ABSTRACT

Introduction: One of the most undesirable consequences of orthodontic treatment is occurrence of enamel demineralization around orthodontic brackets. Numerous in vitro studies have reported the prevention of enamel demineralization by surface treatment with lasers and fluoride varnish.

Aim: To evaluate the changes on the enamel surface and microhardness around orthodontic brackets after surface treatment by CO\textsubscript{2} laser, Er,Cr:YSGG laser and fluoride varnish in vivo.

Materials and Methods: A double blind interventional study was carried out on 100 premolars which were equally divided into five groups, out of which one was the control group (Group 0). The intervention groups (Group I to IV) comprised of patients requiring fixed orthodontic treatment with all 4 first premolars extraction. Brackets were bonded on all 80 premolars which were to be extracted. Enamel surface treatment of Groups I, II and III was done by CO\textsubscript{2} laser, Er,Cr:YSGG laser and 5\% sodium fluoride varnish respectively and Group IV did not receive any surface treatment. A modified T-loop was ligated to the bracket and after two months, the premolars were extracted. Surface changes were evaluated by Scanning Electron Microscopic (SEM) and microhardness testing. Comparison of mean microhardness between all the groups was assessed using post-hoc test with Bonferroni correction.

Results: Group I showed a melted enamel appearance with fine cracks and fissures while Group II showed a glossy, homogenous enamel surface with well coalesced enamel rods. Group III showed slight areas of erosions and Group IV presented areas of stripped enamel. Significant difference was observed between the mean microhardness (VHN) of Group I, Group II, Group III, Group IV and Group 0 with \(p<0.001\). A significant difference of \(p<0.001\) was observed while comparing Group I vs II, III, IV, 0 and Group II vs III, IV, 0. However, difference while comparing Group III vs IV was \(p=0.005\) and difference between the mean microhardness of Group 0 vs Group III was non significant.

Conclusion: Surface treatment with Er,Cr:YSGG laser causes a positive alteration of the enamel surface increasing its ability to resist demineralization with optimum microhardness as compared to CO\textsubscript{2} laser and sodium fluoride varnish.

Keywords: Dental enamel, Sodium fluoride, Surface treatment

INTRODUCTION

The increasing demand to attain a picture perfect smile has led to a rising trend among individuals seeking orthodontic treatment, with aesthetics forming the main reason for seeking the treatment [1]. Unfortunately, one of the possible undesirable consequences of orthodontic therapy is the occurrence of white spot lesions. Despite the advances in orthodontic materials and techniques in recent years, nearly 50\% of orthodontic patients exhibit clinically visible enamel demineralization around orthodontic brackets [2]. There are different methods which can decrease or prevent white spot lesions: improving oral hygiene, modifying diet (low carbohydrate) and treating with topical fluoride. However, most of these methods are patient dependent and compliance is only seen in 13\% of the patients [3]. Thus, attempts have been made to use compliance-free methods.

It has been seen that fluoride varnish application results in 44.3\% decrease in enamel demineralization [4]. However, repeated varnish applications may lead to the temporary discoloration of the teeth and gingival tissue and increase costs to the patient and chair time to the clinician [5].

Introduction of laser in dentistry, in the 1960s by Miaman TH [6], led to a continuous research in the various applications of lasers in dental practice. Several studies have demonstrated the potential of laser pretreatment of enamel to inhibit subsequent acid induced dissolution [6-9]. Previous studies have explored the effect of CO\textsubscript{2} laser and Er,Cr:YSGG laser on the enamel surface and have shown significant inhibition of enamel demineralization of 50\%–98\% upon laser treatment, depending on the laser parameters [6-12]. It has been observed in many studies that the wavelength of CO\textsubscript{2} laser and Er,Cr:YSGG laser is highly compatible with the absorption peak of carbonated hydroxyapatite, which is the major component of the dental enamel (85\%). This energy is efficiently absorbed and converted to heat without damage to the underlying tissues, thereby causing ultrastructural and chemical modifications on the irradiated enamel leading to increased resistance to acid dissolution [8]. Although many approaches which limit enamel demineralization have been tried and tested and also yielded positive results in decreasing demineralization, but most studies to date have evaluated the effect of CO\textsubscript{2} lasers, Er,Cr:YSGG lasers and fluoride applications on the demineralization process in vitro [12-15].
In Group I, the enamel surface of the premolar was treated with CO$_2$ laser, Er:Cr:YSGG laser and 5% sodium fluoride varnish.

