others. Steric restriction on the hammer domain will promote this process, and the domain-swapped arrangement observed in the AfCorA_{1–263} crystal structure (Figure 4A and B) suggests that the hammer domain has an inherent affinity for the closed state. Finally, a local influx of divalent cations, in accordance with the open state, will increase the probability of occupying the divalent cation sensor and help to shut down transport.

Perspectives

Beyond the nature of its pore, perhaps one of the most intriguing features about CorA is its channel-like architecture. Although obviously unique, CorA seems strangely reminiscent of the K^+ -type channels (Zhou *et al.*, 2001; Zhou and MacKinnon, 2003; Shi *et al.*, 2006); and we begin to draw

Figure 7 A schematic model of the CorA translocation cycle. **(A)** The CorA closed state. This conformation presumably represents that observed in the TmCorA_{1–351} crystal structure and the condition in which there is an adequate or high intracellular Mg²⁺ concentration. The color coding is consistent with Figure 1C and major structural features are indicated as follows: MPEL – MPEL loop; 314 – Asn314 gate; HB – hydrophobic belt (residues Met302 and Thr305); HG – hydrophobic girdle (residues Met291 and Leu294); BR – basic ring (residues Lys346–Lys349); AR – acidic ring; NR – neutralizing residues (α 6); ESR – electrostatic sink residues (α 7); DCS – divalent cation sensor; black spheres – Mg²⁺ ions. Only two CorA protomers are shown and, except for α 5, the hammer domain has been omitted for clarity. **(B)** The CorA open state. Implications of the conformational changes in CorA associated with an inadequate or low intracellular Mg²⁺ concentration. With the loss of Mg²⁺ coordination at the DCS, charge repulsion between protomers and the dynamic CorA elbow region have been implicated in bringing about movements of α 6 (the swivel helix), α 5, and the hammer domain. This should impart a force onto the Leu294 gate (at the HG) through a twisting of α 7. A rearrangement of the AR should impinge onto the BR and effect the position and/or orientation of TM2 (i.e. α 8) within the membrane bilayer; this is predicted to open the Asn314 gate through movements of the MPEL loop. As the NR on α 6 have been ‘withdrawn’, the ESR on α 7 will create an acidic environment within the funnel domain interior that will promote ion transport. Factors mentioned in the main text will return CorA to the closed state position shown in (A). Beyond regulating the transport status of CorA, the DCS (in effect) also acts as an intracellular Mg²⁺ sensor.

many parallels with the acetylcholine receptor (Doyle, 2004; Unwin, 2005). Under normal conditions, however, CorA is strictly a ‘transporter’, functioning to move Mg²⁺ against its concentration gradient. Yet, CorA and CorA homologues also function as effluxers; and some of these proteins assume channel-like properties. Thus, the line between an ion channel and an ion transporter has become blurred. CorA, however, does not set this precedent. For example, the CIC superfamily of chloride channels and transporters share nearly identical structures (Dutzler *et al.*, 2003; Lobet and Dutzler, 2006), and the projection structure of the human copper transporter CTR1 has revealed a homo-oligomeric channel-like architecture (Aller and Unger, 2006).

Metals found along the pore in TmCorA_{1–351} have provided clues to the principles of substrate specificity and transport. A hydrophobic gate and a hydrated metal stabilized within the TM region are emerging as channel-specific and cation-specific themes (Zhou *et al.*, 2001; Doyle, 2004; Shi *et al.*, 2006). This work has also unveiled a physiologically tuned divalent cation sensor within TmCorA. Thus, CorA has an intrinsic allosteric regulatory switch that may control Mg²⁺ homeostasis at the cellular level. Based on crystallographic, biochemical, and biophysical data, our model of the open state presents a cohesive structural framework for the translocation cycle of CorA. Hopefully, this will provide a foundation for further study of structure–function relationships within the ubiquitous CorA superfamily.

Materials and methods

Crystallization and data collection

Details on protein purification and crystallization have been reported elsewhere (Payandeh and Pai, 2006). The DK mutant was produced using the QuikChange protocol (Stratagene) and all constructs were confirmed by DNA sequencing. The final protein buffers were 20 mM Tris pH 8.0, 100 mM NaCl, 1 mM TCEP; with or without 0.026% *n*-dodecyl- β -D-maltoside (Anatrace). Besides CaCl₂, no other divalent cations were added during the purification or crystallization of TmCorA_{1–351}.

TmCorA_{1–351} (15 mg/ml; 0.026% *n*-dodecyl- β -D-maltoside) was crystallized in the space group C2 by vapor-diffusion over 20% PEG 400, 0.2 M CaCl₂, 0.1 M HEPES pH 7.0. Native and SeMet-AfCorA_{1–263} (40 mg/ml) were crystallized in the space group P6₁22 over 1.3 M ammonium sulfate, 10 mM CoCl₂, 0.1 M HEPES pH 7.4. X-ray data were collected at the NSLS X8C beamline as described (Payandeh and Pai, 2006). X-ray data were processed using DENZO/SCALEPACK (Otwinowski and Minor, 1997) and the CCP4 suite (CCP4, 1994) (Supplementary Table 1).

Structure determination and refinement

See Supplementary Materials and methods. Structure factors and coordinates for AfCorA_{1–263} and TmCorA_{1–351} have been deposited in the PDB (www.rcsb.org) under the accession codes 2HN1 and 2HN2, respectively.

Alignments and the open state model

ClustalW (Thompson *et al.*, 1994) was used for Figure 1C. LSQMAN (Kleywegt, 1996) was used for all structural alignments. A model of TmCorA_{1–263} was built using LSQMAN to align residues 17–201 and residues 205–263 of TmCorA_{1–351} over equivalent portions of AfCorA_{1–263} as shown in Figure 4C. The CorA elbow was then

rebuilt manually in O (Jones *et al*, 1991), and the TmCorA_{1–263} model was docked onto the full-length TmCorA_{1–351} structure using O.

Molecular figures and electrostatic surface calculations

Structural figures were prepared using MOLSCRIPT (Kraulis, 1991) and RASTER3D (Merritt and Murphy, 1994). After removal of cations, the electrostatic surfaces shown in Figures 3A, 5C and D were generated using SPOCK (Christopher, 1998).

Protease susceptibility assay

For 50 µl reaction volumes, stock solutions were prepared to produce the final concentrations desired upon dilution: 38.75 µl of protein (2.58 mg/ml), 1.25 µl of trypsin (10 mg/ml; Sigma), and 10 µl of 0–100 mM of 'substrate'. Reaction solutions were mixed and equilibrated at 4°C for 20 min. Trypsin was then added, and reactions were incubated at 4 or 37°C for the specified times. After adding 50 µl of SDS–PAGE sample buffer, samples were boiled and run immediately on 15% SDS–PAGE gels. To rule out the possibility of trypsin inhibition, controls were performed on a test protein over the full range of conditions.

Circular dichroism and gel filtration

See Supplementary Materials and methods.

Supplementary data

Supplementary data are available at *The EMBO Journal* Online.

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Note added in proof

While this manuscript was in press, two related reports have appeared. Eshaghi *et al* (2006, *Science* **313**: 354–357) have described the crystal structure of TmCorA at 2.9 Å resolution, which is largely the same as the TmCorA_{1–351} structure presented here. Wang *et al* (2006, *JBC* PMID: 16835234) have performed electron microscopy analysis on a soluble portion of *E. coli* CorA. Although these authors describe EcCorA_{1–223} as a periplasmic tetramer, their electron density envelope seems highly reminiscent of our CorA 'open state' model.

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