**MATERIALS AND METHODS**

The present in vivo double blind intervention study was conducted from March 2015 onwards till March 2016 on patients undergoing fixed orthodontic treatment in the Department of Orthodontics and Dentofacial orthopaedics at Maulana Azad Institute of Dental Sciences, New Delhi, India. Approval was taken from the Institutional Ethical Committee.

Keeping the power of the study at 80% and the probability of type I error at 5%, the interventional study was carried out on 100 premolars which were divided into five groups, each comprising of 20 teeth. The control group was formed by 20 sound enamel samples which were devoid of any surface malformations, cracks and discoloration and the rest 80 premolars were the interventional group. The interventional group had patients in the age group of 13 to 25 years requiring fixed orthodontic treatment with extraction of all 4 first premolars.

The inclusion criteria of the samples were fully erupted first premolars, low caries risk as assessed by caries risk assessment tool given by American dental association [16], premolars devoid of any evident enamel lesions or restoration on the buccal surfaces of teeth. Patients with dental fluorosis or any systemic disease affecting the topography and surface microhardness of sound enamel where no enamel was lost giving rise to a smooth surface [Table/Fig-4].

**In Group II**, the enamel surface of the premolar was treated with a modified T loop was fabricated with 0.016” stainless steel continuous archwire. The T loop was incorporated with a helix of 1 mm on both ends of the arm of the loop. Two helices 1 mm in diameter were incorporated onto the arch wire at a distance of 3 mm from the vertical arm of the loop on each side. The T loop was engaged using an elastomeric ligature to increase plaque accumulation [Table/Fig-1].

In Group III, the enamel surface of the premolar was treated with 5% sodium fluoride varnish (VOCO, Profuorid, Inc. Germany). Application of 5% sodium fluoride varnish was accomplished as per ADA guidelines for Fluoride application.

In Group IV, the bonded premolar received no surface treatment.

After two months, the premolars were extracted and the teeth were disinfected by keeping them in 10% formalin for 48 hours as per OSHA regulations [17] and then stored at room temperature in distilled water till the time of SEM analysis and microhardness testing. For the SEM analysis, the samples were coated with 40 nm of gold using a sputter coater (Palaron-SC 7640, United Kingdom) and then observed in the microscope (Carl Zeiss EVO 40) at the magnification of 2000X [18]. For the purpose of testing the enamel microhardness of all the samples, Vickers hardness testing machine (Microhardness testing machine, MITUTOYO YO, MVK-1) was used. A constant load of 100 g for 10 seconds was applied perpendicular to the enamel surface with the diamond intender to create indentations at three points that were measured using a computerized program. An average of the three readings obtained for per sample was taken.

**STATISTICAL ANALYSIS**

All statistical analysis was performed using SPSS 18.0 software. Descriptive statistics including the mean and standard deviations for enamel microhardness were calculated for each of the five groups. One-way analysis of variance (ANOVA) was used to determine whether significant differences existed between the various groups. Post-hoc multiple comparison test with Bonferroni correction was used to identify which of the groups were significantly different.

**RESULTS**

SEM evaluation of Group 0 revealed circumferentially arranged enamel rods filled with inter rod material producing a typical keyhole appearance [Table/Fig-2]. Evaluation of Group I showed a homogenous enamel surface with fine cracks and fissures over the enamel surface [Table/Fig-3]. Group II showed melting of enamel rods giving rise to a glossy enamel surface with well-coalesced enamel rods. Porous structure of enamel was lost giving rise to a smooth surface [Table/Fig-4]. Evaluation of Group III showed streaks of particle deposition of less
than 1 μm in size on the enamel surface. Areas of slight erosions and porosities were also visible near the bracket tooth interface [Table/Fig-5].

The SEM evaluation of Group IV revealed areas of stripped enamel revealing the enamel rods vividly. Areas of isolated cavitations caused by erosion of the intact enamel surface showed deposition of bacterial colonies [Table/Fig-6].

Results of surface microhardness were obtained using Vickers microhardness testing machine. [Table/Fig-7] depicts the mean, minimum and maximum values of surface microhardness (VHN) and standard deviations for each group.

The highest microhardness value was exhibited by Group I followed by Group II, Group 0 and Group III. The least microhardness value was observed in Group IV.

Level of significance of mean microhardness within groups and between groups was analysed using one-way ANOVA test. There was a significant difference of mean microhardness among five groups with significance of p<0.001 [Table/Fig-8]. Post-Hoc test with Bonferroni correction was done for comparison of mean microhardness between five groups [Table/Fig-9]. Significant difference was observed between the mean microhardness (VHN) of Group I (CO2 Laser), Group II (Er,Cr:YSGG Laser), Group III (5% NaF varnish), Group IV (no surface treatment) and Group 0 (control) with p<0.001. A significant difference of p<0.001 was observed while comparing Group I vs II, III, IV, 0 and Group II vs III, IV, 0. However, significant difference while comparing Group III vs IV was p=0.005 and difference between the mean microhardness of Group 0 vs Group III was non significant.

### DISCUSSION
Orthodontic treatment usually lasts for two years during which almost 50% of patients exhibit clinically visible white spot lesions [2]. Studies state that the rate of decalcification in orthodontic patients especially teenagers is higher than those without orthodontic treatment [19-22]. Post orthodontic treatment, white spot lesions are shown to decrease during the first and second years after debonding. However, areas of demineralization will still remain on the teeth for up to five years after orthodontic treatment and thus posing a cosmetic problem [23].

Some Indian studies like that of Malik A et al., and Chand BR et al., have evaluated the efficiency of lasers and fluoride and their combination in preventing enamel demineralization [24,25]. Although these studies have found positive results in regard to the laser group and fluoride+ laser combination group. However, the limitation in most of these studies was that they were performed in vitro.

Hence, the present in vivo study was undertaken with the intention that greater clarity could be achieved regarding which method of surface treatment could provide us with the best results in decreasing demineralization around orthodontic brackets. In the...
In the present study, the three surface treatment agents employed were CO\textsubscript{2} laser, Er, Cr:YSGG laser and 5% sodium fluoride varnish. First premolars were chosen as sample teeth for our study as not only they are most commonly extracted teeth for orthodontic treatment but also most commonly affected by white spot lesions [26,27]. Since the increased incidence of white spot lesions is due to greater number of stagnation areas owing to fixed appliances, in addition to the bonded appliance, a loop was designed to increase the plaque accumulating surface. Regarding the time required for demineralization, it has been preferentially studied and proven that demineralization of enamel around brackets can be an extremely rapid process with lesions of significant depth (75 µm) can develop in four weeks, a shorter time than many orthodontic appointment intervals of 6 to 10 weeks [1,10,20]. In the present study, the first premolars were extracted after a period of two months (within which alignment of the teeth was carried out using the 0.016” stainless steel continuous arch wire) and were later subjected for evaluation of the enamel surface by SEM and surface microhardness testing.

One of the surface treatment agents we used in the present study was CO\textsubscript{2} laser. Featherstone JDB et al., examined the roles of wavelength (9.3 µm, 9.6 µm, 10.3 µm, or 10.6 µm) in the prevention of caries progression in vitro in enamel [28]. Inhibition of caries progression from 40% to 85% was achieved over the range of laser conditions tested with the optimal results produced by the wavelength of 10.6 µm. In Group I, CO\textsubscript{2} laser was applied on to the enamel surface around the orthodontic bracket in vivo with the specifications (wavelength 10.6 µm, 1 watt power, 1 mm of beam diameter, 20 HZ frequency) for 12 seconds which according to the literature gave the maximum results in decreasing demineralization [15]. The SEM images revealed melting of enamel surface giving a homogenous appearance along with fine cracks and fissures distributed over the surface. The mean microhardness of enamel achieved after CO\textsubscript{2} laser surface treatment was significantly higher than the control group as also observed in respective studies by Miresmaeili A et al., [27] and Strangler L et al., [29] wherein they observed that enamel samples treated with CO\textsubscript{2} laser exhibited a higher value of microhardness as compared to enamel which did not receive any surface treatment. The change in the ultrastructure of enamel can be explained by the fact that surface alteration is achieved from the melting and the subsequent recrystallization of the enamel during cooling. This phenomenon could lead to formation of a resistant enamel surface produced by prominent chemical and mineral content changes [30]. These mineral and structural changes brought by the laser irradiation could be the possible reason that a significantly high microhardness value was achieved. As mentioned earlier, along with the melting of enamel surface, fine cracks and fissures could also be visualized which might act as retentive areas for plaque as also observed in other studies like that of Contreras-Bulnes R et al., [13] where in after surface treatment with CO\textsubscript{2} laser (10.6 µm), SEM analysis of enamel surface revealed rough areas, cracks and strains. In contrast to this, there have been studies which do not report any unwanted appearance of cracks over the enamel surface after the application of CO\textsubscript{2} laser and claim that a smooth morphology of enamel is obtained with CO\textsubscript{2} laser treatment [10,14,15,27,31]. The differences observed in results obtained after application of CO\textsubscript{2} laser in various studies could be due to non uniformity of methodology and laser parameters such as power, beam diameter, frequency and duration of exposure. This notion stresses on the fact that a wider range of understanding and evaluation of different parameters of CO\textsubscript{2} laser is required to attain its maximum benefit in decreasing demineralization.

In Group II, the SEM evaluation revealed a homogenously melted, smooth and glossy looking enamel surface. No cracks and fissures were observed. Out of all the groups evaluated, this group presented with optimal microscopic alterations in the enamel surface showing coalescing of enamel rods making them impervious to acidic dissolution. A significantly higher microhardness for enamel was achieved as compared to the control group. These finding are well in accordance with studies by Geraldo-Martins VR et al., Hossain M et al., and Moslemi M et al., who in their individual studies reported a significant decrease in enamel demineralization after surface treatment with Er,Cr:YSGG laser owing to the appearance of a smooth enamel surface devoid of any cracks [26,32,33].

The evaluation of enamel samples falling under Group III by both the SEM analysis and surface microhardness test revealed that demineralization had occurred around the orthodontic bracket in spite of varnish application. These findings were in accordance with the findings of Basdra EK et al., who used SEM to study the effect of fluoride releasing agents and found that maximum fluoride release was seen in the first 24 hours of application with a steady decline thereafter. They found fluoride release to be negligible after 90 days [34]. Our findings for fluoride varnish were also in accordance with the study conducted by Vitaldi Rodrigues G et al., where they evaluated the effect of varnish in limiting demineralization. They found no statistically significant difference between the mean enamel decalcification index for the control and experimental groups (fluoride varnish) [4].

Group IV samples which received no surface treatment showed significant decrease in microhardness as compared sound enamel. SEM view also showed a rough enamel surface with surface erosions and shallow cavitations with bacterial colonies deposited on them in agreement with the finding that demineralization starts as soon as four weeks after placement of the orthodontic appliance [10, 20].

LIMITATION

The present interventional study was a short term in vivo study wherein a single surface treatment by CO\textsubscript{2} laser, Er,Cr:YSGG laser and 5% sodium fluoride varnish was done. Long term studies need to be carried out to assess the sustainability of the results obtained with a single application of surface treatment agents and whether multiple applications are required for efficiently preventing demineralization for the complete duration of orthodontic treatment.

CONCLUSIONS

According to our present investigation, it was concluded that both the CO\textsubscript{2} laser and the Er, Cr:YSGG laser were successful in increasing the enamel surface microhardness whereas surface treatment with fluoride varnish did not significantly affect the microhardness of enamel. The SEM evaluation showed that surface treatment with CO\textsubscript{2} laser created cracks and fissures on the surface which may act as retentive areas for plaque accumulation. It could also be stated that surface treatment with Er, Cr:YSGG laser causes a positive alteration of the enamel surface increasing its ability to resist demineralization with optimum microhardness.

REFERENCES

Financial or other competing interests: None.

PARTICULARS OF CONTRIBUTORS:
1. Postgraduate Student, Department of Orthodontics and Dentofacial Orthopaedics, Maulana Azad Institute of Dental Sciences, New Delhi, India.
2. Senior Professor and Head, Department of Orthodontics and Dentofacial Orthopaedics, Maulana Azad Institute of Dental Sciences, New Delhi, India.
3. Professor, Department of Orthodontics and Dentofacial Orthopaedics, Maulana Azad Institute of Dental Sciences, New Delhi, India.
4. Professor, Department of Orthodontics and Dentofacial Orthopaedics, Maulana Azad Institute of Dental Sciences, New Delhi, India.

NAME, ADDRESS, E-MAIL ID OF THE CORRESPONDING AUTHOR:
Dr. Tulika Tripathi,
Senior Professor and Head, Department of Orthodontics and Dentofacial Orthopaedics, Maulana Azad Institute of Dental Sciences, MAMC complex, Baharudur Shah Zafar Road, New Delhi-110002, India.
E-mail: dr.tulikatpathi@yahoo.com

Date of Submission: May 22, 2017
Date of Peer Review: Jun 03, 2017
Date of Acceptance: Aug 24, 2017
Date of Publishing: Sep 01, 2017